

Photoelectric Effect Questions

1. (a) The following equation describes the release of electrons from a metal surface illuminated by electromagnetic radiation.

$$hf = k.e._{\max} + \phi$$

Explain briefly what you understand by each of the terms in the equation.

hf

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$k.e._{\max}$

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ϕ

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(3)

- (b) Calculate the momentum p of an electron travelling in a vacuum at 5% of the speed of light.

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$$p = \dots\dots\dots$$

(3)

What is the de Broglie wavelength of electrons travelling at this speed?

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$\lambda =$

(2)

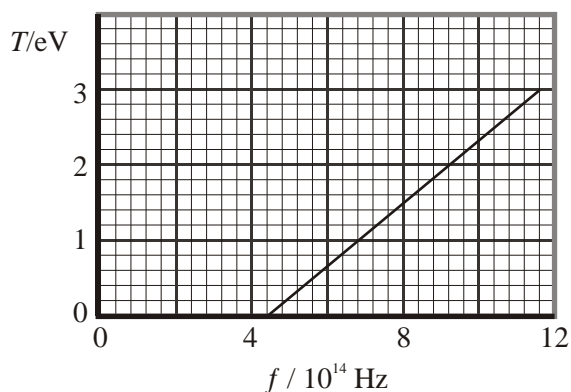
Why are electrons of this wavelength useful for studying the structure of molecules?

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(2)

(Total 10 marks)

2. The graph shows how the maximum kinetic energy T of photoelectrons emitted from the surface of sodium metal varies with the frequency f of the incident radiation.



Why are no photoelectrons emitted at frequencies below $4.4 \times 10^{14} \text{ Hz}$?

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(1)

Calculate the work function ϕ of sodium in eV.

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Work function =

(3)

Explain how the graph supports the photoelectric equation $hf = T + \phi$

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(2)

How could the graph be used to find a value for the Planck constant?

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(1)

Add a line to the graph to show the maximum kinetic energy of the photoelectrons emitted from a metal which has a greater work function than sodium.

(2)

(Total 9 marks)

3. Experiments on the photoelectric effect show that

- the kinetic energy of photoelectrons released depends upon the frequency of the incident light and not on its intensity,
- light below a certain threshold frequency cannot release photoelectrons.

How do these conclusions support a particle theory but not a wave theory of light?

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(6)

Calculate the threshold wavelength for a metal surface which has a work function of 6.2 eV.

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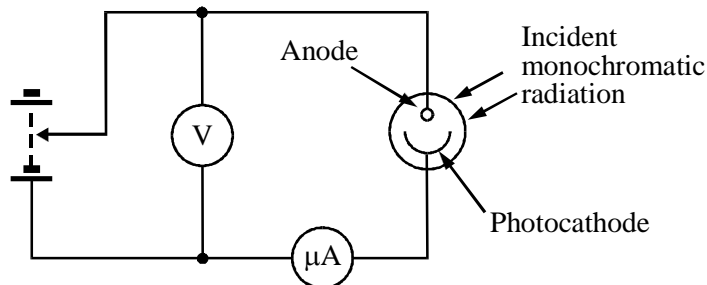
Threshold wavelength =

To which part of the electromagnetic spectrum does this wavelength belong?

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(4)
(Total 10 marks)

4. The diagram shows monochromatic light falling on a photocell.



As the reverse potential difference between the anode and cathode is increased, the current measured by the microammeter decreases. When the potential difference reaches a value V_s , called the stopping potential, the current is zero.

Explain these observations.

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(5)

What would be the effect on the stopping potential of

(i) increasing only the intensity of the incident radiation,

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(ii) increasing only the frequency of the incident radiation?

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(2)
(Total 7 marks)

5. Experiments on the photoelectric effect show that

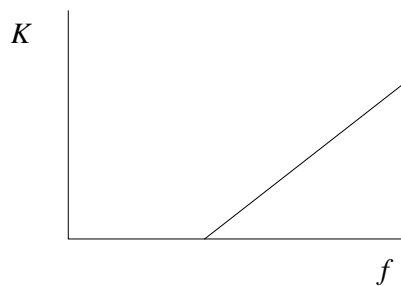
- the kinetic energy of photoelectrons released depends upon the frequency of the incident light and not on its intensity.
- light below a certain threshold frequency cannot release photoelectrons.

How do these conclusions support a particle theory but not a wave theory of light? You may be awarded a mark for the clarity of your answer.

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(6)

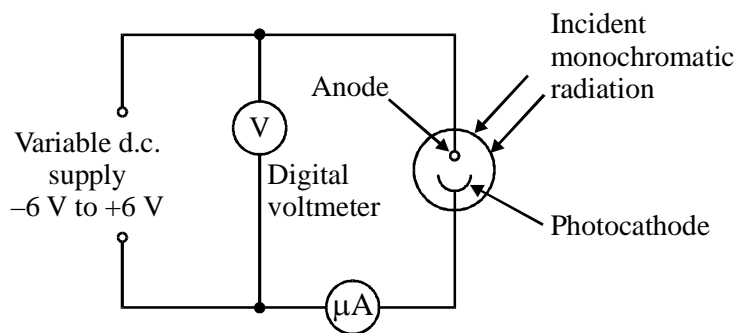
The graph shows how the kinetic energy K of emitted photoelectrons from one metal varies with the frequency f of the incident light.



Add a second line to the graph showing how K will vary with f for a second metal which has a *smaller* work function.

(2)
(Total 8 marks)

6. The diagram shows monochromatic radiation falling on a photocell connected to a circuit.



The incident radiation has a wavelength of 215 nm. The metal surface of the photocathode has a work function of 2.26 eV.

