

# GCSE

# **Additional Science B**

Unit B722/02: Modules B4, C4, P4 (Higher Tier)

General Certificate of Secondary Education

# Mark Scheme for June 2016

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All examiners are instructed that alternative correct answers and unexpected approaches in candidates' scripts must be given marks that fairly reflect the relevant knowledge and skills demonstrated.

Mark schemes should be read in conjunction with the published question papers and the report on the examination.

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Annotations used in scoris

Annotation	Meaning
✓	correct response
×	incorrect response
BOD	benefit of the doubt
NBOD	benefit of the doubt <u>not</u> given
ECF	error carried forward
<b>^</b>	information omitted
I	ignore
R	reject
CON	contradiction

Abbreviations, annotations and conventions used in the detailed Mark Scheme.

- / = alternative and acceptable answers for the same marking point
- (1) = separates marking points
- **allow** = answers that can be accepted
- **not** = answers which are not worthy of credit
- **reject** = answers which are not worthy of credit
- **ignore** = statements which are irrelevant
- () = words which are not essential to gain credit
  - = underlined words must be present in answer to score a mark (although not correctly spelt unless otherwise stated)
- ecf = error carried forward
- AW = alternative wording
- ora = or reverse argument

June 2016

### MARK SCHEME

Question	Answer	Marks	Guidance
1	adding sugar       draws water out of any microorganisms         canning       stops enzymes working in the microorganisms         freezing       stops microorganisms getting to the berries	2	all correct = 2 one or two correct = 1
	Total	2	

Question	Answer	Marks	Guidance	
2 a i	they feed on dead material / dead trees / dead organisms / decaying matter / detritus / rotting material / decomposing material (1)	1	allow dead animals ignore breaks down dead material etc	
ii	any three from:	3		
	population is all (the organisms of) one species living in the same habitat (1)		<b>allow</b> all the same type of organism living in the same area (1) <b>not</b> the number of species	
	community is all the organisms / species / populations living in the same habitat (1)		need reference to same habitat or area at least once for marking points one or two	
	population is all the ash trees or horseshoe bats or thorn moths or woodpeckers (1)			
	community is all the organisms living in the ash woodland (1)			
iii	<b>any two from</b> plantation only contains one type of tree (1)	2		
	idea of less (variety of) habitats (1)		ignore just affects habitats	
	idea that less variety of animals can feed from trees / less variety of food for animals (1)		allow idea of disruption of food chains ignore just affects food chains	
	(trees often planted closer together so) less light reaches floor of forest (1)			
	fewer other plants can grow (1)			
			ignore decreases biodiversity	

Question	Answer	Marks	Guidance
b	first section:         less / no water passes up the xylem (1)         less / no water reaches the leaves (1)         leaves wilt (1)         guard cells         lose water / become flaccid / change         shape (1)	4	Marking points must appear in the correct section to score but no limit on each section allow idea that it interrupts the transpiration stream allow this marking point once only in either section ignore shrivels
	<b>second section:</b> so less carbon dioxide available for photosynthesis (1) less minerals available (to leaves) (1)		<b>ignore</b> reference to respiration but answer must specifically link carbon dioxide to photosynthesis
	Total	10	

#### Question Marks Guidance Answer 4 cows in the barn produce more milk than those in the а 2 field (1) this is because they need to move less / lose less energy keeping warm (1) b both groups would make equal profit (1) 2 Independent marking points difference in milk production is 60kg / £30 (which is **allow** barn = $\pounds$ 90 and field = $\pounds$ 60 (but food for the barn costs the cost of the extra food) (1) £30)(1) allow both make £60 (2) any **two** from 2 С they cannot move around much / less exercise (1) allow not free to move around / cannot roam freely / confined ignore just less freedom lack of stimulation (1) disease may spread / wastes may build up (1) may get aggressive / injure each other (1) Total 6

