	Il parts accept by implication reference to X . in (i) accept "it covers a greater range of temperatur			
		(i)	X covers greater range of <u>temperature</u> OR (goes to) higher temperature OR greater range expressed numerically	B1
		(ii)	liquid in X expands uniformly (with temperature rise)	B1
		(iii)	(for the same temperature rise,) the liquid in X expands more	B1
	(b)	(i)	two junctions correctly connected to each other and to meter OR one junction between wires and other junction at connection to meter temperature difference between junctions two wires correctly labelled as made of different materials, accept labels metal A &	M1 A1
			metal B NOT 3 different metals labelled	B1
		(ii)	junction (in liquid) has low mass/small heat capacity/small size temperature of junction reacts quickly/quickly reaches temperature of liquid/heat or cools faster	B1 B1
			[Tota	l: 8]
2	(a	volu	two from: ime (of a liquid/gas); resistance (of a metal);	
2	(a	volu		В2
		volu volt	ime (of a liquid/gas); resistance (of a metal);	B1 B1
		volu volt	 Ime (of a liquid/gas); resistance (of a metal); age (of a thermocouple); other appropriate examples; 1 place bulb in ice and water mixture AND mark liquid level 2 place bulb in steam from boiling water AND mark liquid level pure ice OR pure water mentioned in 1 OR at normal atmospheric pressure mentione 	B1 B1 d

3	(a	(i)	(liquid) has a uniform expansion/expands at a constant rate/expands evenly/expands linearly	B1
		(ii)	any two from: larger bulb/wider/longer bulb more liquid narrower capillary/tube use liquid with greater expansion	B2
		(iii)) thermometer must be longer	B1
		(111)		ы
	(b)	res vol vol col am col	y 2 from: istance/conductance of a metal/wire/conductor/thermistor tage/current of a thermocouple ume/pressure/expansion/contraction of a gas our of a metal iount of radiation OR frequency OR wavelength of radiation from a metal/furnace our/arrangement of liquid crystals bansion of a solid/any dimension of a solid	
			nding of a bimetallic strip	B2
			[To	tal: 6]
4	(a		liquid evaporates) at any temperature/below the boiling point/over a range of nperatures/below 100°C/at different temperatures/not at a fixed temperature	B1
		(du	ring evaporation) vapour forms at/escapes from the surface of the liquid	B1
			thout a supply of thermal energy,) evaporation continues/occurs/doesn't stop a causes liquid to cool/is slower/reduces	
	(b)	(i)	(Q =) mL OR 0.075 × 2.25 × 10 ⁶	C1
			1.7 × 10 ⁵ J	A1
		(ii)	(<i>E</i> =) <i>VIt</i> OR 240 × 0.65 × (20 × 60) OR <i>P</i> = <i>IV</i> <u>and</u> <i>P</i> = <i>E</i> / <i>t</i> OR energy/time	C1
			1.9 × 10 ⁵ J	A1
		(iii)	energy is transferred <u>to the surroundings</u> OR in heating the surroundings/air/atmosphere/hot-plate	
			[То	tal: 8]

5	(a	(the	ermal) energy/heat to heat unit mass/1kg/1g	B1
		by	unit temperature/1°C/1K	B1
	(b)		SHC= $Q/(m\Delta T)$ in any form or $Q/(m\Delta T)$ words, symbols or numbers	C1
			(SHC = $8700/800 \times 12$ =) 0.91 J/(g °C) or 910 J/(kg °C)	A1
		(ii)	th. cap. = $Q/\Delta T$ in any form or $Q/\Delta T$ or $m \times$ SHC words, symbols or numbers	C1
			(th. cap. = 8700/12 or 0.906 \times 800 or 906 \times 0.8 =) 730 J/°C or 725 J/°C	A1
	(c)	lag	(cylinder)/wait after heating until temperature stable/at max. value	M1
		prevents/reduces heat losses or heat (energy) takes time to flow throughout block throughout 4(c) , reward correct alternative physics which answers the question e.g. use greater power to reduce expt time and hence energy lo ignore: repeats or use thermometer with low thermal capacity		
				[Total: 8]
6	(a	(m	=) $Pt/l \mathbf{OR} 460 \times 180/2.3 \times 10^6 \mathbf{OR} 82 800/2.3 \times 10^6$	C1
		0.0	36 kg OR 36 g	A1
	(b)	(i)	any two from: (surface) area draught temperature (of water/room) humidity of air	B2
		(ii)	any two from:	
			evaporation at any temperature/below boiling point evaporation (only) at the surface	
			evaporation influenced by surface area/draught/temperature/humidity (not if given in (b)(i))	B2
				[Total: 6]

7	(a	(i)	A OR left hand thermometer	B1
		(ii)	E AND longest length and smallest range/more length per degree/liquid moves more per degree/increases the most per degree	B1
	(b)	nar larg	two from: row bore/tube e amount of liquid/mercury/ethanol/alcohol/bulb id with large expansivity OR ethanol instead of mercury	B2
	(c)	80	(°C) OR 80/120 OR 18/120	
		120	cm	A1

[Total: 6]