



# Wednesday 15 June 2016 - Afternoon

# GCSE GATEWAY SCIENCE ADDITIONAL SCIENCE B

**B721/01** Additional Science modules B3, C3, P3 (Foundation Tier)

Candidates answer on the Question Paper. A calculator may be used for this paper.

OCR supplied materials:

None

### Other materials required:

- Pencil
- Ruler (cm/mm)

**Duration:** 1 hour 15 minutes



Candidate forename					Candidate surname				
Centre number						Candidate nu	ımber		

### **INSTRUCTIONS TO CANDIDATES**

- Write your name, centre number and candidate number in the boxes above. Please write clearly and in capital letters.
- Use black ink. HB pencil may be used for graphs and diagrams only.
- Answer all the questions.
- Read each question carefully. Make sure you know what you have to do before starting your answer.
- Write your answer to each question in the space provided. Additional paper may be used if necessary but you must clearly show your candidate number, centre number and question number(s).
- Do not write in the bar codes.

### **INFORMATION FOR CANDIDATES**

- The quality of written communication is assessed in questions marked with a pencil ( ).
- A list of equations can be found on page 2.
- The Periodic Table can be found on the back page.
- The number of marks is given in brackets [ ] at the end of each question or part question.
- The total number of marks for this paper is 75.
- This document consists of 28 pages. Any blank pages are indicated.

### **EQUATIONS**

energy = mass x specific heat capacity x temperature change
energy = mass x specific latent heat

efficiency = 
$$\frac{\text{useful energy output (x 100\%)}}{\text{total energy input}}$$

wave speed = frequency × wavelength

power = voltage × current

energy supplied = power x time

average speed = 
$$\frac{\text{distance}}{\text{time}}$$

distance = average speed x time

$$s = \frac{(u+v)}{2} \times t$$

$$acceleration = \frac{change in speed}{time taken}$$

force = mass x acceleration

weight = mass x gravitational field strength

work done = force  $\times$  distance

$$power = \frac{work done}{time}$$

 $power = force \times speed$ 

$$KE = \frac{1}{2}mv^2$$

momentum = mass x velocity

$$force = \frac{change \ in \ momentum}{time}$$

$$GPE = mgh$$

$$mgh = \frac{1}{2}mv^2$$

$$resistance = \frac{voltage}{current}$$

3 BLANK PAGE

Question 1 begins on page 4

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1

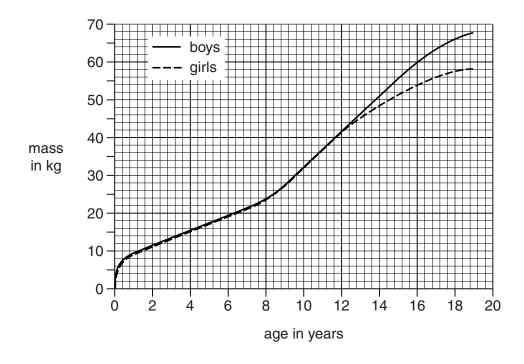
Answer all the questions.

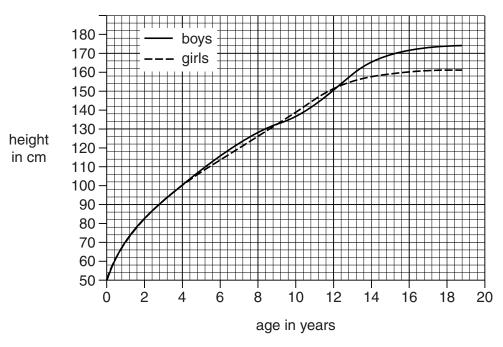
### **SECTION A – Module B3**

1 This question is about growth in boys and girls.

Look at the graphs below.

They show how mass and height change as boys and girls grow.





(a)	Write down the age when the mass of boys becomes greater than the mass of girls.  [1]						
(b)	Compare the growth in boys and girls during the first ten years.						
	[2]						
(c)	Cell division occurs throughout growth.						
	During which age range is cell division fastest?						
	Look at the list.						
	age range in years						
	0 to 2						
	4 to 6						
	6 to 8						
	12 to 14						
	16 to 18						
	Choose your answer from the list.						
	[1]						
	[Total: 4]						

2 Mike competes in the triathlon.

This event involves swimming, cycling and running.

(a) Mike wants to make sure that his training will help him to improve.

Mike needs to keep within his target heart rate zone when training.

Look at the table below.

It shows predicted maximum heart rates and target heart rate zones.