Question	Answer	Marks	Guidance
5 a	electrons (1)	1	<b>allow</b> correct answer ticked, circled or underlined in list if the answer line is blank or fully crossed out <b>not</b> electron shells
b	any two from:	2	
	he split the atom (1)		
	discovered the nucleus (which was part of an atom) (1)		ignore references to protons and neutrons
	showed that the nucleus was positive (1)		
	idea that he showed that the atom had (lots of) empty space (1)		allow showed that the atom is not a solid mass
	Total	3	

Question	Answer	Marks	Guidance
5	[Level 3] Constructs the balanced symbol equation and draws the 'dot and cross' diagram for $O_2$ and draws the 'dot and cross' diagram for CaO Quality of written communication does not impede communication of the science at this level (5 – 6 marks) [Level 2] Constructs the balanced symbol equation and draws the 'dot and cross' diagram for either $O_2$ or CaO	6	<ul> <li>This question is targeted at grades up to A*.</li> <li>Indicative scientific points at all levels may include: <ul> <li>Correct 'dot and cross' diagram for oxygen.</li> <li>Correct electronic structure for Ca<sup>2+</sup></li> <li>Correct electronic structure for O<sup>2-</sup></li> <li>2Ca + O<sub>2</sub> → 2CaO or any other correct multiple</li> </ul> </li> </ul>
	<b>OR</b> Draws the 'dot and cross' diagram for $O_2$ <b>and</b> draws the 'dot and cross' diagram for CaO Quality of written communication partly impedes communication of the science at this level $(3 - 4 \text{ marks})$ <b>[Level 1]</b> Constructs the balanced symbol equation <b>OR</b> Describes the bonding in CaO and $O_2$ <b>OR</b> Describes or draws the 'dot and cross' diagram for $O_2$ <b>OR</b> Describes or draws the 'dot and cross' diagram for $O_2$ <b>OR</b> Describes or draws the 'dot and cross' diagram for $O_2$ <b>OR</b> Describes or draws the 'dot and cross' diagram for $O_2$ <b>OR</b> Describes or draws the 'dot and cross' diagram for $O_2$ <b>OR</b> Describes or draws the 'dot and cross' diagram for $O_2$ <b>OR</b> Describes or draws the 'dot and cross' diagram for $O_2$ <b>OR</b> Describes or draws the 'dot and cross' diagram for $O_2$ <b>OR</b> Describes or draws the 'dot and cross' diagram for $O_2$ <b>OR</b> Describes or draws the 'dot and cross' diagram for $O_2$ <b>OR</b>		<ul> <li>ignore inner shells</li> <li>allow calcium ion outer shell can be shown as empty.</li> <li>No need to show the charge on the ions – this is given in the question but if it is incorrect if the wrong charge is shown</li> <li>Relevant points describing the bonding at levels 1 could include</li> <li>Calcium oxide is ionic</li> <li>Oxygen is covalent</li> </ul>
	[Level 0] Insufficient or irrelevant science. Answer not worthy of credit. (0 marks)		Use the L1, L2, L3 annotations in scoris. Do not use ticks.
	Total	6	

Question	Answer	Marks	Guidance
7 a	(yes because)	2	no marks for just saying yes – marks are for explanation if no, 0 for question
	(potassium gives a) lilac flame (in the flame test) (1)		<b>allow</b> result from test 2 shows that it contains potassium (1)
	(iodide because a) pale yellow precipitate (with silver nitrate) (1)		<b>allow</b> result from test 5 shows that it contains iodide (1) <b>not</b> iodine
b	any three from:	3	marks can be awarded from a labelled diagram if heating in a test tube / beaker etc scores 0 for question
	idea of use a (flame test) wire (1)		<b>allow</b> use a wooden splint / spray solution (1)
	dip wire into solution / dip wire into solid (1)		
	put wire or substance into a (blue) (Bunsen) flame (1)		
	observe the colour of the flame (1)		
	Total	5	

