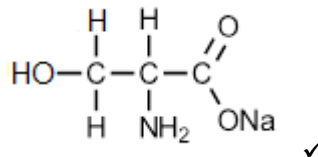
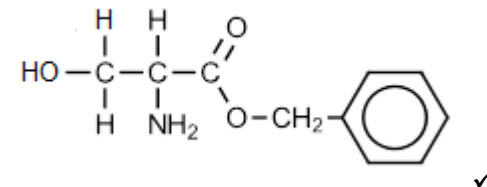
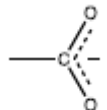
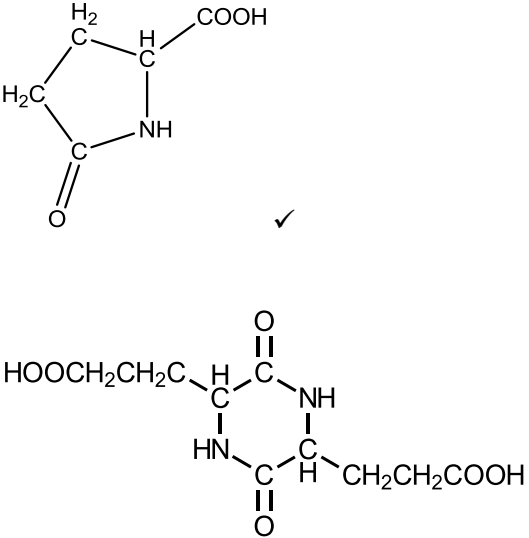
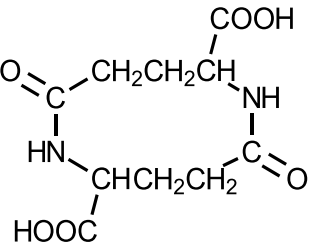
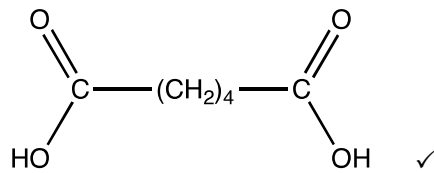
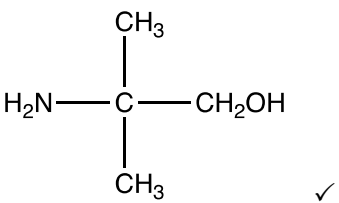
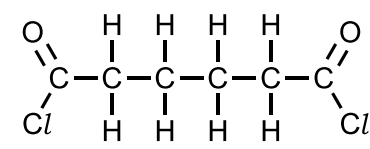


Question			Answer	Mark	Guidance
1	(a)	(i)	  —NH <sub>3</sub> <sup>+</sup> in second product ✓	3	<b>ALLOW</b> correct structural <b>OR</b> displayed <b>OR</b> skeletal formulae <b>OR</b> a combination of above as long as unambiguous <b>ALLOW</b> —O <sup>-</sup> Na <sup>+</sup> <b>OR</b> —O <sup>-</sup> (cation not required) <b>DO NOT ALLOW</b> —O—Na (covalent bond) <b>DO NOT ALLOW</b> —O (without the sodium) <b>ALLOW</b> delocalised carboxylate 
		(ii)	perfume/fragrance/flavouring ✓	1	<b>IGNORE</b> solvent <b>OR</b> food additive
		(iii)	Reaction 3: (hot) ethanolic ammonia ✓  Reaction 4: oxidation ✓  Reaction 5: hydrolysis ✓	3	<b>ALLOW</b> NH <sub>3</sub> (dissolved) in ethanol <b>IGNORE</b> other conditions  <b>ALLOW</b> oxidisation/oxidised <b>DO NOT ALLOW</b> redox  <b>ALLOW</b> nucleophilic addition-elimination <b>DO NOT ALLOW</b> nucleophilic substitution <b>IGNORE</b> acid/base