Predicted maximum heart rate in beats per minute	Target heart rate zone in beats per minute
200	120 – 170
195	117 – 166
190	114 – 162
185	111 – 157
180	108 – 153

Table 2

To find his target heart rate zone, Mike must calculate his predicted maximum heart rate.

### predicted maximum heart rate = 220 - age

Mike is 30 years old.

(i)	What is Mike's target heart rate zone?
	Use the predicted maximum heart rate formula and Table 2 above to help you to answer

.....[2]

[Total: 7]

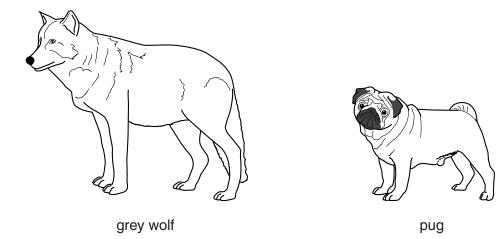
(ii) Mike measures his heart rate during each stage of his training.Look at the table of his results.

	Heart rate in beats per minute
Swimming	110
Cycling	162
Running	160

	Mike will gain the most benefit from his training if his heart rate stays within the targetheart rate zone.	∍t
	What changes should Mike make to his training to get the most benefit?	
	[1	]
(iii)	Part of Mike's training is to measure his recovery time after exercise.	
	How could Mike measure his recovery time?	
	[2	<u>?]</u>
Wh	en Mike takes part in a triathlon the amount of aerobic respiration increases.	
Exp	olain why.	
	[2	<u>?]</u>

(b)

3 (a) Look at the pictures below.



The grey wolf is a wild animal.

All dog breeds like the pug are thought to have been bred originally from the grey wolf.

Some pug dogs have difficulty breathing.

(i) Describe how humans could have bred a dog like the pug from the grey wolf.
Suggest why some pugs may have breathing problems.

The quality of vi question.	vritten communi	ication will be a	assessed in your	answer to this
				[6]

9

							[Total: 9]
	a	0.0001110	aipioid	ioi tillood	gamete	2,9010	[1]
	a	crosome	diploid	fertilised	gamete	zygote	
	(ii)	Put a ring arou	and the word that	t describes dog skil	n cells.		[1]
	(i)	How many chror	mosomes are in	a dog sperm cell?			[4]
(b)	Dog	skin cells have 7	78 chromosomes	S.			
		Mutations are ch	nanges to				[1]
		Complete the se	entence.				
	(ii)	The different cha	aracteristics of d	ogs may be caused	d by <b>mutations</b> in	cells.	

Question 4 begins on page 10

4 (a) This question is about enzymes.

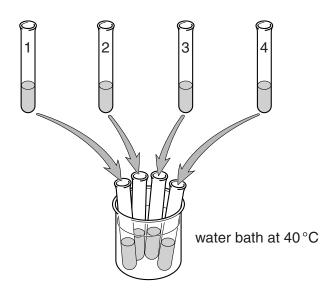
Pepsin is an enzyme that breaks down protein.

Egg-white is a protein that makes water cloudy.

Look at the table below.

It shows an investigation into the effect of adding the enzyme pepsin to egg-white.

Tube 1	Tube 2	Tube 3	Tube 4
5 cm <sup>3</sup> egg-white	5 cm <sup>3</sup> egg-white	5 cm <sup>3</sup> egg-white	5 cm <sup>3</sup> egg-white
3 drops distilled water	3 drops hydrochloric acid	3 drops hydrochloric acid	3 drops hydrochloric acid
1 cm <sup>3</sup> pepsin	1 cm <sup>3</sup> distilled water	1 cm <sup>3</sup> pepsin	1 cm <sup>3</sup> of boiled pepsin



The tubes were put in a water bath at 40 °C for 5 minutes.

Look at the results below.

		Observations of tube contents		
Tube	Contents	At start	At end	
1	egg-white, water and pepsin	cloudy	almost clear	
2	egg-white, hydrochloric acid and water	cloudy	cloudy	
3	egg-white, hydrochloric acid and pepsin	cloudy	clear	
4	egg-white, hydrochloric acid and boiled pepsin	cloudy	cloudy	

	(1)	Write a conclusion explaining what the results show about the conditions pepsin needs to work.
		[2]
	(ii)	How does the 'lock and key' mechanism explain why pepsin will <b>only</b> break down protein and <b>not</b> other food groups like starch?
		You may draw a diagram to help your answer.
		[2]
(b)	Wh	y were all the tubes kept at 40 °C?
		F41
		[1]
		[Total: 51

# 12 SECTION B – Module C3

- Pete and Helen investigate the reaction between marble chips (calcium carbonate) and nitric acid.
   Calcium nitrate, carbon dioxide and water are made.
  - (a) Write the word equation for this reaction.

