



Wednesday 15 June 2016 - Afternoon

GCSE GATEWAY SCIENCE ADDITIONAL SCIENCE B

B721/02 Additional Science modules B3, C3, P3 (Higher Tier)

Candidates answer on the Question Paper. A calculator may be used for this paper.

OCR supplied materials:

None

Other materials required:

- Pencil
- Ruler (cm/mm)

Duration: 1 hour 15 minutes



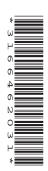
Candidate forename				Candidate surname			
Centre number				Candidate nu	umber		

INSTRUCTIONS TO CANDIDATES

- Write your name, centre number and candidate number in the boxes above. Please write clearly and in capital letters.
- Use black ink. HB pencil may be used for graphs and diagrams only.
- Answer all the questions.
- Read each question carefully. Make sure you know what you have to do before starting your answer.
- Write your answer to each question in the space provided. If additional space is required, you should use the lined page(s) at the end of this booklet. The question number(s) must be clearly shown.
- Do not write in the bar codes.

INFORMATION FOR CANDIDATES

- The quality of written communication is assessed in questions marked with a pencil ().
- A list of equations can be found on page 2.
- The Periodic Table can be found on the back page.
- The number of marks is given in brackets [] at the end of each question or part question.
- The total number of marks for this paper is 75.
- This document consists of 28 pages. Any blank pages are indicated.



EQUATIONS

energy = mass x specific heat capacity x temperature change
energy = mass x specific latent heat

efficiency =
$$\frac{\text{useful energy output (x 100\%)}}{\text{total energy input}}$$

wave speed = frequency × wavelength

power = voltage × current

energy supplied = power x time

average speed =
$$\frac{\text{distance}}{\text{time}}$$

distance = average speed x time

$$s = \frac{(u+v)}{2} \times t$$

$$acceleration = \frac{change in speed}{time taken}$$

force = mass x acceleration

weight = mass x gravitational field strength

work done = force \times distance

$$power = \frac{work done}{time}$$

 $power = force \times speed$

$$KE = \frac{1}{2}mv^2$$

momentum = mass x velocity

$$force = \frac{change \ in \ momentum}{time}$$

$$GPE = mgh$$

$$mgh = \frac{1}{2}mv^2$$

$$resistance = \frac{voltage}{current}$$

3 BLANK PAGE

Question 1 begins on page 4

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4

Answer **all** the questions.

SECTION A – Module B3

1 (a) Mike competes in the triathlon.

This event involves swimming, cycling and running.

He records his heart rate in beats per minute (bpm) during each stage of the triathlon.

Look at his results in the table.

	Heart rate in bpm
Swimming	163
Cycling	165
Running	160

The maximum heart rate is the highest heart rate that is achievable during exercise.

One method to calculate a predicted maximum heart rate uses this formula:

predicted maximum heart rate = 220 - age

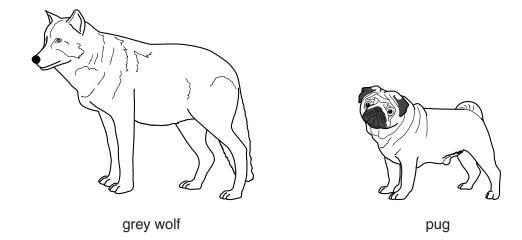
(i) Mike is 29 years old.

Calculate his heart rate during **cycling** as a percentage of his predicted maximum heart rate.

.....%

	(ii)	Mike will gain the most benefit from his training if his heart rate stays within a certain range.
		This range is called the target heart rate zone and is between 60% and 85% of his maximum heart rate.
		Heart rates above the target heart rate zone mean that anaerobic respiration will be used.
		Will Mike gain maximum benefit from his cycling training?
		Explain your answer using ideas about respiration.
		[2]
(b)	Blo	od doping is a way of cheating in sport.
	The	edrug EPO increases the amount of haemoglobin in the blood.
	Exp	plain why taking EPO gives athletes an unfair advantage.
		[2]
		[Total: 6]

2 Look at the pictures below.



The grey wolf is a wild animal.

All dog breeds like the pug are thought to have been bred originally from the grey wolf.

(a) As a result of selective breeding, some pug dogs have breathing problems.

