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## GCSE (9–1)

# **Combined Science A (Physics) A (Gateway Science)**

## J250/11: Paper 11 (Higher Tier)

General Certificate of Secondary Education

## Mark Scheme for Autumn 2021

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This mark scheme is published as an aid to teachers and students, to indicate the requirements of the examination. It shows the basis on which marks were awarded by examiners. It does not indicate the details of the discussions which took place at an examiners' meeting before marking commenced.

All examiners are instructed that alternative correct answers and unexpected approaches in candidates' scripts must be given marks that fairly reflect the relevant knowledge and skills demonstrated.

Mark schemes should be read in conjunction with the published question papers and the report on the examination.

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## 1. Annotations available in RM Assessor

Annotation	Meaning
$\checkmark$	Correct response
X	Incorrect response
<b>^</b>	Omission mark
BOD	Benefit of doubt given
CON	Contradiction
RE	Rounding error
SF	Error in number of significant figures
ECF	Error carried forward
L1	Level 1
L2	Level 2
L3	Level 3
NBOD	Benefit of doubt not given
SEEN	Noted but no credit given
I	Ignore

2. Abbreviations, annotations and conventions used in the detailed Mark Scheme (to include abbreviations and subject-specific conventions).

Annotation	Meaning
1	alternative and acceptable answers for the same marking point
✓	Separates marking points
DO NOT ALLOW	Answers which are not worthy of credit
IGNORE	Statements which are irrelevant
ALLOW	Answers that can be accepted
()	Words which are not essential to gain credit
	Underlined words must be present in answer to score a mark
ECF	Error carried forward
AW	Alternative wording
ORA	Or reverse argument

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#### 3. Subject-specific Marking Instructions

### INTRODUCTION

Your first task as an Examiner is to become thoroughly familiar with the material on which the examination depends. This material includes:

- the specification, especially the assessment objectives
- the question paper
- the mark scheme.

You should ensure that you have copies of these materials.

You should ensure also that you are familiar with the administrative procedures related to the marking process. These are set out in the OCR booklet **Instructions for Examiners**. If you are examining for the first time, please read carefully **Appendix 5 Introduction to Script Marking: Notes for New Examiners**.

Please ask for help or guidance whenever you need it. Your first point of contact is your Team Leader.

The breakdown of Assessment Objectives for GCSE (9-1) in Combined Science A:

	Assessment Objective
A01	Demonstrate knowledge and understanding of scientific ideas and scientific techniques and procedures.
AO1.1	Demonstrate knowledge and understanding of scientific ideas.
AO1.2	Demonstrate knowledge and understanding of scientific techniques and procedures.
AO2	Apply knowledge and understanding of scientific ideas and scientific enquiry, techniques and procedures.
AO2.1	Apply knowledge and understanding of scientific ideas.
AO2.2	Apply knowledge and understanding of scientific enquiry, techniques and procedures.
AO3	Analyse information and ideas to interpret and evaluate, make judgements and draw conclusions and develop and improve experimental procedures.
AO3.1	Analyse information and ideas to interpret and evaluate.
AO3.1a	Analyse information and ideas to interpret.
AO3.1b	Analyse information and ideas to evaluate.
AO3.2	Analyse information and ideas to make judgements and draw conclusions.
AO3.2a	Analyse information and ideas to make judgements.
AO3.2b	Analyse information and ideas to draw conclusions.
AO3.3	Analyse information and ideas to develop and improve experimental procedures.
AO3.3a	Analyse information and ideas to develop experimental procedures.
AO3.3b	Analyse information and ideas to improve experimental procedures.

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Question	Answer	Marks	AO element	Guidance
1	A✓	1	1.1	
2	B√	1	1.2	
3	B √	1	1.1	
4	D✓	1	2.2	
5	A✓	1	1.2	
6	C✓	1	1.1	
7	D✓	1	2.2	
8	B✓	1	2.1	
9	D✓	1	2.1	
10	C √	1	1.2	