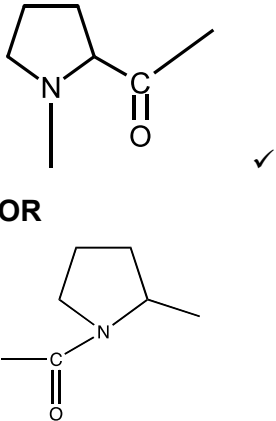
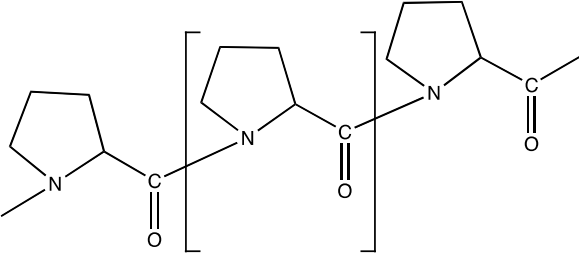
Question	Answer	Mark	Guidance
(b)	<p><b>M1</b> Compound E</p> $\begin{array}{c} \text{H} \quad \text{H} \\   \quad   \\ \text{H}_2\text{C}=\text{C}-\text{C}-\text{CHO} \\   \\ \text{NH}_2 \end{array}$ <p><b>M2</b> Compound F</p> $\begin{array}{c} \text{H} \quad \text{H} \\   \quad   \\ \text{H}_2\text{C}=\text{C}-\text{C}-\text{COOH} \\   \\ \text{NH}_2 \end{array}$ <p><b>M3</b> Compound G</p> $\begin{array}{c} \text{H} \quad \text{H} \\   \quad   \\ \left[ \text{C}-\text{C} \right] \\   \quad   \\ \text{H} \quad \text{CHNH}_2 \\ \quad \quad   \\ \quad \quad \text{COOH} \end{array}$ <p><b>M4</b> Compound H</p> $\left[ \begin{array}{c} \text{H} \quad \text{O} \\   \quad    \\ \text{N}-\text{C}-\text{C} \\   \quad   \\ \text{H} \quad \text{CH}=\text{CH}_2 \end{array} \right]$	6	<p><b>ANNOTATE ANSWER WITH TICKS AND CROSSES ETC</b></p> <p><b>ALLOW</b> correct structural <b>OR</b> displayed <b>OR</b> skeletal formulae <b>OR</b> a combination of above as long as unambiguous</p> <p>Labels are not required for compound E, F, G or H</p> <p><b>IGNORE</b> labels for <b>M1</b>, <b>M2</b>, <b>M3</b> and <b>M4</b></p> <p>CH<sub>2</sub>=CH must be shown in <b>E</b></p> <p><b>ALLOW</b> C<sub>2</sub>H<sub>3</sub> <b>OR</b> CHCH<sub>2</sub> for CH=CH<sub>2</sub> in <b>F</b></p> <p><b>ALLOW ECF</b> from error in structure of <u>aldehyde</u> E</p> <p><b>ALLOW</b> multiple repeat units but must be full repeat units</p> <p><b>ALLOW</b> end bonds shown as .....</p> <p><b>DO NOT ALLOW</b> if structures have no end bonds</p> <p><b>IGNORE</b> brackets unless they are used to pick out the repeat unit from a polymer chain</p> <p><b>IGNORE</b> n</p> <p><b>ALLOW</b> C<sub>2</sub>H<sub>4</sub>NO<sub>2</sub> for CH(NH<sub>2</sub>)COOH in polymer <b>G</b></p> <p><b>ALLOW</b> C<sub>2</sub>H<sub>3</sub> <b>OR</b> CHCH<sub>2</sub> for CH=CH<sub>2</sub> in polymer <b>H</b></p> <p><b>ALLOW ECF</b> from NH<sub>2</sub>CH<sub>2</sub>CH=CHCOOH for the formation of compound G or compound H</p>

