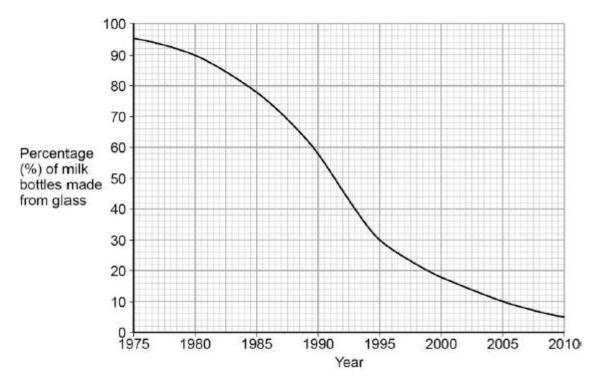
Q1. Plastic and glass can be used to make milk bottles.

The figure below shows the percentage of milk bottles made from glass between 1975 and 2010.



- (a) Plot the points and draw a line on the figure above to show the percentage of milk bottles made from materials **other** than glass between 1975 and 2010.
- (b) The table below gives information about milk bottles.

	Glass milk bottle	Plastic milk bottle
Raw materials	Sand, limestone, salt	Crude oil
Bottle material	Soda-lime glass	HD poly(ethene)
Initial stage in production of bottle material	Limestone and salt used to produce sodium carbonate.	Production of naphtha fraction.
Maximum temperature in production process	1600 °C	850 °C
Number of times bottle can be used for milk	25	1

(3)

Size(s) of bottle	0.5 dm ³	0.5 dm ³ , 1 dm ³ , 2 dm ³ , 3 dm ³
Percentage (%) of recycled material used in new bottles	50 %	10 %

Evaluate the production and use of bottles made from soda-lime glass and those made from HD poly(ethene).

Use the information given and your knowledge and understanding to justify your ch material for milk bottles.	oice of
	(6) (Total 9 marks)

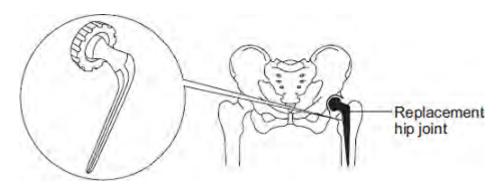
	a) E	xplain how electricity is conducted in a metal.
Describe how the structure of an alloy is different from the structure of a pure metal. Alloys are used to make dental braces and coins. (i) Nitinol is an alloy used in dental braces.	Т	o gain full marks you must include a description of the structure and bonding of a metal.
b) Describe how the structure of an alloy is different from the structure of a pure metal.		
b) Describe how the structure of an alloy is different from the structure of a pure metal. c) Alloys are used to make dental braces and coins. (i) Nitinol is an alloy used in dental braces.		
b) Describe how the structure of an alloy is different from the structure of a pure metal. c) Alloys are used to make dental braces and coins. (i) Nitinol is an alloy used in dental braces.		
b) Describe how the structure of an alloy is different from the structure of a pure metal. c) Alloys are used to make dental braces and coins. (i) Nitinol is an alloy used in dental braces.		
c) Alloys are used to make dental braces and coins. (i) Nitinol is an alloy used in dental braces.		
(c) Alloys are used to make dental braces and coins. (i) Nitinol is an alloy used in dental braces.		
c) Alloys are used to make dental braces and coins. (i) Nitinol is an alloy used in dental braces.		
(i) Nitinol is an alloy used in dental braces.		
(i) Nitinol is an alloy used in dental braces.	b) D	rescribe how the structure of an alloy is different from the structure of a pure metal.
(i) Nitinol is an alloy used in dental braces.	b) D 	escribe how the structure of an alloy is different from the structure of a pure metal.
(i) Nitinol is an alloy used in dental braces.	b) D 	rescribe how the structure of an alloy is different from the structure of a pure metal.
(i) Nitinol is an alloy used in dental braces.	b) D	rescribe how the structure of an alloy is different from the structure of a pure metal.
	b) D	escribe how the structure of an alloy is different from the structure of a pure metal.
Why is Nitinol used in dental braces?		
	 	lloys are used to make dental braces and coins.
	 	lloys are used to make dental braces and coins.) Nitinol is an alloy used in dental braces.