Suggest why pug dogs have breathing problems and explain why continued selective breeding could result in other health problems.

The quality of written communication will be assessed in your answer to this question	n.
	31

7

(b)	(i)	Write down the	name of the ty	pe of cell division	n that makes d	og sperm cells	
							[1]
	(ii)	Put a ring arou	ınd the word th	nat describes do	g skin cells.		
		acrosome	diploid	fertilised	gamete	zygote	[1]
							[Total: 8]

Question 3 begins on page 8

3 (a) This question is about enzymes.

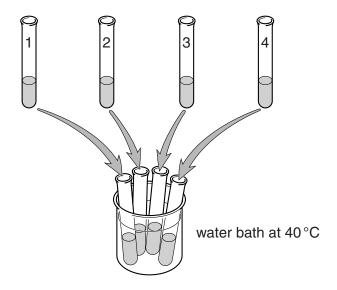
Pepsin is an enzyme that breaks down protein.

Egg-white is a protein that makes water cloudy.

Look at the table below.

It shows an investigation into the effect of adding the enzyme pepsin to egg-white.

Tube 1	Tube 2	Tube 3	Tube 4
5 cm ³ egg-white	5 cm ³ egg-white	5 cm ³ egg-white	5 cm ³ egg-white
3 drops distilled water	3 drops hydrochloric acid	3 drops hydrochloric acid	3 drops hydrochloric acid
1 cm ³ pepsin	1 cm ³ distilled water	1 cm ³ pepsin	1 cm ³ of boiled pepsin



The tubes were put in a water bath at 40 °C for 5 minutes.

Look at the results below.

		Observations of tube contents		
Tube	Contents	At start	At end	
1	egg-white, water and pepsin	cloudy	almost clear	
2	egg-white, hydrochloric acid and water	cloudy	cloudy	
3	egg-white, hydrochloric acid and pepsin	cloudy	clear	
4	egg-white, hydrochloric acid and boiled pepsin	cloudy	cloudy	

(i)	Write a conclusion explaining what the results show about the conditions pepsin needs to work.
	[2]
(ii)	How does the 'lock and key' mechanism explain why pepsin will only break down protein and not other food groups like starch?
	You may draw a diagram to help your answer.
	[2]

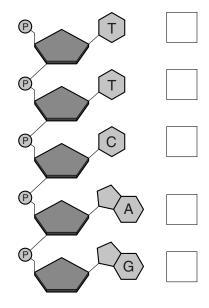
(b) DNA is important in the production of different enzymes.

There are four bases in DNA.

The bases are A, C, G and T.

The diagram below shows one strand of DNA.

Write down in each box, the letter of the base that would be found on the complementary strand of DNA.



[2]

(c) The table below shows the order of bases on DNA that code for different amino acids.

Base order	Amino acid	Base order	Amino acid	Base order	Amino acid	Base order	Amino acid	
AAA	Dhanylalanina	AGA		ATA	Turacina	ACA	Outstains	
AAG	Phenylalanine	AGG	Serine	ATG	Tyrosine	ACG	Cysteine	
AAT		AGT	Serine					
AAC		AGC				ACC	Tryptophan	
GAA	Loucino	GGA	Drolino	GTA	Histidine	GCA	Argining	
GAG	Leucine	GGG		GTG	nistiaine	GCG		
GAT	GGT	Proline	GTT Glutamine	GCT	Arginine			
GAC		GGC		GTC	Giutamine	GCC		
TAA		TGA		TTA	Assoragina	TCA	Corino	
TAG	Isoleucine	TGG	Threonine	TTG	Asparagine	TCG	Serine	
TAT		TGT	Trireonine	TTT	Leading	TCT	A section in a	
TAC	Methionine	TGC		TTC	Lysine	TCC	Arginine	
CAA		CGA		СТА	A a a a with a said	CCA		
CAG	CGG	Alonino	CTG	Aspartic acid	CCG	Oharia a		
CAT	Valine	CGT	Alanine	CTT	01.4	CCT	Glycine	
CAC		CGC		CTC Glutamic a		CCC		

This is part of the DNA base sequence that codes for the enzyme pepsin.