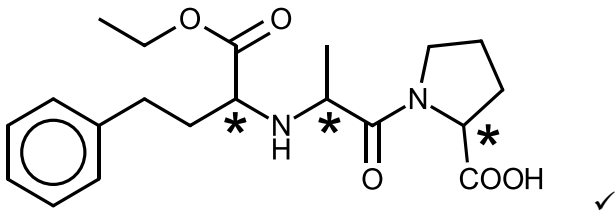
Question	Answer	Mark	Guidance
	<p><b>M5</b> Compound G <b>OR</b></p> $\left[ \begin{array}{c} \text{H} \quad \text{H} \\   \quad   \\ -\text{C}-\text{C}- \\   \quad   \\ \text{H} \quad \text{CHNH}_2 \\ \quad \quad   \\ \quad \quad \text{COOH} \end{array} \right]$ <p>Is an addition polymer ✓</p> <p><b>M6</b> Compound H <b>OR</b></p> $\left[ \begin{array}{c} \text{H} \quad \text{O} \\   \quad    \\ -\text{N}-\text{C}-\text{C}- \\   \quad   \\ \text{H} \quad \text{CH}=\text{CH}_2 \end{array} \right]$ <p>is a condensation polymer ✓</p>		<p><b>ALLOW</b> alkene forms addition polymer/polymer with same empirical formula as monomer</p> <p><b>ALLOW</b> equation for reaction</p> $n \text{ H}_2\text{C}=\begin{array}{c} \text{H} \quad \text{H} \\   \quad   \\ \text{C}-\text{C}-\text{COOH} \\   \\ \text{NH}_2 \end{array} \longrightarrow \left[ \begin{array}{c} \text{H} \quad \text{H} \\   \quad   \\ -\text{C}-\text{C}- \\   \quad   \\ \text{H} \quad \text{CHNH}_2 \\ \quad \quad   \\ \quad \quad \text{COOH} \end{array} \right]_n$ <p><b>ALLOW</b> amino acid forms condensation polymer</p> <p><b>OR</b> (molecules of) compound F join/bond/add/react/form polymer and water/small molecule</p> <p><b>ALLOW</b> equation for reaction</p> $n \text{ H}_2\text{C}=\begin{array}{c} \text{H} \quad \text{H} \\   \quad   \\ \text{C}-\text{C}-\text{COOH} \\   \\ \text{NH}_2 \end{array} \longrightarrow \left[ \begin{array}{c} \text{H} \quad \text{O} \\   \quad    \\ -\text{N}-\text{C}-\text{C}- \\   \quad   \\ \text{H} \quad \text{CH}=\text{CH}_2 \end{array} \right]_n + \text{H}_2\text{O}$
(c) (i)	$\begin{array}{ccccccccc} & \text{H} & \text{H} & \text{H} & \text{H} & \text{O} & & & & \\ &   &   &   &   &    & & & & \\ \text{H} & -\text{N} & -\text{C} & -\text{C} & -\text{C} & -\text{C} & -\text{OH} & & & \\ & &   &   &   & & & & & \\ & & \text{HOOC} & \text{H} & \text{H} & & & & & \end{array}$ <p style="text-align: right;">✓</p>	1	<p><b>ALLOW</b> correct structural <b>OR</b> displayed <b>OR</b> skeletal formulae</p> <p><b>OR</b> a combination of above as long as unambiguous</p>

Question	Answer	Mark	Guidance
(ii)	 <p>✓</p> <p>✓</p>	2	<p><b>ALLOW</b> correct structural <b>OR</b> displayed <b>OR</b> skeletal formulae <b>OR</b> a combination of above as long as unambiguous</p> <p><b>ALLOW</b> a cyclic amide with a 3 membered ring</p> <p><b>ALLOW</b></p>  <p><b>OR</b> a structure obtained by condensation of a glutamic acid molecule with the first cyclic amide</p>

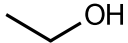
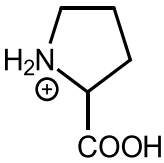
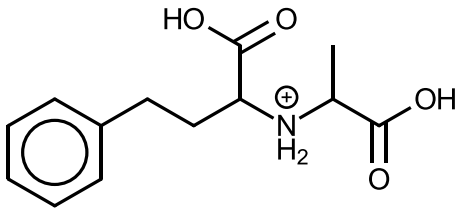


Question		Answer	Mark	Guidance
(d)	(i)	Ester <b>AND</b> amide ✓	1	<b>ALLOW</b> peptide for amide
	(ii)	 	2	<p><b>ALLOW</b> correct structural <b>OR</b> displayed <b>OR</b> skeletal formulae <b>OR</b> a combination of above as long as unambiguous</p> <p>Functional groups do not need to be fully displayed</p> <p><b>ALLOW</b> structures as shown; the O-H bond and the N-H bonds in the functional groups <b>do not</b> need to be displayed</p> <p><b>DO NOT ALLOW</b> -COOH</p> <p><b>ALLOW</b></p>  <p>Penalise incorrect connectivity to OH once in this question</p>
	(iii)	(The molecule/amide/ester) can be <u>hydrolysed</u> ✓	1	<p><b>ALLOW</b> (the molecule/amide/ester) can form hydrogen/H-bonds <u>with water</u></p> <p><b>IGNORE</b> acid/base</p>
<b>Total</b>			<b>20</b>	

