Q1.A student used the apparatus in **Figure 1** to obtain the data needed to calculate the specific heat capacity of copper.

12 V
Power supply

Joulemeter

Copper block

The initial temperature of the copper block was measured.

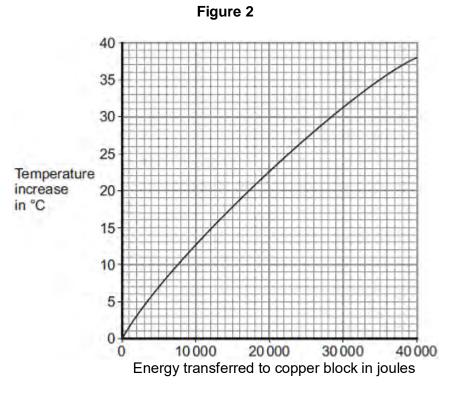
The power supply was switched on.

The energy transferred by the heater to the block was measured using the joulemeter.

The temperature of the block was recorded every minute.

The temperature increase was calculated.

Figure 2 shows the student's results.

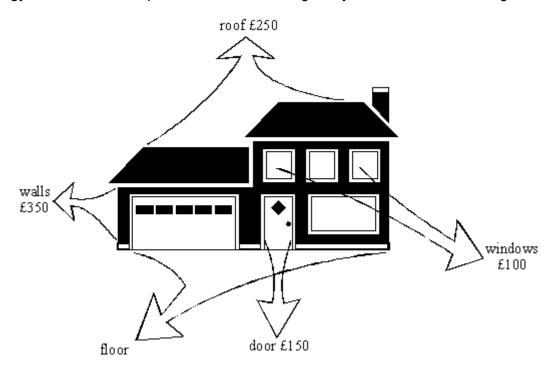


(a) Energy is transferred through the copper block.

What is the name of the process by which the energy is transferred?

Tick (✓) one box.	
Conduction	
Convection	
Radiation	
	(1)
Use Figure 2 to determine how much energy was needed to increase the temperature of the copper block by 35 °C.	
joules	(1)
	` ,
The copper block has a mass of 2 kg.	
Use your answer to part (b) to calculate the value given by this experiment for the specific heat capacity of copper. Give the unit.	
Specific heat capacity =	(2)
	(3)
This experiment does not give the correct value for the specific heat of copper.	
Suggest one reason why.	
(Total 6	(1)
	Convection Radiation Use Figure 2 to determine how much energy was needed to increase the temperature of the copper block by 35 °C. The copper block has a mass of 2 kg. Use your answer to part (b) to calculate the value given by this experiment for the specific heat capacity of copper. Give the unit. Specific heat capacity =

Q2. The diagram below shows a house which has **not** been insulated. The cost of the energy lost from different parts of the house during one year is shown on the diagram.



(a)	The total coet of the	e energy lost during	one year is £1000
141	THE IDIAL COST OF HI	e energy rost diffilia	One veal is a look

(i)	What is the cost of the energy lost through the floor?	
		(2)
		(4)

(11)	Suggest one way of reducing this loss.

(1)

(b) The table below shows how some parts of the house may be insulated to reduce energy losses. The cost of each method of insulation is also given.

WHERE LOST COST OF ENERGY LOST PER YEAR (£)	METOD OF INSULATION	COST OF INSULATION (£)
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roof	250	fibre-glass in loft	300
walls	350	foam filled cavity	800
windows	100	double glazing	4500
doors	150	draught proofing	5

(i)	Which method of insulation would you install first? Explain why.	
		(3)
		(-)
(ii)	Which method of insulation would you install last? Explain why.	
	/T-4-10	(3)
	(Total 9 i	narks)

Q3. The table gives information about some methods of conserving energy in a house.