\_\_\_\_\_[1]

**(b)** Pete and Helen measure the mass of carbon dioxide made every 30 seconds during the reaction.

They do the experiment again.

They use the same amount of acid and marble chips.

This time they use warm nitric acid instead of cold nitric acid.

Look at the graph below. It shows their results.

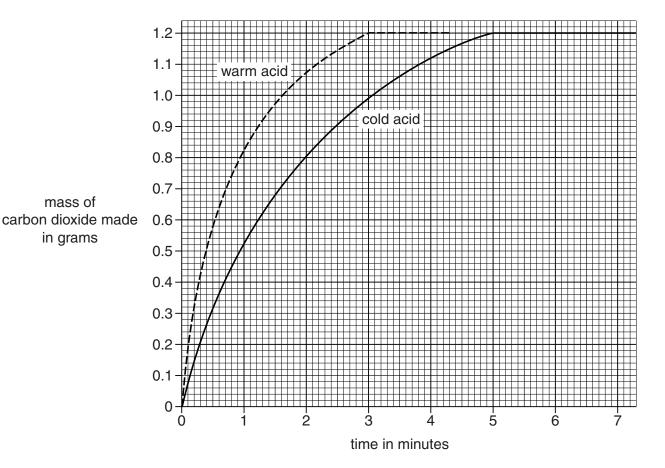


Fig. 5

Look at the graph in Fig. 5 for the **cold** acid.

	(i)	What mass of carbon dioxide is made after 1 minute?	
		answer g	[1]
	(ii)	When is the reaction fastest?	
		Tick (✓) the correct box.	
		between 0 and 1 minute	
		between 1 and 2 minutes	
		between 2 and 3 minutes	
		between 3 and 4 minutes	[1]
	(iii)	Pete thinks that the graph for the <b>cold</b> acid shows that the mass of gas made e minute is the same.	ach
		Is Pete right? Explain your answer.	
			[1]
(c)	The	reaction with warm acid stops after 3 minutes.	
	Ехр	plain why the reaction stops.	
			. [1]
(d)	The	e reaction with <b>cold</b> nitric acid is slower than the reaction with warm nitric acid.	
	Exp	plain, in terms of the reacting particle model, why the reaction is <b>slower</b> with cold acid.	
			[2]

14

eratures.	Reactions are faster at <b>nigher t</b>	(e)
g reactions faster.	Write down <b>two</b> other ways of m	
[2]		
[Total: 9]		

6 Soldiers use 'flameless heaters' to heat their meals.



The 'flameless heater' heats the food safely and quickly without using a flame.

The heater uses a chemical reaction between magnesium metal and water.

$$Mg + 2H_2O \rightarrow Mg(OH)_2 + H_2$$

(a)	Look at the equation.
	Write down the formula for one <b>reactant</b> in this reaction.
	[1]
(b)	The reaction is <b>exothermic</b> .
	What is meant by an exothermic reaction?
	[1]
(c)	A scientist is trying to improve the 'flameless heaters' so that they heat the food more quickly.
	Look at her results.

Heater Temperature rise in the heater in °C		Time taken in minutes
Α	40	8
В	42	7
С	24	6
D	50	10

	[2]
Explain your answer using the temperature rise and time taken.	
Which heater heats up quickest?	

[Total: 4]

7 Pensby pharmaceuticals are making a new painkiller.



They make the drug using a **batch** process rather than a continuous process.

(a)	Write about <b>two</b> differences between a batch process and a continuous process.
	[2]
(b)	It is often expensive to make and develop new drugs.
	Explain <b>two</b> reasons why.
	[2]

(c) Pensby pharmaceuticals investigate four different methods of making the new painkiller.

Look at the table below.

It gives information about the four methods they use.

Method	Predicted mass in g	Actual mass in g	Percentage yield	Atom economy
Α	6.7	4.0	60%	50%
В	14.2	7.1	50%	85%
С	11.5	6.9		40%
D	13.3	12.0	90%	80%

Use the information to calculate the percentage yield for method **C**.

Decide which method the	y should use to	make the	painkiller and ex	plain your choice.

The quality of written communication will be assesse	
 	[6]

[Total: 10]

8 Diamonds are used in jewellery.

One physical property of diamond is that it is colourless.



Graphite is used in pencil leads.