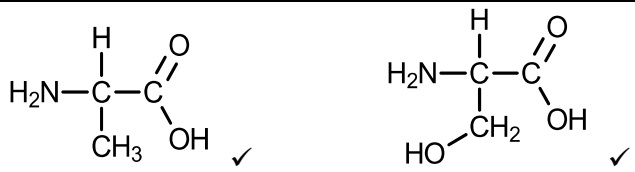
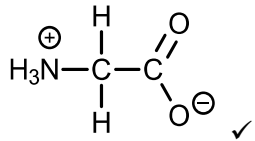
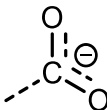
Question			Answer	Mark	Guidance
2	(a)	(i)	$\begin{array}{ccccccc} & \text{H} & \text{O} & & \text{CH}_2\text{OH} & & \\ &   &    & &   & & \\ \text{H}_2\text{N} & -\text{C} & -\text{C} & -\text{N} & -\text{C} & -\text{COOH} \\ &   & &   &   & & \\ & \text{CH}_3 & & \text{H} & \text{H} & & \end{array}$ $\begin{array}{ccccccc} & \text{H} & \text{O} & & \text{CH}_3 & & \\ &   &    & &   & & \\ \text{H}_2\text{N} & -\text{C} & -\text{C} & -\text{N} & -\text{C} & -\text{COOH} \\ &   & &   &   & & \\ & \text{HOH}_2\text{C} & & \text{H} & \text{H} & & \end{array}$ <div style="text-align: right;">✓</div>	2	<p><b>ALLOW</b> correct structural <b>OR</b> displayed <b>OR</b> skeletal formulae</p> <p><b>OR</b> combination of above as long as unambiguous</p> <p><b>DO NOT ALLOW</b> peptide chains</p>
	(a)	(ii)	<p>alanine at pH 6.0</p> $\begin{array}{ccc} & \text{H} & \text{O} \\ &   &    \\ \text{H}_3\text{N}^{\oplus} & -\text{C} & -\text{C} & -\text{O}^{\ominus} \\ &   & & \\ & \text{CH}_3 & & \end{array}$ <div style="text-align: right;">✓</div> <p>serine at pH 10.0</p> $\begin{array}{ccc} & \text{H} & \text{O} \\ &   &    \\ \text{H}_2\text{N} & -\text{C} & -\text{C} & -\text{O}^{\ominus} \\ &   & & \\ & \text{CH}_2\text{OH} & & \end{array}$ <div style="text-align: right;">✓</div>	2	<p><b>ALLOW</b> correct structural <b>OR</b> displayed <b>OR</b> skeletal formulae</p> <p><b>OR</b> combination of above as long as unambiguous</p> <p><b>ALLOW</b> + charge on N or H: <i>i.e.</i> <math>\text{NH}_3^+</math> or <math>\text{NH}_3^{\oplus}</math></p> <p><b>DO NOT ALLOW</b> ‘-’ charge on C <i>i.e.</i> <math>\text{COO}^-</math></p> <p><b>DO NOT ALLOW</b> if structure is incomplete</p>

