- 1 What is used in the production of ethanol from ethene?
 - A hydrogen and oxygen
 - B oxygen only
 - C steam
 - D yeast
- 2 Which term describes the formation of ethanol from glucose?
 - A cracking
 - **B** distillation
 - **C** fermentation
 - D polymerisation
- 3 Which statement about ethanol is correct?
 - A It burns in air to form ethene and water.
 - **B** It is prepared from ethene by fermentation.
 - **C** It is prepared from glucose in an addition reaction.
 - **D** It is the only product when ethene reacts with steam.
 - 4 Ethanol can be formed by
 - 1 fermentation
 - 2 reaction between steam and ethene

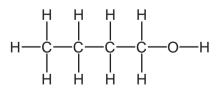
Which of these processes uses a catalyst?

	1	2
Α	\checkmark	\checkmark
в	\checkmark	x
С	x	\checkmark
D	X	x

5 The flow chart shows the preparation of ethanol and some important chemistry of ethanol.

substa	ince X ferment	ethanol	ocess Y ► car	bon dioxide + substance Z
What are 2	X, Y and Z?			
	х	Y	Z	
A	ethane	combustion	yeast	
в	glucose	combustion	steam	
С	water	polymerisation	water	
D	yeast	fermentation	glucose	

- 6 By which of the following methods is ethanol formed?
 - 1 fractional distillation of petroleum
 - 2 fermentation
 - 3 catalytic addition of steam to ethene
 - A 1 and 2 only
 - B 1 and 3 only
 - C 2 and 3 only
 - **D** 1, 2 and 3
- 7 An organic compound has the molecular structure shown.



Which type of organic compound is this?

- A alcohol
- B alkane
- C alkene
- D carboxylic acid

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	can be made from glucose	can be made from ethene	is used as a fuel	is used as a solvent
A	1	1	1	1
в	1	x	5	1
С	x	1	1	x
D	x	1	x	1

8 Which row correctly describes the production of ethanol and its properties?

- 9 Which statement about the names of organic compounds is correct?
 - **A** Compounds containing C=C double bonds are alkanes.
 - **B** The compound of formula CH_3CO_2H is methanoic acid.
 - **C** The compound of formula C_2H_4 is ethane.
 - **D** The compound of formula C_2H_5OH is an alcohol.
- 10 Ethanol is manufactured from petroleum by reacting ethene with steam.

Which statements about this process are correct?

- 1 Ethene is obtained from the cracking of alkanes.
- 2 The process is carried out in the presence of yeast.
- 3 The reaction is an addition reaction.
- 4 The rate of reaction is increased by a catalyst.

A 1 and 3 only **B** 1 and 4 only **C** 1, 2 and 3 **D** 1, 3 and 4

11 Ethene reacts with Y to produce ethanol.

ethene + Y \rightarrow ethanol

What is Y?

- A hydrogen
- B oxygen
- C steam
- D yeast
- 12 Ethanol is made by fermentation.

How is ethanol obtained from the fermentation mixture?

- A chromatography
- B crystallisation
- C electrolysis
- D fractional distillation
- 13 Ethanol can be manufactured from substance X.

substance X + steam _____ ethanol

What is substance X?

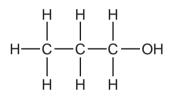
- A carbon dioxide
- B ethene
- C hydrogen
- D oxygen

14 The equation shows an industrial process.

 $H_2O + C_2H_4 \longrightarrow compound X$

What is the name of compound X?

- A ethane
- B ethanoic acid
- C ethanol
- D methanol
- 15 Which type of compound is shown?



- A alcohol
- B alkane
- C alkene
- D carboxylic acid
- 16 Which statements about ethanol are correct?
 - 1 It can be made by fermentation.
 - 2 It is an unsaturated compound.
 - 3 It burns in air and can be used as a fuel.
 - A 1, 2 and 3 B 1 and 2 only C 1 and 3 only D 2 and 3 only

17 Ethanol is produced from either ethene or sugar.

	ethene \rightarrow ethanol	sugar $ ightarrow$ ethanol
Α	addition	fermentation
в	addition	fractional distillation
С	distillation	fermentation
D	distillation	fractional distillation

Which type of chemical reaction is used in each case?

18 The table shows the boiling points of four members of the homologous series of alcohols.

comp	boiling point	
name	formula	/°C
methanol	CH ₃ OH	65
ethanol	C ₂ H ₅ OH	78
propanol	C ₃ H ₇ OH	х
butanol	C ₄ H ₉ OH	117

What is the value of X?

Α	55°C	В	82°C	С	98°C	D	115°C
		_		-		_	

- 19 Which of the statements about ethanol are correct?
 - 1 Ethanol can be formed by an addition reaction.
 - 2 Ethanol can be formed by fermentation.
 - 3 When ethanol burns in air, it forms carbon dioxide and water.
 - **A** 1, 2 and 3 **B** 1 and 2 **C** 1 and 3 **D** 2 and 3

- 20 Which process is **not** used during the production of ethanol?
 - A addition of steam to ethene
 - **B** fermentation
 - **C** fractional distillation
 - **D** reacting ethane with oxygen

21 Which row is correct for ethanol?

	burns	made by fermentation
Α	\checkmark	✓
в	\checkmark	x
С	x	\checkmark
D	X	x

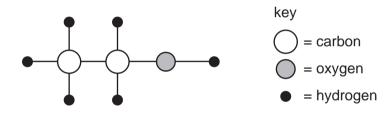
- 22 Which statement about ethanol is not correct?
 - A It can be used as a fuel.
 - **B** It can be used as a solvent.
 - **C** It is formed by reaction between steam and ethane.
 - **D** Its functional group is –OH.

23 Ethanol is an important chemical produced by the1..... of2.

	1	2	
A	combustion	ethane	
в	combustion	glucose	
С	fermentation	ethane	
D	fermentation	glucose	

Which words correctly complete gaps 1 and 2?

24 The diagram represents the molecule of an organic compound.

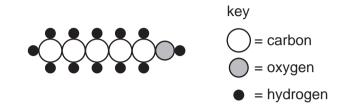


What is the name of the compound?

- A ethane
- B ethanoic acid
- C ethanol
- D ethene
- 25 When glucose is fermented, ethanol is formed together with
 - A carbon dioxide.
 - **B** ethene.
 - **C** methane.
 - D oxygen.

26 Compounds containing five carbon atoms in a molecule may have names beginning with 'pent...'.

What is the name of the compound shown?



- A pentane
- B pentanoic acid
- **C** pentanol
- D pentene
- 27 What is **not** essential for the formation of ethanol by fermentation?
 - A light
 - B sugar
 - C yeast
 - D water
- 28 Ethene reacts with Y to produce ethanol.

ethene + Y \rightarrow ethanol

What is Y?

- A hydrogen
- **B** oxygen
- **C** steam
- D yeast