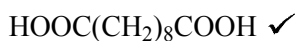




*ALLOW*  $\text{H}_2\text{NCH}_2\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_2\text{NH}_2$



*ALLOW*  $\text{HOOCCH}_2\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_2\text{COOH}$

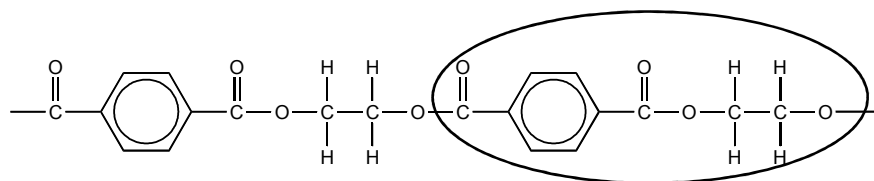
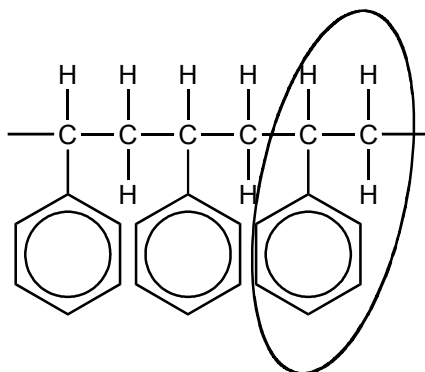
*ALLOW*  $\text{CO}_2\text{H}$  for  $\text{COOH}$

*ALLOW* acid chloride,  $\text{ClOC}(\text{CH}_2)_8\text{COCl}$

*ALLOW* displayed formulae or skeletal formulae

[2]

2. (a) (i)



1 mark for each repeat unit (1)(1)

2

(ii) 1 mark for each monomer (1)(1)

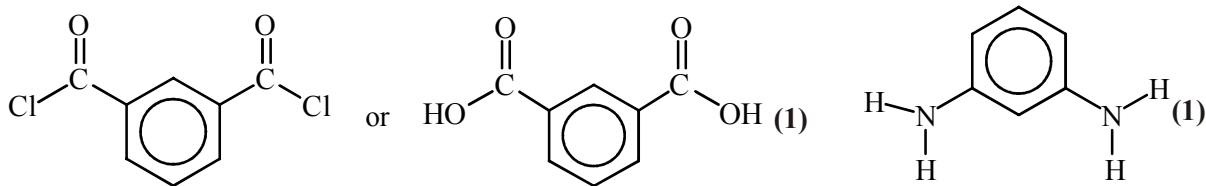
2

(b)  $\text{C}=\text{O}$  absorbs radiation/breaks (1)  
ester linkage hydrolysed (1)

2

[6]

3. (i)



2

- (ii) any valid suggestion to explain or describe stronger intermolecular forces – e.g. Nomex is planar so packs together more easily / greater H-bonding / Van der Waals' / forces between molecules **(1)** AW  
(ignore arguments based on Mr)

1

[3]

4. (a)

$\begin{array}{c} \text{HO}-\overset{\text{O}}{\parallel}{\text{C}}-(\text{CH}_2)_4-\overset{\text{O}}{\parallel}{\text{C}}-\text{OH} \\ \text{H}_2\text{N}-(\text{CH}_2)_e-\text{NH}_2 \end{array}$	$\begin{array}{c} \text{H} \quad \text{CN} \\ \diagdown \quad / \\ \text{C}=\text{C} \\ / \quad \diagdown \\ \text{H} \quad \text{H} \end{array} \quad \text{(1)}$
$\left[ \begin{array}{c} \text{O} \quad \quad \text{O} \\ \parallel \quad \quad \parallel \\ -\text{C}-\text{(CH}_2)_4-\text{C}-\text{N}-\text{(CH}_2)_6-\text{N}- \\   \quad \quad \quad   \\ \text{H} \quad \quad \quad \text{H} \end{array} \right]$ <p>monomers connected by NHCO <b>(1)</b> correct repeat shown <b>(1)</b></p>	$\left[ \begin{array}{c} \text{H} \quad \text{CN} \\   \quad   \\ -\text{C}-\text{C}- \\   \quad   \\ \text{H} \quad \text{H} \end{array} \right]$
condensation	addition

**(1)** for both

4

(b) (i)  $\text{PCl}_5 / \text{SOCl}_2$

1

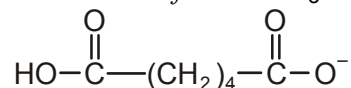
(ii)  $\text{HCl}$

1

(c)  $\text{H}_3\text{N}^+-(\text{CH}_2)_6-\text{NH}_3^+ \quad \text{(1)} \quad ^-\text{O}-\overset{\text{O}}{\parallel}{\text{C}}-(\text{CH}_2)_4-\overset{\text{O}}{\parallel}{\text{C}}-\text{O}^-$

2

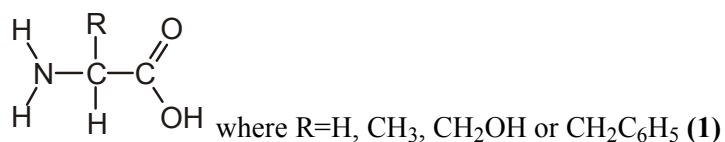
allow 1 mark for: both  $\text{H}_3\text{N}^+-(\text{CH}_2)_6-\text{NH}_3^+$  and



(d) (i) 4

1

(ii)



1

(iii) any three different chemically or biologically correct differences between amino acids and the nylon monomers **(1)(1)(1)** - eg

- protein monomers are amino acids / nylon monomers are a (di)amine/base and a (di)acid
- protein monomers have different types/R groups / nylon monomers are two types/no variation
- protein monomers have stereo/optical isomers/are chiral
- protein monomers have higher melting points/ form zwitterions

other possible answers include:

- nylon monomers have longer chain length/no other functional groups / no aromatic content / are symmetrical etc  
*don't allow comparisons solubility or  $M_r$*

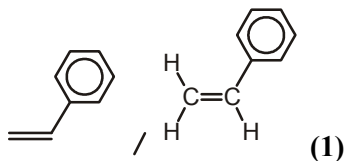
3

**[13]**

5. (i) addition (polymerisation) **(1)**  
*NOT additional*

1

(ii)



1

(iii)  $\pi$ -bond breaks **(1)**

**many** molecules join / a **long** chain forms /  
equation to show this using 'n' **(1)**

2

**[4]**