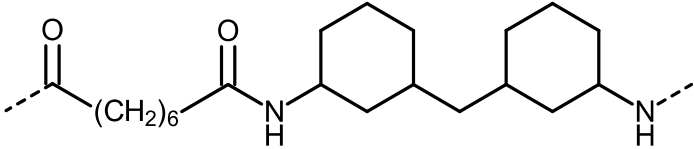
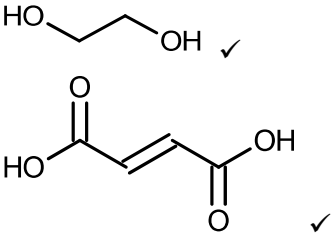
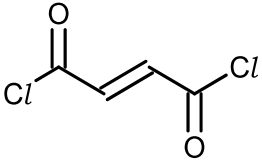
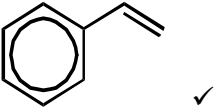
Question	Answer	Mark	Guidance
(a) (iii)	 <p>OR</p>	1	<p><b>ALLOW</b> correct structural <b>OR</b> displayed <b>OR</b> skeletal formulae  <b>OR</b> combination of above as long as unambiguous</p> <p><b>IGNORE</b> bond angles</p> <p><b>DO NOT ALLOW</b> more than one repeat unit</p> <p><b>ALLOW</b> end bonds shown as - - - - -</p> <p><b>DO NOT ALLOW</b> if structure has no end bonds</p> <p><b>IGNORE</b> brackets unless they are used to pick out the repeat unit from a polymer chain</p> <p><b>IGNORE</b> <i>n</i></p> 

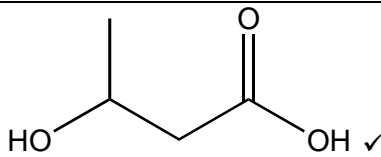
Question		Answer	Mark	Guidance									
	(b)	<p style="text-align: center;"><sup>1</sup>H NMR spectrum for serine</p> <table border="1" style="margin-left: auto; margin-right: auto;"> <thead> <tr> <th>chemical shift, <math>\delta</math> /ppm</th> <th>relative peak area</th> <th>splitting pattern</th> </tr> </thead> <tbody> <tr> <td>2.0 to 3.0</td> <td>1</td> <td>triplet</td> </tr> <tr> <td>3.3 to 4.2</td> <td>2</td> <td>doublet</td> </tr> </tbody> </table> <p>One mark for each correct row ✓✓</p>	chemical shift, $\delta$ /ppm	relative peak area	splitting pattern	2.0 to 3.0	1	triplet	3.3 to 4.2	2	doublet	2	<p><b>ALLOW</b> <math>\delta</math> values <math>\pm 0.2</math> ppm, as a range or a value within the range</p> <p><b>ALLOW</b> a response that implies a splitting into three for a triplet/into two for a doublet</p>
chemical shift, $\delta$ /ppm	relative peak area	splitting pattern											
2.0 to 3.0	1	triplet											
3.3 to 4.2	2	doublet											
	(c) (i)		1	<b>ALL</b> correct for one mark									
	(c) (ii)	<p>any <b>two</b> from:</p> <p>no/fewer side effects</p> <p>increases the (pharmacological) activity/effectiveness</p> <p>Reduces/stops the need for/cost/difficulty in separating stereoisomers/optical isomers</p> <p style="text-align: right;">✓✓</p>	2	<p><b>IGNORE</b> toxic/harmful</p> <p><b>IGNORE</b> a response that implies a reduced dose</p> <p><b>IGNORE</b> "it takes (less) time to separate"</p>									



Question		Answer	Mark	Guidance
(c)	(iii)	 ✓ one mark for ethanol  ✓ one mark for proline with NH <b>OR</b> NH <sub>2</sub> <sup>+</sup>  ✓ one mark for remaining fragment with  or  ✓ <b>Fourth</b> mark for structure of <b>both</b> ions shown correctly with NH <sub>2</sub> <sup>+</sup>	4	<b>ALLOW</b> correct structural <b>OR</b> displayed <b>OR</b> skeletal formulae <b>OR</b> combination of above as long as unambiguous <b>ALLOW</b> + charge on H of NH <sub>2</sub> groups, <i>i.e.</i> NH <sub>2</sub> <sup>+</sup> <b>IGNORE</b> negative (counter) ions
(c)	(iv)	idea of separating (the components/compounds) <b>AND</b> idea of (identifying compounds by) comparison with a (spectral) database ✓	1	<b>ALLOW</b> (identifies compounds) using fragmentation (patterns)/fragment ions (but <b>IGNORE</b> molecular ions) <b>IGNORE</b> retention times
<b>Total</b>			<b>15</b>	