Question	Answer	Marks	Guidance
8 a	sodium (carbonate) <b>and</b> potassium (carbonate) (1)	1	both needed
bi	0.96 (g) (1)	1	
bii	all metal carbonates (that decomposed) had the same starting mass (1) idea that this is the least amount of solid left / most mass lost (so the greatest amount of gas produced) (1)	2	allow decomposed the most lowest mass of solid left in relation to mass of carbonate (2)
C	$MnCO_3 \rightarrow MnO + CO_2(1)$	1	allow four correct percentage calculations (2) allow any correct multiple not heat in the equation rather than over the equation all formulae must be completely correct
	Total	5	

Que	estion	Answer		Guidance	
9	а	material has (very) low resistance (1)		allow material has no resistance	
	b	idea of super-fast electronic circuits / loss free power transmission (1)	2	allow idea of less or no heat / energy lost	
		only work at (very) low temperatures (1)		<b>ignore</b> cannot work at room temperature <b>allow</b> temperature in the range -100 to -273	
Total		3			

Question	Answer		Marks	Guidance
10			3	
	Number of protons in particle	<b>26</b> (1)		
	Number of electrons in particle	<b>24</b> (1)		
	Number of neutrons in particle	<b>29</b> (1)		
	Total		3	

Question	Answer	Marks	Guidance
11 a	(idea of) electron movement (1)	2	mention of positive electrons or moving ions = 0 marks
	(correct direction idea) from rod (to cloth) (1)		<b>accept</b> cloth gains electrons from rod (2)
			but electrons move from cloth to rod scores (1) only
b	B (1)	1	mark answer on the line first
			<b>allow</b> answer ringed, underlined or ticked on diagram if no answer on the answer line
С	any two ideas from these categories	2	
	quicker (1)		allow more likely to find a solution
	more (different) designs can be tested / range of experiments (1)		<b>allow</b> they can compare results / can check results
	more expertise in different areas / technology / knowledge / information / evidence / perspective (1)		<b>not</b> just 'do things differently' <b>allow</b> more ideas
	spread the cost / share equipment (1)		ignore just cheaper
d	any one from	1	
	better finish (1)		<b>allow</b> does not drip / even coating (1) <b>ignore</b> it sticks better
	shadows painted (1)		allow idea of better penetration of dents or scratches
	less paint used / less waste (1)		ignore just cheaper
	quicker (1)		
	Total	6	

Question	Answer	Marks	Guidance
12 a	2.5 (Ω) (2)	2	
	but if incorrect		
	$\frac{5}{2}$ (1)		
b	40 (cm) (2)	2	
	but if incorrect allow the idea of:		
	2/5ths of 100 (1) <b>or</b> 0.05 ohms/cm (1)		
C İ	3.04 / 3.0 / 3 (2)	2	allow 3.04347826 A or any correct rounding (2)
	but if incorrect		
	<u>700</u> (1) 230		
ii	5 A (1)	1	answer dependent on answer in (c)(i)
			e.g. an answer of 0.33 A is a fuse of 3A answer of 6A is a 10A fuse
			If no answer in ci) answer is 5A
			mark answer on line first
			<b>allow</b> answer ringed, underlined or ticked on diagram if no answer on the answer line
	Total	7	

Question	Answer	Marks	Guidance
Question 13	[Level 3] Answer includes a description of a beta particle and describes two or more changes in the nucleus including some reference to atomic number or mass number and constructs fully correct balanced nuclear equation Quality of written communication does not impede communication of the science at this level (5 – 6 marks) [Level 2] Answer includes a description of a beta particle and one change in the nucleus OR Constructs fully correct balanced nuclear equation Quality of written communication partly impedes communication of the science at this level (3 – 4 marks) [Level 1] Answer includes a description of a beta particle OR one change in the nucleus OR constructs a partially correct balanced nuclear equation Quality of written communication impedes	Marks 6	Guidance         This question is targeted up to A*         Indicative scientific points about the Beta particle include         • Beta particle is a high speed / energy electron         Indicative scientific points about changes in the nucleus include         • Nucleus gains a proton       • Nucleus loses a neutron         • Neutron changes to a proton (counts as two changes)       • Atomic number increases by one         • Mass number unchanged       Indicative scientific points involving the balanced nuclear equation         (14)       14       0         +         (6)       7       -1         Use the L1, L2, L3 annotation in Scoris; do not use ticks
	one change in the nucleus OR constructs a partially correct balanced nuclear equation Quality of written communication impedes communication of the science at this level		Use the L1, L2, L3 annotation in Scoris; do not use ticks
	(1 – 2 marks) <b>[Level 0]</b> Insufficient or irrelevant science. Answer not worthy of credit. (0 marks)		
	Total	6	