TAACCACTG

(i)	Write down the order of amino acids that are coded for by this section of DNA.
	[2]
(ii)	Explain why the order of amino acids is important for the correct function of pepsin.
	[2]
(iii)	A gene mutation causes a change in the DNA.
	TATCCACTG
	What effect will this have on the function of pepsin?
	Explain your answer.
	[1]
	[Total: 11]

12 SECTION B – Module C3

4 Pete and Helen investigate the reaction between sodium carbonate, Na_2CO_3 , and ethanoic acid, $C_2H_4O_2$.

Sodium ethanoate, C₂H₃O₂Na, carbon dioxide and water are made.

(a) Complete the balanced symbol equation for this reaction.

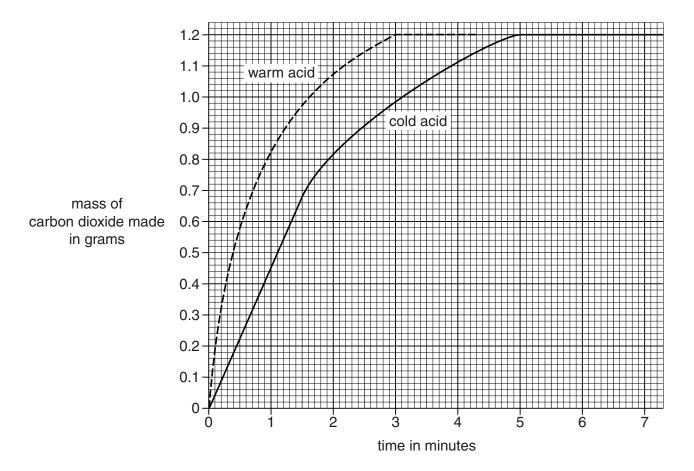
(b) Pete and Helen measure the mass of carbon dioxide made every 30 seconds during the reaction.

They do the experiment again.

They use the same amount of acid and sodium carbonate.

This time they use warm ethanoic acid instead of cold ethanoic acid.

Look at the graph below. It shows their results.



[Total: 10]

	Loo	k at the graph for the warm acid.	
	Hov	v long does it take for the reaction to finish?	
	ans	wer minutes	[1]
(c)	(i)	Look at the graph for the cold acid.	
		Calculate the rate of this reaction during the first 1.5 minutes of the experiment.	
		Give your answer to 2 significant figures.	
		answer g/min	[2]
	(ii)	The rate of reaction during the first 1.5 minutes is greater than at 4 minutes.	
		How can you tell this from the graph?	
			[1]
(d)	The	e reaction with cold ethanoic acid is slower than the reaction with warm ethanoic acid.	
	Ехр	plain, in terms of the reacting particle model, why this reaction is slower with cold acid.	
			[3]
(e)	Cor	mplete the sentence.	
	The	e ethanoic acid is all used up at the end of the reaction because it is the	
		reactant.	[1]

5 Soldiers use 'flameless heaters' to heat their meals.



The 'flameless heater' heats the food safely and quickly without using a flame.

The heater uses a chemical reaction between magnesium metal and water.

$$\rm Mg \, + \, 2H_2O \, \longrightarrow \, Mg(OH)_2 \, + \, H_2$$

(a)	Use the balanced symbol equation to show that mass is conserved during this reaction.
	The relative atomic mass, A_r , of H = 1, Mg = 24 and O = 16.
	[1
(b)	The reaction between magnesium and water is an exothermic reaction.
	Explain why, using ideas about bond breaking and bond making.
	[3

(c) Soldiers need the 'flameless heaters' to heat 227 g of food by 56 °C in less than 12 minutes.

A scientist investigates 'flameless heaters'.

Look at her results below.

Heater	Mass of food heated in g	Temperature rise of food in °C	Time taken in minutes
Α	200	40	8
В	227	45	10
С	227	24	6
D	227	30	5

Which heater will heat 227 g of food by 56 °C in less than 12 minutes?	
Explain your answer.	
	[2]
[То	otal: 6]

6 Pensby pharmaceuticals are making a new painkiller.



They make the drug using a batch process rather than a continuous process.

(a) Explain why batch processes are used for making pharmaceutical drugs.

[1]

(b) It is often expensive to make and develop new drugs.