Question			Answer	Marks	Guidance
3	(a)	(i)	monomers join/bond/add/react/form polymer/form chain AND another product/small molecule e.g. H <sub>2</sub> O/HCl ✓	1	IGNORE 'two' when referring to monomers, i.e. (two) monomers...
		(ii)		2	ALLOW correct structural OR displayed OR skeletal formulae OR combination of above as long as unambiguous ALLOW zwitterions
		(iii)	The pH at which the zwitterion exists ✓ 	2	ALLOW correct structural OR displayed OR skeletal formulae OR combination of above as long as unambiguous ALLOW pH at which there is no overall/net charge IGNORE pH at which there is no charge/ neutral charge <i>ie overall/net is required</i> ALLOW pH at which contains COO <sup>-</sup> AND NH <sub>3</sub> <sup>+</sup>  ALLOW delocalized carboxylate  ALLOW + on N or H; - must be on O
	(b)	(i)	Adsorption ✓	1	DO NOT ALLOW absorption ALLOW partition ALLOW adsorbtion
		(ii)	R <sub>f</sub> = 0.53 to 0.62 ✓ Amino acid is <u>methionine</u> ✓	2	Values vary if distance measured to middle or top of spot Independent marks. No need to show working as question asks for estimate of R <sub>f</sub>

Question	Answer	Marks	Guidance
(c)	 <p>amide link ✓ correct structure ✓</p>	2	<p><b>ALLOW</b> correct structural <b>OR</b> displayed <b>OR</b> skeletal formulae  <b>OR</b> combination of above as long as unambiguous  <b>ALLOW</b> 'terminal' —NH— at other end  'End bonds' <b>MUST</b> be shown (solid or dotted)  <b>IGNORE</b> brackets and/or <i>n</i>  <b>DO NOT ALLOW</b> aromatic rings in amine residue  <b>ALLOW</b> CONH for amide link</p>
(d) (i)	 <p>Penalise connectivity once (i.e. not —HO)</p>	2	<p><b>ALLOW</b> correct structural <b>OR</b> displayed <b>OR</b> skeletal formulae  <b>OR</b> combination of above as long as unambiguous  For dicarboxylic acid:</p>  <p><b>ALLOW</b> diethyl chloride</p> <p><b>DO NOT ALLOW</b> the CIS monomer</p>
(ii)		1	<p><b>ALLOW</b> correct structural <b>OR</b> displayed <b>OR</b> skeletal formulae  <b>OR</b> combination of above as long as unambiguous</p>
<b>Total</b>		<b>13</b>	

Question			er	Marks	Guidance
4	(a)	(	photodegradable <b>OR</b> light/sunlight/UV ✓	1	<b>IGNORE</b> IR/heat <b>IGNORE</b> bacteria  <b>DO NOT ALLOW</b> burn/combustion
		(ii)		1	<b>DO NOT ALLOW</b> structure with any C shown (especially as part of C=O)  <b>DO NOT ALLOW</b> OH—
	(b)	(	ammonia/NH <sub>3</sub> <b>AND</b> ethanol <b>OR</b> ethanolic ammonia ✓	1	<b>ALLOW</b> ammonia in a sealed tube <b>IGNORE</b> heat  <b>ALLOW</b> dilute ethanolic ammonia /NH <sub>3</sub>  <b>DO NOT ALLOW</b> any reference to water or hydroxide ions, e.g. <b>DO NOT ALLOW</b> dilute ethanolic NH <sub>3</sub> (aq) e.g. <b>DO NOT ALLOW</b> ethanolic NH <sub>3</sub> + NaOH
		(ii)	<b>Nitrogen</b> electron pair/lone pair accepts a proton/H <sup>+</sup> ✓ <i>Requires position of electron pair on N</i>  Cl <sup>-</sup> H <sub>3</sub> N <sup>+</sup> (CH <sub>2</sub> ) <sub>4</sub> N <sup>+</sup> H <sub>3</sub> Cl <sup>-</sup>  <b>OR</b> ClH <sub>3</sub> N(CH <sub>2</sub> ) <sub>4</sub> NH <sub>3</sub> Cl ✓	2	<b>DO NOT ALLOW</b> Nitrogen/N lone pair accepts hydrogen proton/H <sup>+</sup> required  <b>ALLOW nitrogen</b> donates an electron pair <b>IGNORE</b> NH <sub>2</sub> group donates electron pair  <b>ALLOW</b> + charge (if shown) on N or H of NH <sub>3</sub> e.g. Cl <sup>-</sup> H <sub>3</sub> N <sup>+</sup> (CH <sub>2</sub> ) <sub>4</sub> NH <sub>3</sub> <sup>+</sup> Cl <sup>-</sup>  <b>DO NOT ALLOW</b> just H <sub>3</sub> N <sup>+</sup> (CH <sub>2</sub> ) <sub>4</sub> NH <sub>3</sub> <sup>+</sup> i.e. <b>2 x Cl<sup>-</sup> MUST</b> be included