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Q	Question		Answer	Marks	AO element	Guidance
Q 11		ion (i) (ii)	[1 before 5] 5 before 4 $\checkmark$ 4 before 3 $\checkmark$ [3 before 2] Peer Review <b>Any one from:</b> to confirm the experiment is reproducible $\checkmark$ to check the method $\checkmark$ to check for mistakes $\checkmark$ to ensure quality of research / uphold standards / prevent unethical behaviour $\checkmark$ to offer advice by experts $\checkmark$ Communication <b>Any one from:</b> other scientists can check / test / verify / validate results (against own results) $\checkmark$	Marks 2 2		ALLOW to check it is carried out correctly IGNORE to see if others think the same ALLOW to ensure the experiments are performed correctly / to know of any downfalls / checking the findings / check the validity / to cross check ALLOW to detect false claims ALLOW to add more ideas/points / to give another opinion
			other scientists can develop or use ideas or theories $\checkmark$ others can use or compare the data $\checkmark$ improve knowledge or education $\checkmark$ more data available / allow further development or research $\checkmark$			ALLOW to make corrections ALLOW other investigations or experiments could be done afterwards / for more opinions and theories ALLOW to see similarities and differences
			to gain credit or acknowledgement for their work $\checkmark$			ALLOW help with new technologies

Q	Question		Answer	Marks	AO element	Guidance	
11	(b)	(i)	Any two from: Density of solids is the greatest / AW ✓ Density of gases is the least / AW ✓ Density of solids > density of liquids / ORA ✓ Density of solids > density of gases / ORA ✓	2	2 × 3.1a		
			Density of liquids > density of gases / ORA $\checkmark$				
		(ii)	Atoms more tightly packed in a solid than liquid / ORA $\checkmark$ So greater mass in the same volume / ORA $\checkmark$	2	2 × 1.1	ALLOW answers on a labelled diagram	
	(c)	(i)	FIRST CHECK THE ANSWER ON ANSWER LINE If answer = 169 200 (J) award 2 marks $E = 0.2 \times 846\ 000 \checkmark$ $E = 169\ 200\ (J) \checkmark$	2	2 × 2.1		
		(ii)	Increasing temp (of 1kg of water by 1°C) only requires molecules to move faster ✓ Evaporation requires intermolecular forces to be overcome ✓	2	2 × 2.1	ALLOW increases the kinetic energy of molecules ALLOW to break (intermolecular) bonds	

Q	Question		Answer		AO element	Guidance	
12	(a)		Rate or how fast work is done, or energy is transferred / AW ✓	1	1.1		
	(b)	(i)	FIRST CHECK THE ANSWER ON ANSWER LINE If answer = 908 (N) award 3 marks	3			
			200 lbs = 200 × 0.454 kg = 90.8 (kg) $\checkmark$ W = 90.8 × 10 $\checkmark$ W = 908 (N) $\checkmark$		1.2 2 × 2.1		
		(ii)	FIRST CHECK THE ANSWER ON ANSWER LINE If answer = 248 (W) award 4 marks	4	4 × 2.1		
			$W = 620 \times 1.6 \checkmark W = 992 (J) \checkmark P = 992 / 4 \checkmark P = 248 (W) \checkmark$			ALLOW ecf for incorrect work done for MPs 3 and 4	

Question	Answer	Marks	AO element	Guidance
13 *	<ul> <li>Please refer to the marking instructions on page 4 of this mark scheme for guidance on how to mark this question.</li> <li>Level 3 (5–6 marks)</li> <li>Detailed evaluation of statements in terms of distance, displacements and velocity.</li> <li>AND</li> <li>Calculations included to explain answer.</li> <li>There is a well-developed line of reasoning which is clear and logically structured. The information presented is relevant and substantiated.</li> <li>Level 2 (3–4 marks)</li> <li>Detailed evaluation of statements in terms of distance, displacements and velocity.</li> <li>OR</li> <li>Outline evaluation of statements in terms of distance, displacements and velocity.</li> <li>AND</li> <li>An attempt to include calculations to explain answer.</li> <li>There is a line of reasoning presented with some structure. The information presented is relevant and supported by some evidence.</li> <li>Level 1 (1–2 marks)</li> <li>Evaluation of statements in terms of distance OR displacements OR velocity.</li> <li>OR</li> <li>Calculations included to explain answer.</li> <li>There is an attempt at a logical structure with a line of reasoning. The information is in the most part relevant.</li> <li>O marks</li> <li>No response or no response worthy of credit.</li> </ul>	6	2 × 1.1 4 × 3.1b	<ul> <li>AO3.1b - Analyses information to evaluate speed and velocity</li> <li>Distance = 90 m</li> <li>Displacement = 18 m</li> <li>S = distance / time = 90/120 = 0.75 m/s</li> <li>V = displacement / time = 18/120 = 0.15 m/s</li> <li>Velocity &lt; speed</li> <li>Displacement &lt; distance</li> <li>Therefore, the statements are false</li> </ul> AO1.1 - Demonstrates knowledge of vector and scalar distinction as it applies to speed, velocity, distance, displacement <ul> <li>Velocity = displacement / time</li> <li>Speed = distance / time</li> <li>Displacement = shortest distance between 2 points in a certain direction</li> <li>Velocity and displacement are vectors as they have direction</li> </ul>