Calculate the energy in eV of a photon of the incident radiation.

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Energy = eV (4)

What is the maximum kinetic energy in eV of the emitted electrons?

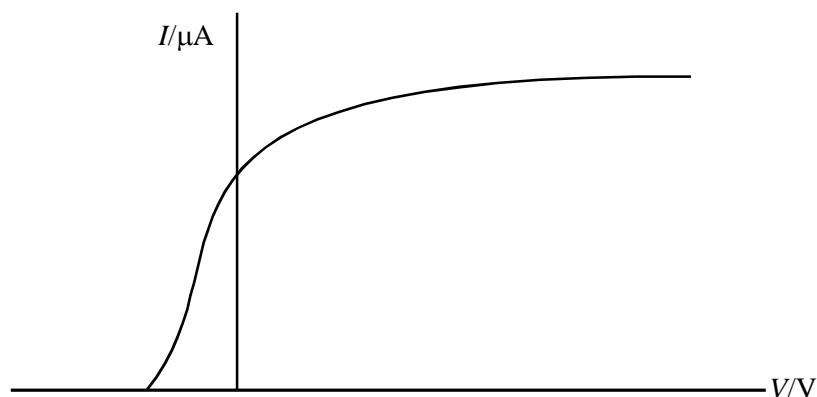
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Maximum k.e. = eV

Write down the value of the stopping potential.

Stopping potential = (2)

If the wavelength and intensity of the incident radiation is kept constant, a graph of the current I through the photocell against p.d. V is as shown.



Mark a letter S on the graph to show the stopping potential.

The photocathode is replaced with one whose metal surface has a greater work function. On the graph above, sketch how I would vary with V given that the wavelength and intensity of the incident radiation remain unchanged.

(3)
(Total 9 marks)

7. The photoelectric effect supports a particle theory of light but not a wave theory of light.

Below are two features of the photoelectric effect. For each feature explain why it supports the particle theory and not the wave theory.

(a) Feature 1: The emission of photoelectrons from a metal surface can take place instantaneously.

Explanation

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(2)

(b) Feature 2: Incident light with a frequency below a certain threshold frequency cannot release electrons from a metal surface.

Explanation

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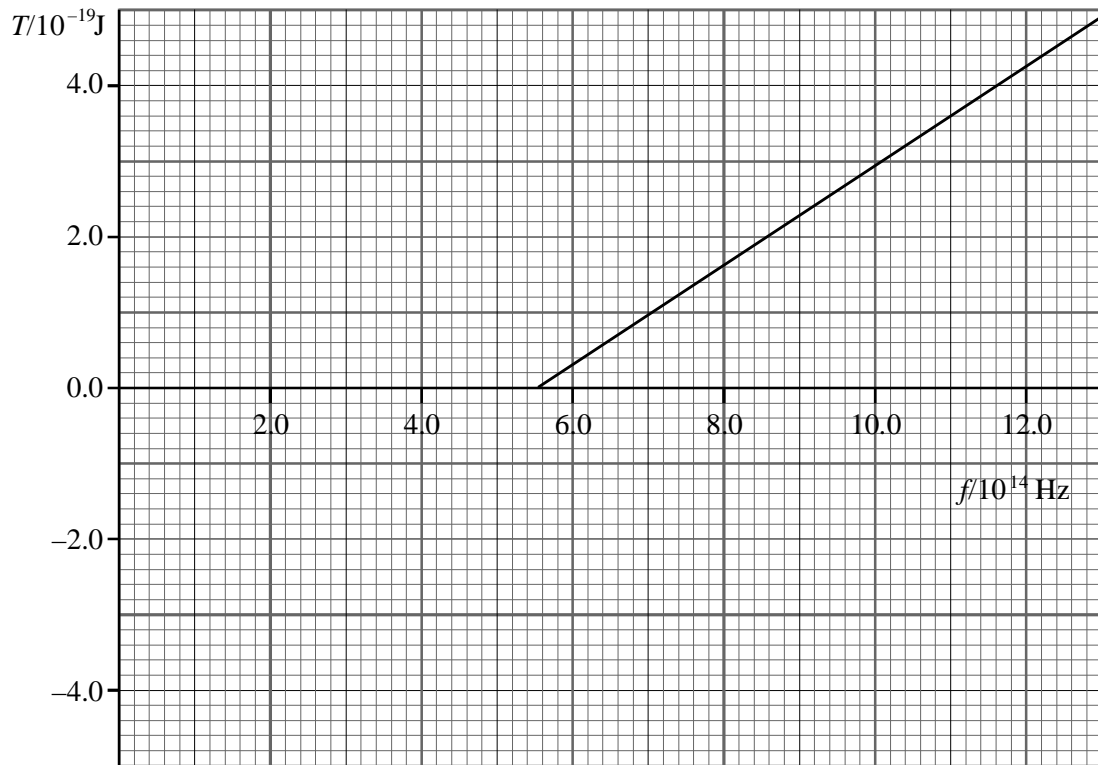
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(2)
(Total 4 marks)

8. The graph shows how the maximum kinetic energy T of photoelectrons emitted from the surface of sodium metal varies with the frequency f of the incident electromagnetic radiation.



Use the graph to find a value for the Planck constant.

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Planck constant =

(3)

Use the graph to find the work function ϕ of sodium metal.

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Work function =

(2)

Calculate the stopping potential when the frequency of the incident radiation is 9.0×10^{14} Hz.

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Stopping potential =

(3)

(Total 8 marks)