Explain two reasons why.

.....[2]

[Total: 9]

(c) Pensby pharmaceuticals investigate four different methods of making the new painkiller.

Look at the table below.

It gives information about the four methods they use.

Method	Predicted mass in g	Actual mass in g	Percentage yield	Atom economy	Cost of raw materials per gram
Α	6.7	5.0	75%	50%	£1.92
В	14.2	8.5	60%	85%	£1.30
С	11.5	6.9		40%	£2.16
D	13.3	12.0	90%	80%	£1.36

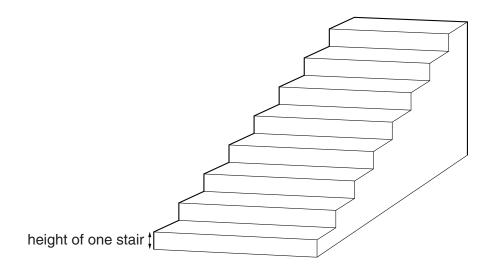
Use the information to calculate the percentage yield for method ${\bf C}$.

Decide which method they should use to make the painkiller and explain your choice.

	The quality of written communication will be assessed in your answer to this question.
•••••	
	[6]

18 SECTION C - Module P3

- 7 This question is about work and power.
 - (a) Janna walks upstairs.



The height of one stair is 0.15 m.

Janna walks up **four** stairs. She has a mass of 50 kg.

Assume that the gravitational field strength (g) is 10 N/kg.

[3
above.
above.
[1]

(b) Janna walks up to the top of the eight stairs every day.

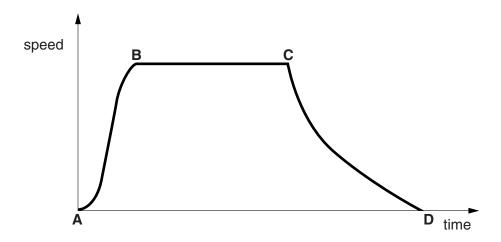
She walks at different speeds each day.

Look at the information below.

Day	Number of stairs	Power developed in watts
Monday	8	107
Tuesday	8	104
Wednesday	8	108
Thursday	8	112
Friday	8	91

[Total:	-
	[2 [.]
Explain why.	
Janna takes the shortest time to reach the top of the stairs on Thursday.	

8 Look at the speed-time graph below for a journey.



Describe the changes in acceleration during the journey and explain how to calculate the distance travelled between time ${\bf B}$ and ${\bf C}$.

Use letters A, B, C and D in your answer.

The	e quality of writte	n communication	will be assesse	d in your answer	to this question.
•••••					
					[6]

21 BLANK PAGE

Question 9 begins on page 22

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9	Seat belts a	re a safety	feature	in cars.
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The design of seat belts has changed since they were first fitted in cars.

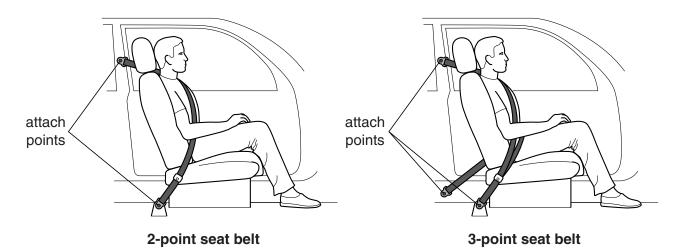
(a)	Scientists collect test data to help them design new seat belts.	
-----	--	--

(i)	Suggest some methods the scientists use to collect valid test data for seat belts.	
		[2
(ii)	Why is it important for scientists to publish the test data they collect?	
		<u>L</u> Z

(b) Some seat belts are attached to the car in two places.

Others are attached in three places.

Look at the diagram below.



Suggest why 3-point seat belts are better at reducing injuries.

[1]

	23
(c)	Test data produced by scientists show that the material which seat belts are made from is important.
	Write down one property which seat belt material must have and explain why this property is useful in an accident.
	[1]
(d)	Pregnant women can find it uncomfortable to wear a seat belt.
	Describe one risk and one benefit of a pregnant woman wearing a seat belt.
	[2]

[Total: 8]

Turn over © OCR 2016

10 Taran wants to buy a new car.

He uses the internet to find data about fuel consumption and emissions.