Q	uesti	ion	Answer	Marks	AO element	Guidance
14	(a)		Force of floor on ball / Normal contact/reaction ✓ Force of ball on floor / Weight ✓	2	2 × 2.2	ALLOW for one mark if no other marks awarded, arrows of same size in opposite directions vertically and forces labelled in some incomplete way e.g., force of floor, force of ball
	(b)	(i)	First <b>AND</b> Forces on ball are balanced / resultant force = $0 \checkmark$	1	1.1	BOTH needed.
		(ii)	Third ✓	3	1.1	
			Forces act in pairs $\checkmark$ Force of foot on ball = force of ball on foot / one force acts on foot / AW $\checkmark$		2 × 2.1	ALLOW equal and opposite reaction
	(c)	(i)	FIRST CHECK THE ANSWER ON ANSWER LINE If answer = 500 (m / s <sup>2</sup> ) award 2 marks acceleration = $25 / 0.05 \checkmark$ acceleration = $500 (m / s2) \checkmark$	2	2 × 2.1	
		(ii)	FIRST CHECK THE ANSWER ON ANSWER LINE If answer = 480 (m / s <sup>2</sup> ) award 3 marks Rearrangement: acceleration = force $\div$ mass $\checkmark$ acceleration = 216 / 0.45 $\checkmark$ acceleration = 480 (m / s <sup>2</sup> ) $\checkmark$	3	1.2 2 × 2.1	

Q	Question		Answer		AO element	Guidance	
15	(a)		Voltmeter AND 0.1 (V) ✓	1	1.1	BOTH needed. IGNORE unit	
	(b)		Complete circuit <b>AND</b> cell/battery/power source ✓ Voltmeter in parallel with resistor <b>AND</b> ammeter in series with resistor ✓ Correct symbol for resistor ✓	3	3 × 1.2		
	(c)	(i)	3.3 (V) ✓ 44 (mA) ✓	2	2 × 1.2	<b>ALLOW</b> 0.044 or $44 \times 10^{-3}$ or $4.4 \times 10^{-2}$ but only if units amended to A on answer line	
		(ii)	FIRST CHECK THE ANSWER ON ANSWER LINE If answer = 75 ( $\Omega$ ) award 3 marks Recall and rearrange: Resistance = p.d. ÷ current $\checkmark$ R = 3.3 / 0.044 $\checkmark$ R = 75 ( $\Omega$ ) $\checkmark$	3	1.2 2 × 2.1	ALLOW ECF from values of p.d and current in (c)(i) ALLOW 0.075 ( $\Omega$ ) when using value of current in mA rather than A for 2 marks	

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Q	Question		Answer	Marks	AO element	Guidance
15	(d)	(i)	Wire will get hot when current flows / temperature of wire will increase ✓	2	2 × 3.3a	IGNORE no energy lost / more current / less current
			Any one from: Water reduces risk of burns ✓			
			To keep wire at constant temperature $\checkmark$			ALLOW stabilises the temperature
			To keep resistance of wire constant $\checkmark$			
		(ii)	Any two from: Allows p.d. across wire to be varied $\checkmark$	2	2 × 3.3b	
			Allows repeat readings ✓			
			Allows mean resistance to be calculated $\checkmark$			
			Allows (I-V) graph to be plotted $\checkmark$			
			Can check if wire has linear dependence (on p.d.) $\checkmark$			

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