Question	Answer	Marks	Guidance
14 a	A (1) idea that half-life is time to reduce count rate to half its original value / idea that half-life is time taken to reduce to 3000 (1) <b>BUT</b> idea that A reaches half its original value (3000) in a shorter time than the others (2)	3	If A not chosen, 0 for question allow time for activity to halve / time for half the mass of isotope to decay / time for half the atoms or nuclei to decay (1) ignore time to give out half the radiation / half of the time taken for the substance to decay ignore time for half the atom or nucleus to decay ignore just it has the shortest half-life (in stem of question) allow detail from the graphs eg reaches half original value between 20 and 30 sec for A 40 and 50 sec for B
			60 and 70 sec for C (1) <b>allow</b> it has the steepest gradient at the start (1) eg A's activity halves in the shortest time (3) <b>ignore</b> faster or quicker time
b i	<ul> <li>any two from</li> <li>C-14 can only be used to date materials that were once living (1)</li> <li>idea that when the plants were living, C-14 levels remain constant (1)</li> <li>idea of C-14 in the plant reducing after plants die (1)</li> <li>idea of comparison of activity in living and dead material / age can be read off a decay curve (1)</li> </ul>	2	<b>allow</b> idea that all living plants have same amount of C-14 (1) <b>allow</b> idea that dead plants do not take in any more C-14 (1)
ii	(idea of) little difference in count rate over 100 years (1)	1	<b>allow</b> 100 years is small compared to the half-life of C-14 (1) <b>ignore</b> the half-life is too long
	Total	6	

Question	Answer	Marks	Guidance
15 a	any three from:	3	
	similarities:		
	both photosynthesise faster / make sugar faster in summer / named months (1)		not makes / produces energy ignore works better allow produces energy in sugars
	in October / April rates are equal (1)		allow traps / stores energy
	differences:		
	deciduous has a higher maximum rate (1)		<b>allow</b> in summer / named month the rate for deciduous is higher (than the evergreen)
	evergreen photosynthesises/ makes sugar throughout the year but deciduous does not / the minimum rate for deciduous tree is lower / deciduous has a faster rate of fall or increase in production (1)		<b>allow</b> only evergreen photosynthesises / makes sugar in winter / named months <b>not</b> makes / produces energy <b>ignore</b> works better
	correct reference to data for both trees in any marking point (1)		<b>allow</b> produces energy in sugars <b>allow</b> traps / stores energy
			so 1600 max v 1000 max (2)
b i	any two from: same month / time of the year / both in July (1)	2	ignore same brightness or intensity of sun
	same place / area of forest (1)		allow same environment / habitat (1)
	same area of each tree (1)		
ii	no, they both trapped 32 000 (1)	1	allow 32000 shown for both on the table

#### Question Marks Guidance Answer 5(%) (2) allow 0.05 (1) c i 2 **but**, if answer incorrect 1600/32000 x 100 (1) ii any two from: 2 It = deciduous tree deciduous trees do not photosynthesise / makes sugar in **not** makes / produces energy / takes in sugar the winter / **allow** produces energy in sugars allow do not make sugar all the year round deciduous trees lose leaves in the winter (1) but deciduous trees have a higher efficiency (1) idea that they can photosynthesise / makes more sugar at certain times or idea that they can photosynthesise / makes more sugar in the summer (1) Total 10

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