Conservation method	Installation cost in £	Annual saving on energy bills in £
Cavity wall insulation	500	60
Hot water tank jacket	10	15
Loft insulation	110	60
Thermostatic radiator valves	75	20

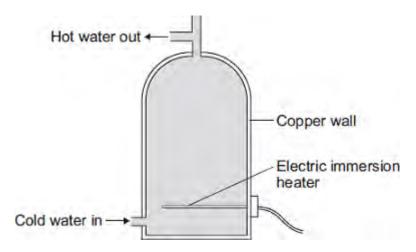
(a)	energy over a 10 year period. To obtain full marks you must support your answer with calculations.	
		(3)
		, ,
(b)	Describe what happens to the energy which is 'wasted' in a house.	
	(Total 5 n	(2)
	(10tal 5 II	11 a1 15)

Q4. (a) The table gives information about some ways of reducing the energy consumption in a house.

Method ofreducing energy consumption	Installation cost in £	Annual saving on energy bills in £
Fit a newhot water boiler	1800	200
Fit a solarwater heater	2400	100
Fitunderfloor heating	600	50
Fitthermostatic radiator valves	75	20

	Which way of reducing energy consumption is most cost effective over a 10-year period?	
	To obtain full marks you must support your answer with calculations.	
		(3)
		(0)
(b)	Explain why using an energy-efficient light bulb instead of an ordinary light bulb reduces the amount of carbon dioxide emitted into the atmosphere.	
		(2)
	(Total	5 marks)

Q5.An electric immersion heater is used to heat the water in a domestic hot water tank. When the immersion heater is switched on the water at the bottom of the tank gets hot.



(a) Complete the following sentence.

The main way the energy is transferred through the copper wall of the water tank is by

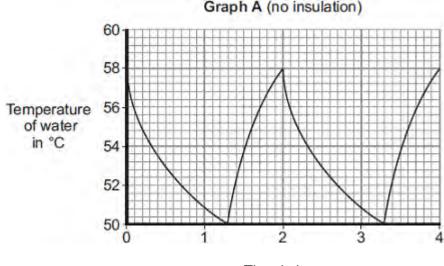
the process of

(1)

(b) The immersion heater has a thermostat to control the water temperature.

When the temperature of the water inside the tank reaches 58°C the thermostat switches the heater off. The thermostat switches the heater back on when the temperature of the water falls to 50°C.

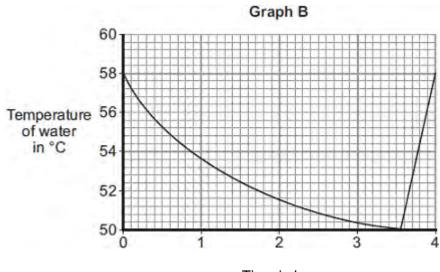
Graph A shows how the temperature of the water inside a hot water tank changes with time. The tank is **not** insulated.



Time in hours

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(i)	The temperature of the water falls at the fastest rate just after the heater switches off.	
	Explain why.	
		(2)
(ii)	To heat the water in the tank from 50°C to 58°C the immersion heater transfers 4032 kJ of energy to the water.	
	Calculate the mass of water in the tank.	
	Specific heat capacity of water = 4200 J/kg°C	
	Mass = kg	(3)
<i></i>		
(iii)	An insulating jacket is fitted to the hot water tank.	
	Graph B shows how the temperature of the water inside the insulated hot water tank changes with time.	

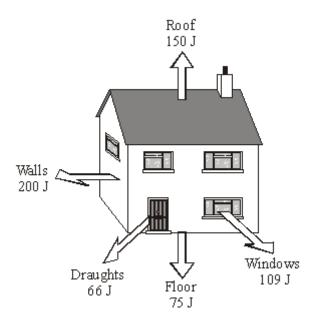


Time in hours

An insulating jacket only costs £12.

o a hot water tank saves money.	д јаскет
	(3) (Total 9 marks)

Q6. (a) The diagram shows how much heat is lost each second from different parts of an uninsulated house.



(i) Each year, the house costs £760 to heat.

How much money is being wasted because of heat lost through the roof?

Show clearly how you work out your answer.

(2)

(ii) Insulating the loft would cut the heat lost through the roof by 50 $\,\%$.

The loft insulation has a payback time of $1\frac{1}{2}$ years.

How much did the loft insulation cost to buy?

Cost of loft insulation = £

(1)

(b) What happens to the wasted energy?

(1) (Total 4 marks)