Question number	Answer	Notes	Marks
1 (a)	mass surface area / size / volume	Accept weight Ignore length / width / height / thickness / shape / type Ignore temperature / purity / density	1 1
1 (b)	(2) time / how long (to collect gas)	Accept length of reaction/experiment	1
	(3) number of / how many (marble) chips	Ignore amount of marble Accept amount of (marble) <u>chips</u>	1
	(4) volume of gas / how much gas collected	Accept carbon dioxide/CO ₂ in place of gas Accept amount of gas	1
	(5) per tage / concentration (of acid)	Accept strength (of acid) Ignore volume of acid Reject amount (of acid)	1

Question number	Answer	Notes	Marks
1 (c) i	all six points plotted to nearest gridline	Deduct 1 mark for each error up to max 2, including extra points, but no penalty for point at 2.4 and 50%	2
	straight line of best fit	Line need not be extrapolated Must be drawn with a ruler CQ on candidate's plotted points	1
ii	anomalous point circled	at 1.5, 20 or CQ on candidate's line of best fit	1
iii	timer started late / stopped early	Ignore just wrong time	2
	concentration/% too high/more than 20%	Ignore just wrong concentration	
	marble chips bigger / more marble chips	Ignore just wrong mass/size	
	air in measuring cylinder before experiment started	Accept smaller volume of gas collected	
	(room) temperature higher	Ignore references to calculation Any two for 1 mark each	
		M1 to M5 CQ on position of anomalous point, so if anomalous point below line of best fit, then: M1 timer started too early/stopped too late	
		M2 concentration/% too low/less than 20% M3 marble chips smaller / fewer marble chips	
		M4 larger volume of gas collected M5 (room) temperature lower	
iv	vertical line from 50% to drawn graph line	Accept short vertical line crossing graph line at 50% Accept point on graph line at 50%	1
	2.4 (cm ³ /s)	CQ on drawn graph line but can be awarded without vertical line	1
		Do not penalise incorrect units Total	14

Question number	Answer	Accept	Reject	Marks
2 (a)	M1 (reactants) s aq	capital letters		1
	M2 (products) aq I g			1
(b) (i)	to prevent acid escaping/spraying out/spitting out IGNORE to prevent water escaping	solution/liquid/HCl		1
(ii)	С			1
(c) (i)	M1 A			1
	M2 gas produced/collected more quickly / experiment over in shorter time / (gradient of) line steeper	reaction is faster		1
	M2 dep on M1			1
(ii)	M1 0.1(0)	Half the products are		1
	M2 volume of gas is half/40 ÷ 80 = ½ / 80 = 40 x 2	produced		
	M2 dep on M1			
(d) (i)	M1 & M2 - all points plotted to nearest gridline deduct 1 mark			2
	for each incorrect plot up to a max. of 2			1
	M3 suitable straight line of best fit (csq on plotted points)			
(ii)	(must be drawn with the aid of a ruler). Line need not beextrapolated.	(show a) positive correlation		1
	M1 as concentration increases rate increases	as one doubles the other doubles/directly proportional		'

	M2 proportional / in proportion	for 2 marks		
(iii)			molecules/atoms	1
				1
	M1 more ions/particles (in a given volume) IGNORE more reactants			1
	M2 collide (successfully)			
	M3 more per second/more frequently		any reference to greater energy	
	Must be reference to frequency or number of collisions per unit time			
	I GNORE greater chance of collision			
			Total	16

Question number	Answer	Notes	Marks
3 a	mass / amount	Accept weight Ignore number of marble chips	1
	surface area / size / volume	Ignore length / width / height / thickness / shape / type Ignore temperature / purity / density	1
b	gas/carbon dioxide escapes / OWTTE	Ignore references to solid dissolving Ignore references to acid spray / vapour Do not penalise incorrectly named gas (eg hydrogen)	1
С	prevents loss of acid (spray)/liquid	Ignore references to evaporation / water vapour / spilling of liquid Reject references to stopping gases/marble chips escaping	1
d i	210 (s) some indication of mark on curve OR vertical line from 50% / horizontal line from 210 s	Mark M1 and M2 independently	1
ii	B (the loss of mass was greater than 1.0g		1
е	1 ÷ 210 evaluation of M1 / 0.00476	Accept any time value in range 200 - 210 Accept answer in range 0.004- 0.005 Accept any number of sig fig	1 1

Question number	Answer	Notes	Marks
3 f	all five points plotted to nearest gridline straight line of best fit	Deduct 1 mark for each error up to max 2 Line need not be extrapolated to origin If line not extrapolated, it should go to the origin if extrapolated Must be drawn with a ruler CQ on candidate's plotted points	2

Question number	Answer	Notes	Marks
3 g	more particles (in a given volume)	Accept ions Reject atoms / molecules Accept quantitative answer such as twice as many particles when concentration doubles	1
	more collisions (between particles) / OWTTE		1
	per unit time / OWTTE	more frequent collisions scores M2 and M3 Ignore greater chance/likelihood of collisions Accept reverse argument if clear that decreasing concentration is being considered MAX 1 if any reference to particles moving faster / having more energy	1
	L	Total 1	5 marks

Question number	Answer	Accept	Reject	Marks
4 (a) (i)	M1 & M2- all points correctly plotted to nearest gridline			2
	deduct 1 mark for each incorrectly plotted point			
	M3 smooth curve of best fit drawn			1
	170— 160— 150— X Time taken in seconds 140—			
	120- 110- 10 20 30 40 50 60 70 Temperature in °C			
	value from candidate's graph to nearest gridline			
(ii)	Penalise incorrect units			1
(iii)	as temperature increases, time (taken) decreases	reverse argument		1
	IGNORE references to rate and inverse proportionality	negative correlation		

Question number	Answer	Accept	Reject	Marks
4 (b)	 M1 (average kinetic) energy of particles/ions increases M2 more collisions/particles/ions have energy ≥ activation energy M3 more (successful) collisions per second / more frequent (successful) collisions IGNORE references to chance of collisions Penalise reference to molecules once only 	particles move faster sufficient energy to react	molecules/atoms (but once only)	1 1 1
(c)	(same) concentration (of each solution)	(same) volume (of each solution) (same) amount of (each) solution rate of mixing		1
			Total	9