Look at the table below with the information he finds about two different car models.

Model R	Model S			
Fuel consumption in litres per 100 km:	Fuel consumption in litres per 100 km:			
• In town 5.7	• In town 8.2			
On motorways 4.1	On motorways 5.2			
Combined 4.6	Combined 6.3			
Average carbon dioxide emission 124.0 g/km	Average carbon dioxide emission 149.0 g/km			

(a)		s data is collected using cars that have been driven with the same driving styles eds.	and
	Writ	te down another condition that is kept constant to allow the cars to be compared.	[41]
(b)	(i)	Model S has a higher carbon dioxide emission than Model R.	יין.
		Use the data to suggest why.	
			. [1]
	(ii)	Suggest why the carbon dioxide emissions quoted in the table are average values.	
			. [1]

(c)	Taran's friend, Charlie, has a van.
	Charlie's van has double the mass of Taran's old car.
	Charlie says 'when my van goes twice the speed of your car, it will have four times as much kinetic energy (KE)'.
	Is Charlie correct?
	Explain your answer.
	[2]
	[Total: 5]

END OF QUESTION PAPER

26 ADDITIONAL ANSWER SPACE

If additional space is required, you should use the following lined page(s). The question number(s) must be clearly shown in the margins.				

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The Periodic Table of the Elements

0	4 He helium 2	20 Ne	40 Ar argon 18	84 Kr krypton 36	131 Xe xenon 54	[222] Rn radon 86	t fully
7		19 F fluorine 9	35.5 C t chlorine 17	80 Br bromine 35	127 I iodine 53	[210] At astatine 85	orted but no
9		16 0 oxygen 8	32 S sulfur 16	79 Se selenium 34	128 Te tellurium 52	[209] Po polonium 84	ve been repo
2		14 N nitrogen 7	31 P phosphorus 15	75 As arsenic 33	122 Sb antimony 51	209 Bi bismuth 83	s 112-116 hav authenticated
4		12 C carbon 6	28 Si silicon	73 Ge germanium 32	119 Sn tin 50	207 Pb lead 82	Elements with atomic numbers 112-116 have been reported but not fully authenticated
3		11 B boron 5	27 A1 aluminium 13	70 Ga gallium 31	115 In indium 49	204 T t thallium 81	nts with ator
	·			65 Zn zinc 30	112 Cd cadmium 48	201 Hg mercury 80	Eleme
				63.5 Cu copper 29	108 Ag silver 47	197 Au gold 79	[272] Rg roentgenium
				59 Ni nickel 28	106 Pd palladium 46	195 Pt platinum 78	[271] Ds damstadtium 110
				59 Co cobalt 27	103 Rh rhodium 45	192 Ir iridium 77	[268] Mt meitnerium 109
	1 H hydrogen 1			56 Fe iron 26	101 Ru ruthenium 44	190 Os osmium 76	[277] Hs hassium 108
				55 Mn manganese 25	[98] Tc technetium 43	186 Re rhenium 75	[264] Bh bohrium 107
		mass ool number		52 Cr chromium 24	96 Mo molybdenum 42	184 W tungsten 74	Sg seaborgium 106
	Key	relative atomic mass atomic symbol name atomic (proton) number		51 V vanadium 23	93 Nb niobium 41	181 Ta tantalum 73	[262] Db dubnium 105
		relati atc atomic		48 Ti titanium 22	91 Zr zirconium 40	178 Hf hafnium 72	[261] Rf rutherfordium 104
	·			45 Sc scandium 21	89 Y yttrium 39	139 La* lanthanum 57	[227] Ac* actinium 89
2		9 Be beryllium 4	24 Mg magnesium 12	40 Ca calcium 20	Sr Strontium 38	137 Ba barium 56	[226] Ra radium 88
_		7 Li lithium 3	23 Na sodium 11	39 K potassium 19	85 Rb rubidium 37	133 Cs caesium 55	[223] Fr francium 87

* The lanthanoids (atomic numbers 58-71) and the actinoids (atomic numbers 90-103) have been omitted.

The relative atomic masses of copper and chlorine have not been rounded to the nearest whole number.