(ii) Suggest **one** reason why coins are not made of pure copper.

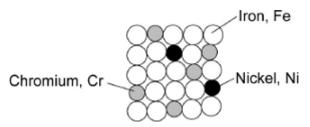
		Do not give cost as a reason.	
			(1)
	(iii)	Some coins are made from an alloy of aluminium.	
		Complete the sentence.	
		Aluminium is manufactured by the electrolysis of a molten mixture of cryolite and	
			(1)
	(iv)	Banks keep coins in poly(ethene) bags. These bags are made from low density poly(ethene).	
		High density poly(ethene) can also be made from the same monomer.	
		How can the same reaction produce two different products?	
			(1)
(d)	Give meta	two reasons why instrumental methods of analysis are used to detect impurities in als.	
			(1)
		(Total 11 ma	` '

Q3.The hip joint sometimes has to be replaced.

Early replacement hip joints were made from stainless steel.



Stainless steel is an alloy of iron, chromium and nickel. The diagram below represents the particles in stainless steel.



Paticle diagram of stainless steal

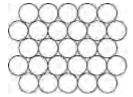
(a) Use the diagram to complete the percentages of metals in this stainless steel.

The first one has been done for you.

Element	Percentage (%)
Iron, Fe	72
Chromium, Cr	
Nickel, Ni	

(2)

(b) Pure iron is a soft, metallic element.



(i)	Why is iron described as an <i>element</i> ?	
		(1)
/::\	Done in a consulation to the control of the consulation in the	
(ii)	Pure iron would not be suitable for a replacement hip joint.	
	Suggest why.	
		(1)
		(-)
(iii)	The three metals in stainless steel have different sized atoms. Stainless steel is harder than pure iron.	
	Explain why.	
		(2)
		(Total 6 marks)

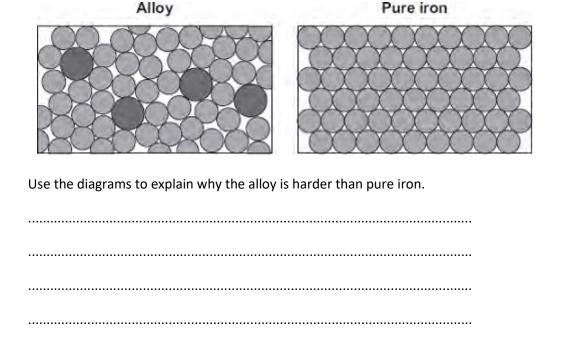
Q4.Oil rigs are used to drill for crude oil.



© Digital Vision/Photodisc

(a) Drills are made from an alloy of iron.

The diagrams show the particles in the alloy and in pure iron.



(b) Drill heads contain diamonds.

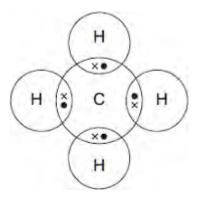
Tick (**✓**) **two** reasons why diamonds are hard.

Reason	Tick (✓)
--------	----------

Diamonds have a giant covalent structure.	
Diamonds have high melting points.	
Diamonds are unreactive.	
Diamonds have strong bonds between carbon atoms.	

(2)

(c) Methane gas is often found where crude oil is found.The diagram shows how atoms bond in methane.Only the outer electrons are shown.



(i) Draw a ring around the correct answer to complete the sentence.

a compound.

Methane is an element.

a mixture.

(1)

(ii) Draw a ring around the correct answer to complete each sentence.