One physical property of graphite is that it is slippery.



[Total:	2]
	[2]
Write down <b>two</b> of these properties.	
Diamond and graphite have some physical properties in common.	

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Question 9 begins on page 20
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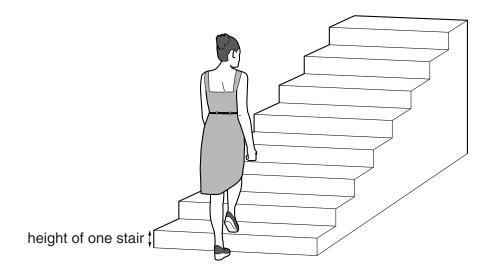
# 20 SECTION C – Module P3

- **9** This question is about work, energy and power.
  - (a) What is the unit for energy?

Choose from

hertz	joule	newton	watt
 			[1]

**(b)** Janna walks up stairs.



(i) The height of one stair is 0.15 m.

Janna walks up	) <b>tour</b> stairs.	. She weigh	าร 500 N.
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Calculate the work done by Janna using this data.

(ii) Janna walks up another four stairs.

She has now walked up eight stairs.

Compare the amount of work done walking up eight stairs with the work done walking up four stairs.

	[4]

(c) Janna walks up the stairs every day.

She walks at different speeds each day.

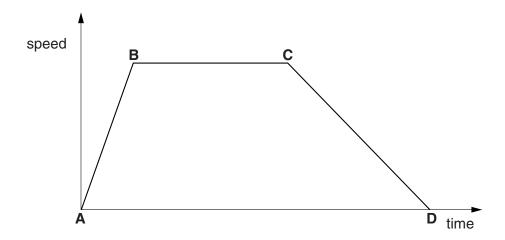
Look at the information below.

Day	Number of stairs	Time in seconds		
Monday	8	4.2		
Tuesday	8	4.4		
Wednesday	8	4.1		
Thursday	8	3.9		
Friday	8	5.4		

llking up the stairs?	nich day did she develop the <b>most</b> powe	On which
[1]		
[Total: 5]		

Question 10 begins on page 22

10 Look at the speed-time graph below for a journey.



Describe the changes in speed **and** acceleration that take place during the whole journey.

Use letters **A**, **B**, **C** and **D** in your answer.

The quality of written communication will be assessed in your answer to this question.	
[6	
	-

# 23 BLANK PAGE

Question 11 begins on page 24

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	11	Seat belts	are a	safety	feature	in	cars.
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(i)

(ii)

The design of seat belts has changed since they were first fitted in cars.

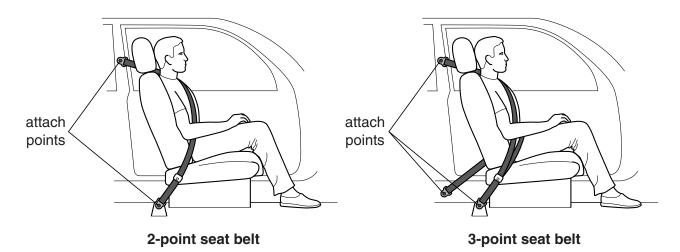
(a) Scientists collect test data to help them design new seat belts.

Suggest methods the scientists use to collect valid test data for seat belts.	
	[2
	· L—,
Why is it important for scientists to publish the test data they collect?	

**(b)** Some seat belts are attached to the car in two places.

Others are attached in three places.

Look at the diagram below.



Suggest why 3-point seat belts are better at reducing injuries.

[11]

(c)	Test data produced by scientists show that the material which seat belts are made from is important.
	Write down one property which seat belt material must have and explain why seat belts are replaced after a crash.
	[2]
(d)	Cars have many different safety features.
	Seat belts protect the driver in the car <b>during</b> an accident.
	Write down the name of a car safety feature that is designed to <b>prevent</b> accidents.
	[1]
	[Total: 8]

Question 12 begins on page 26

**12** Taran wants to buy a new car.

He uses the internet to find data about fuel consumption and emissions.

Look at the table below with the information he finds about two different car models.

