

| Question | | Answer | Marks | Guidance | |
|--------------|-----|--|--|--|---|
| 1 | (a) | The neutrons interact with other uranium (nuclei) / the neutrons cause further (fission) reactions | B1 | Not: neutrons interact with uranium <u>atoms</u> / <u>molecules</u> / <u>particles</u> | |
| | (b) | <p>Fuel rod: Contain the <u>uranium</u> (nuclei) / fissile material</p> <p>Control rods: Absorb (some of the) neutrons</p> <p><i>Controlled chain reaction:</i> The control rods are inserted into the reactor so as to allow (on average) one neutron from previous reaction to cause subsequent fission (AW)</p> <p>Moderator: Slows down the (fast-moving) neutrons / lowers the KE of (fast moving) neutrons / makes the (fast moving) neutrons into thermal neutrons</p> <p>Slow moving neutrons have a greater chance of causing fission / of being absorbed (by U-235) / sustaining chain reaction</p> | <p>B1</p> <p>B1</p> <p>B1</p> <p>B1</p> <p>B1</p> | <p>Show annotation on Scoris</p> <p>Not 'contains fuel'</p> <p>QWC mark</p> <p>Allow: Fast moving neutrons are captured (easily) by uranium-238 (nuclei leaving insufficient number of nuclei for fission / chain reaction) for the last B1 mark</p> | |
| | (c) | (i) | <p>power = $3.0 \times 10^9 / 0.22$</p> <p>power = 1.36×10^{10} (W) or 1.4×10^{10} (W)</p> | B1 | |
| | | (ii) | <p>energy = $1.36 \times 10^{10} \times 8.64 \times 10^4$</p> <p>energy = 1.18×10^{15} (J) or 1.2×10^{15} (J)</p> | B1 | Possible ecf from (c)(i) |
| | | (iii) | <p>(number of reactions per day) = $\frac{1.18 \times 10^{15}}{3.2 \times 10^{-11}}$</p> <p>mass = $\frac{1.18 \times 10^{15}}{3.2 \times 10^{-11}} \times 3.9 \times 10^{-25}$</p> <p>mass = 14.4 (kg) or 14 (kg)</p> | <p>C1</p> <p>A1</p> | <p>Possible ecf from (c)(ii)</p> <p>Note: Using 1.2×10^{15} (J) gives an answer of 14.6 (kg); allow 15 (kg)</p> |
| | (d) | Nuclear waste is (radio)active for a long time (AW) Causes ionisation | <p>B1</p> <p>B1</p> | Allow: 'Nuclear waste can have long half life' | |
| Total | | | 12 | | |

| Question | | Expected Answers | Marks | Additional guidance |
|--------------|---------|--|-------------------------------|---|
| | (c) (i) | <p>mass = $235 \times 1.7 \times 10^{-27}$ (= 3.995×10^{-25} kg)</p> <p>volume = $\frac{4}{3} \pi \times (8.8 \times 10^{-15})^3$ (= 2.855×10^{-42} m³)</p> <p>density = mass/volume</p> <p>density = 1.4×10^{17} (kg m⁻³)</p> | <p>C1</p> <p>C1</p> <p>A1</p> | <p>Allow: 1.66×10^{-27} kg for mass of nucleon</p> <p>Allow: 10^{17} (kg m⁻³) for this estimation question Note: Omitting 235 gives 6.0×10^{14} (kg m⁻³), allow 2 mark Allow: 1 mark if 92 or 143 is used to determine the mass of the nucleus; this gives a density value of 5.5×10^{16} (kg m⁻³) and 8.5×10^{16} (kg m⁻³) respectively</p> |
| | (ii) | The nucleons / neutrons and protons are packed together with little or no empty space (AW) | B1 | |
| Total | | | 14 | |

| Question | | Expected Answers | Marks | Additional guidance |
|--------------|-----|--|----------------------------------|--|
| 3 | (a) | A neutron is absorbed by a (massive / uranium) nucleus The nucleus splits into two (smaller/daughter) nuclei and (one or more) neutrons | B1 B1 | |
| | (b) | In a fission reaction there is a decreases in the mass (According to $\Delta E = \Delta mc^2$) mass is converted into energy Or The (total) binding energy of the products / smaller nuclei is greater than the binding energy of the original nucleus The difference in the binding energies is released as energy | M1 A1 M1 A1 | Allow: The 'BE increases (in the reaction)' |
| | (c) | Moderator: water / graphite / carbon It slows down the (fast-moving) neutrons / reduces the (kinetic) energy of neutrons Slow-moving neutrons have greater chance of causing fission (than fast-moving neutrons) | B1 B1 B1 | Note: If boron is mentioned, then do not award this B1 mark Allow: They become thermal neutrons |
| Total | | | 7 | |

| Question | | | Expected Answers | Marks | Additional Guidance |
|----------|---|---|---|----------------------------|---|
| 4 | a | (i) | $A = \lambda N_0 = 4.5 \times 10^{23} \times 0.693 / (12 \times 3600)$ $= 7.22 \times 10^{18} \text{ (s}^{-1}\text{)}$ | C1 A1 | allow one mark if the 12 hours is not converted into seconds. Answer is 2.6×10^{22} Allow one mark if the 12 hours is converted into minutes Answer 4.33×10^{20} |
| | | (ii) | 3 half lives $N = 5.6 \times 10^{22}$ | A1 | |
| | | (iii) | $N = N_0 e^{-\lambda t} = 4.5 \times 10^{23} \times e^{-(0.693 \times 50/12)}$ or use of 2^n $= 2.5 \times 10^{22}$ | C1 A1 | use of 2^n 50/12 half lives |
| | b | material with large λ / short half life have initial high activity hence precautions needed <u>for initial period</u> of disposal OR material with small λ / long half life activity will last for a long period hence need for long term disposal <div style="text-align: right;">MAX 2</div> | (B1) (B1) (B1) (B1) | | |
| | | Total | | [7] | |

| Question | | | Expected Answers | Marks | Additional Guidance |
|----------|---|------|---|--------------|--|
| 5 | a | (i) | e: 0 and -1 N: 15 and 7 + (antineutrino) | B1 | |
| | | (ii) | e: 0 and +1 Si: 30 and 14 + (neutrino) correct 'neutrino' <u>in each case</u> | B1 B1 | Allow 1 for +1 Correct symbols required for the neutrinos: ν and $\bar{\nu}$ Allow ν_e and $\bar{\nu}_e$ |
| | b | (i) | uud \rightarrow udd | B1 | Allow u \rightarrow d |
| | | (ii) | udd \rightarrow uud | B1 | Allow d \rightarrow u |
| | c | | weak(nuclear force) | B1 | |
| | | | Total | [6] | |