 $\begin{array}{c} C_4H_4 \\ \\ The \ formula \ of \ methane \ is \end{array}$ C_4H C_4H_4

	(iii) Name the type of bond between the carbon and hydrogen atoms in methanol	€.	
		(1	1)
(d)	Explain why methane is a gas at 20°C.		
		(2 (Total 9 marks)	2) s)

(1)

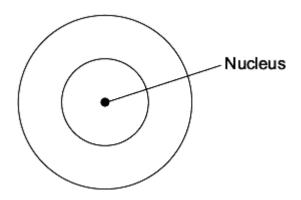
Q5.The picture shows a diamond ring.



Photograph supplied by Comstock/Thinkstock

(a) Diamond is a form of carbon. A carbon atom has six electrons.

Draw the electronic structure of a carbon atom.



(1)

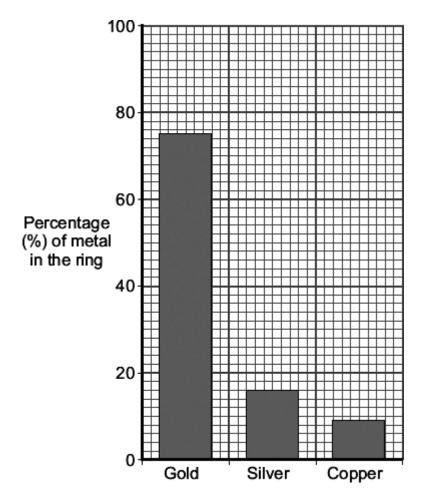
(b) A gold atom has an atomic number of 79 and a mass number of 197.

Complete the table to show the name and number of each sub-atomic particle in this gold atom.

Name	Number
Proton	79
Electron	

(3)

(c) The bar chart shows the composition of this gold ring.



(i) Give the percentage of the other two metals in this gold ring.

Silver is % and copper is %

(ii) This gold ring is not made from 100% gold.

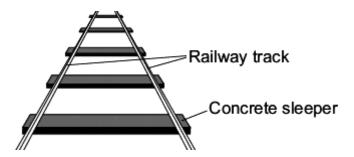
Give two reasons why.

(1)

Page 12

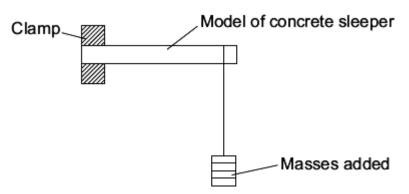
(2)
(Total 7 marks)

Q6. In the UK, railway sleepers were made from wood. They are now often made from concrete.



A scientist was asked to find the best concrete mixture to use so that railway sleepers would not break easily. The scientist made:

- a mould to make small models of concrete sleepers
- concrete mixtures using crushed rock, sand, cement and water
- the equipment shown to add 0.1 kg masses until the model sleeper broke.



The scientist's results are shown in the table.

Concrete mixture in % by volume			Total mass added to break the model sleeper in kg				
Cement	Sand	Crushed rock	Test 1	Test 2	Test 3	Test 4	Mean
10	70	20	1.2	1.1	1.3	1.2	1.2
20	60	20	3.0	2.6	2.5	2.4	
30	50	20	3.5	3.3	3.3	3.3	3.3
40	40	20	3.9	3.8	4.0	3.3	3.9
50	30	20	4.2	4.5	4.2	4.3	4.3

	Mean = kg
State	e one conclusion that the scientist could make from these results.
	scientist sent the results in a report to a company that makes full-size concrete railway pers.
SICC	5013.
	Suggest two other factors that the company should take into consideration before deciding which mixture to use to make a full-size concrete railway sleeper.
	Suggest two other factors that the company should take into consideration before
	Suggest two other factors that the company should take into consideration before deciding which mixture to use to make a full-size concrete railway sleeper.
	Suggest two other factors that the company should take into consideration before deciding which mixture to use to make a full-size concrete railway sleeper.
(i)	Suggest two other factors that the company should take into consideration before deciding which mixture to use to make a full-size concrete railway sleeper.
	Suggest two other factors that the company should take into consideration before deciding which mixture to use to make a full-size concrete railway sleeper.

(4)
(4)
(Total 9 marks)