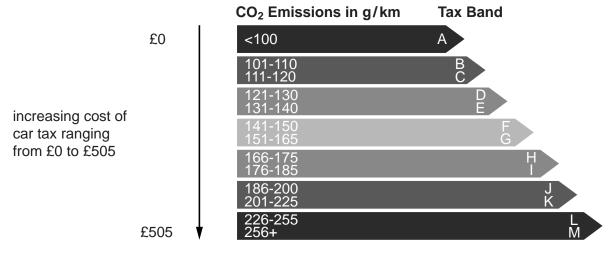
Model R	Model S			
Fuel consumption in litres per 100 km:	Fuel consumption in litres per 100 km:			
• In town 5.7	• In town 8.2			
On motorways 4.1	On motorways 5.2			
<ul> <li>Combined 4.6</li> </ul>	Combined 6.3			
Carbon dioxide emission 124.0 g/km	Carbon dioxide emission 149.0 g/km			
This vehicle is in the UK tax band D and this costs £110.00 a year.				

(a)	Tara	ran looks at the data about <b>fuel consumption</b> .			
	(i)	The fuel consumption for both models is measured in the same units.			
		Why is this important?			
		[1]			
	(ii)	Write down <b>two</b> conclusions from the table about fuel consumption.			
		[2			

[Total: 6]

(b) Taran looks at the carbon dioxide emissions of the two cars, Model R and Model S.

He also uses the chart below to help him find the tax band for model S.



Use the information in the table and the chart to find the tax band for model <b>S</b> and use this to estimate the cost to buy tax for model <b>S</b> .
The tax band for model <b>S</b> is and the estimated cost is £ [2]
The kinetic energy (KE) of a car changes during its journey.
Suggest why the average kinetic energy of a car is less when travelling in a town compared to when travelling on a motorway.
[1]

### **END OF QUESTION PAPER**



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# The Periodic Table of the Elements

0	4 He	20 Ne	40 <b>Ar</b> argon 18	84 <b>Kr</b> krypton 36	131 Xe xenon 54	[222] <b>Rn</b> radon 86	t fully
7		19 F fluorine 9	35.5 C <b>t</b> chlorine 17	80 Br bromine 35	127 <b>I</b> iodine 53	[210] At astatine 85	orted but no
9		16 0 oxygen 8	32 <b>S</b> sulfur 16	79 Se selenium 34	128 Te tellurium 52	[209] Po polonium 84	ve been repo
2		14 N nitrogen 7	31 P phosphorus 15	75 As arsenic 33	122 Sb antimony 51	209 Bi bismuth 83	s 112-116 hav authenticated
4		12 C carbon 6	28 Si silicon 14	73 Ge germanium 32	119 Sn tin 50	207 <b>Pb</b> lead 82	mic numbers a
3		11 <b>B</b> boron 5	27 At aluminium 13	70 <b>Ga</b> gallium 31	115 In indium 49	204 T t thallium 81	Elements with atomic numbers 112-116 have been reported but not fully authenticated
	·			65 <b>Zn</b> zinc 30	112 Cd cadmium 48	201 Hg mercury 80	Eleme
				63.5 Cu copper 29	108 Ag silver 47	197 <b>Au</b> gold 79	Rg roentgenium
				59 Ni nickel 28	106 Pd palladium 46	195 Pt platinum 78	Ds damstadtium 110
				59 Co cobalt 27	103 Rh rhodium 45	192 <b>Ir</b> iridium 77	[268] Mt meitnerium 109
	1 H hydrogen 1			56 Fe iron 26	101 Ru ruthenium 44	190 Os osmium 76	[277] Hs hassium 108
				55 Mn manganese 25	[98] Tc technetium 43	186 Re rhenium 75	[264] Bh bohrium 107
		mass ool number		52 Cr chromium 24	96 Mo molybdenum 42	184 W tungsten 74	Sg seaborgium 106
	Key	relative atomic mass atomic symbol name atomic (proton) number		51 V vanadium 23	93 Nb niobium 41	181 <b>Ta</b> tantalum 73	[262]     Db     dubnium     105
		relati <b>atc</b> atomic		48 Ti titanium 22	91 Zr zirconium 40	178 Hf hafnium 72	[261] Rf rutherfordium 104
			_	45 Sc scandium 21	89 Y yttrium 39	139 La* Ianthanum 57	[227] Ac* actinium 89
2		9 Be beryllium 4	24 Mg magnesium 12	40 Ca calcium 20	88 Sr strontium 38	137 <b>Ba</b> barium 56	[226] <b>Ra</b> radium 88
_		7 Li lithium 3	23 <b>Na</b> sodium 11	39 K potassium 19	85 <b>Rb</b> rubidium 37	133 Cs caesium 55	[223] Fr francium 87

\* The lanthanoids (atomic numbers 58-71) and the actinoids (atomic numbers 90-103) have been omitted.

The relative atomic masses of copper and chlorine have not been rounded to the nearest whole number.