

Question Number	Answer	Mark
1(a)	A radioactive atom has an unstable nucleus which emits α , β , or γ radiation [at least one of α β γ named]	(1) (1) 2
1(b)	$C \rightarrow {}_{5}^{11}B + {}_{1}^{0}e^{+} + \nu_e$ Top line correct Bottom line correct	(1) (1) 2
1(c)	Attempt at mass difference calculation Attempt at conversion from (M)eV to J $\Delta E = 1.4 \times 10^{-13}$ (J) <u>Example of calculation:</u> $\Delta E = 10\,253.6 - 10\,252.2 - 0.5 = 0.889$ MeV $\Delta E = 0.889$ MeV $\times 1.6 \times 10^{-13}$ J MeV ⁻¹ = 1.42×10^{-13} J	(1) (1) (1) 3
1(d)	The idea that the sample will not produce radiation for very long (because carbon-11 has a relatively short half-life) β particles are not very ionising Or positrons are not very ionising Or boron is safe in small amounts	(1) (1) 2
1(e)	Use of $\lambda t_{1/2} = \ln 2$ ($\lambda = 5.68 \times 10^{-4} \text{ s}^{-1}$) Use of $A = A_0 e^{-\lambda t}$ Use $A = 1.58 \times 10^6$ Bq in $A = A_0 e^{-\lambda t}$ $A_0 = 1.2 \times 10^7$ Bq <u>Example of calculation:</u> $\lambda = \frac{0.693}{1220 \text{ s}} = 5.68 \times 10^{-4} \text{ s}^{-1}$ $1.58 \times 10^6 \text{ Bq} = A_0 e^{-5.68 \times 10^{-4} \text{ s}^{-1} \times 60 \times 60 \text{ s}}$ $A_0 = 1.22 \times 10^7 \text{ Bq}$	(1) (1) (1) (1) 4
Total for question		13

Question Number	Answer	Mark
2(a)	${}_{7}^{14}\text{N} + {}_{0}^{1}\text{n} \rightarrow {}_{6}^{12}\text{C} + {}_{1}^{3}\text{H}$ <p>Top line correct (1)</p> <p>Bottom line correct (1)</p>	2
2(b)(i)	<p>Background radiation would increase the count rate (by a constant amount)</p> <p>Or Background count rate has to be subtracted (from the activity) (1)</p>	1
2(b)(ii)	<p>Record the count for a long period of time</p> <p>Or Record the count more than once and find an average value (1)</p>	1
2(b)(iii)	<p>Use of $\lambda t_{1/2} = \ln 2$ (1)</p> <p>Use of $A = A_0 e^{-\lambda t}$ (1)</p> <p>Correct time identified (65 years) (1)</p> <p>$A_0 = 42 \text{ Bq}$ (1)</p> <p>Or</p> <p>Use of $A = \frac{A_0}{2^x}$ (1)</p> <p>Correct time identified (65 years) (1)</p> <p>Use of $x = \frac{t}{t_{1/2}}$ (1)</p> <p>$A_0 = 42 \text{ Bq}$ (1)</p> <p><u>Example of calculation</u></p> $\lambda = \frac{\ln 2}{t_{1/2}} = \frac{0.693}{12.3 \text{ year}} = 0.0563 \text{ year}^{-1}$ $A = A_0 e^{-\lambda t}$ $\therefore 1.08 \text{ Bq} = A_0 e^{-0.0563 \text{ year}^{-1} \times 65 \text{ year}}$ $A_0 = \frac{1.08 \text{ Bq}}{0.0257} = 42.1 \text{ Bq}$	4
2(c)(i)	<p>Mass difference calculation (1)</p> <p>Conversion to kg (1)</p> <p>Use of $\Delta E = c^2 \Delta m$ (1)</p> <p>$\Delta E = 2.8 \times 10^{-12} \text{ (J)}$ (1)</p> <p><u>Example of calculation</u></p> $\Delta m = (3.0155 + 2.0136) \text{ u} - (4.0015 + 1.0087) \text{ u} = 0.0189 \text{ u}$ $\Delta m = 0.0189 \text{ u} \times 1.66 \times 10^{-27} \text{ kg u}^{-1} = 3.14 \times 10^{-29} \text{ kg}$ $\Delta E = c^2 \Delta m = (3 \times 10^8 \text{ m s}^{-1})^2 \times 3.14 \times 10^{-29} \text{ kg} = 2.82 \times 10^{-12} \text{ J}$	4

2(c)(ii)	MAX 2		
	Very high temperatures [accept $T \sim 10^7$ K]	(1)	
	so that nuclei have sufficient energy to come close enough to overcome electrostatic repulsion [accept reference to strong interaction]	(1)	
	A collision rate large enough to sustain fusion (from a very high density)	(1)	2
	Total for Question		14

Question Number	Answer	Mark
3(a)		
(i)	Sun's position identified [single point identified]	(1)
(ii)	White dwarf region Red giant region	(1) (1)
		3
*3(a)(iii)	<p>(QWC – Work must be clear and organised in a logical manner using technical wording where appropriate)</p> <p>White dwarf stars have: high temperature T (because λ_{\max} is small) low luminosity L $L = \sigma AT^4$ linked to a determination of the surface area</p>	(1) (1) (1)
		3
3(b)	The star cools, so temperature T reduces The star contracts (under gravitational forces), so area A reduces $L = \sigma AT^4$ hence L is reduced (mark dependent upon either mp1 or mp2)	(1) (1) (1)
		3
3(c)(i)	${}^7_3\text{Li} + {}^1_1\text{X} \rightarrow 2 \times {}^4_2\text{He}$ <p>X is a proton [Accept X is hydrogen/H]</p>	(1) (1)
		2
3(c)(ii)	Attempt at calculation of mass difference Use of $1 \text{ MeV} = 1.60 \times 10^{-13} \text{ J}$ $\Delta E = 2.77 \times 10^{-12} \text{ (J)}$ <u>Example of calculation:</u> $\Delta m = 6533.8 \text{ MeV}/c^2 + 938.3 \text{ MeV}/c^2 - (2 \times 3727.4 \text{ MeV}/c^2) = 17.3 \text{ MeV}/c^2$ $\Delta E = 17.3 \text{ MeV}$ $\Delta E = 17.3 \text{ MeV} \times 1.60 \times 10^{-13} \text{ J MeV}^{-1} = 2.768 \times 10^{-12} \text{ J}$	(1) (1) (1)
		3

3(d)	<p>Max 4</p> <ul style="list-style-type: none"> • Extremely high temperature and density needed (1) • High temperature because nuclei need high <u>energy</u> to overcome the (electrostatic) repulsive force (1) • Since nuclei must come very close for fusion to occur Or since nuclei must come close enough for (strong) nuclear force to act (1) • Very high density is needed to maintain a sufficient collision rate (1) • Reference to extreme conditions leading to containment problems (1) 	4
Total for Question		18