Question	er	Marks	Guidance
	<p>(iii) 1 mark for amide/peptide link correctly displayed within an attempted repeat unit ✓</p> <p>1 mark for rest of structure correct including side links ✓</p> $\text{---C(=O)---(CH}_2\text{)}_4\text{---C(=O)---N(H)---(CH}_2\text{)}_4\text{---N(H)---}$	2	<p>Minimum requirement is each end of a displayed amide group attached to a carbon atom (could be skeletal)</p> <p>Brackets <b>not</b> required</p> <p><b>IF</b> more than one repeat unit has been drawn a single repeat unit <b>MUST</b> be identified by brackets or clear label</p> <p><b>DO NOT ALLOW 2nd</b> mark if amide/peptide link wrong  <i>1st mark requires amide group fully displayed</i>  <i>For 2nd mark, <b>ALLOW</b> -CONH- in correct structure</i></p> <p><b>ALLOW</b> correct structural <b>OR</b> displayed <b>OR</b> skeletal formula</p> <p><b>ALLOW</b> combination of formulae as long as unambiguous e.</p>

Question		er	Marks	Guidance
(c)	(i)	<p>One mark for each correct structure</p> $  \begin{array}{c}  \text{O} \\  \parallel \\  \text{H}_3\text{N}^+ - \text{CH} - \text{C} - \text{O}^- \\    \\  \text{CHOH} \\    \\  \text{CH}_3  \end{array}  \quad \checkmark  $ $  \begin{array}{c}  \text{O} \\  \parallel \\  \text{H}_3\text{N}^+ - \text{CH} - \text{C} - \text{O}^- \\    \\  (\text{CH}_2)_4 \\    \\  \text{NH}_2  \end{array}  \quad \text{OR} \quad  \begin{array}{c}  \text{O} \\  \parallel \\  \text{H}_2\text{N} - \text{CH} - \text{C} - \text{O}^- \\    \\  (\text{CH}_2)_4 \\    \\  \text{NH}_3^+  \end{array}  \quad \checkmark  $	2	<p><b>ALLOW</b> correct structural <b>OR</b> displayed <b>OR</b> skeletal formula</p> <p><b>ALLOW</b> combination of formulae as long as unambiguous</p> <p><b>ALLOW</b> <math>\text{COO}^-</math></p> <p>'-' charge must be on O of <math>\text{COO}^-</math> but</p> <p><b>ALLOW</b> + sign shown as <math>^+\text{NH}_3</math> <b>OR</b> <math>\text{NH}_3^+</math></p> <p><b>BUT</b> only one <math>\text{NH}_2</math> can be protonated in zwitterion</p>
	(ii)	<p>Zwitterion at pH 9.60/higher pH has <b>one</b> <math>\text{NH}_2</math> group</p> <p><b>OR</b></p> <p>Zwitterion <b>OR</b> amino acid at pH 9.60/higher pH has a side chain with an <math>\text{NH}_2</math> group ✓</p> <p><b>Note:</b></p> <p><b>ASSUME</b> that 'it' refers to zwitterion</p>	1	<p><b>ALLOW</b> amino acid at 9.60/higher pH has <b>two</b> <math>\text{NH}_2</math> groups</p> <p><b>ALLOW</b> amino acid at 9.60/higher pH has more <math>\text{NH}_2</math> groups</p> <p><b>ALLOW</b> amine <b>OR</b> amino for <math>\text{NH}_2</math></p> <p><b>IGNORE</b> <math>\text{CHOH}</math> slightly acidic</p>
<b>Total</b>			<b>10</b>	