Q1.The anticancer drug cisplatin operates by reacting with the guanine in DNA.

Figure 1 shows a small part of a single strand of DNA. Some lone pairs are shown.

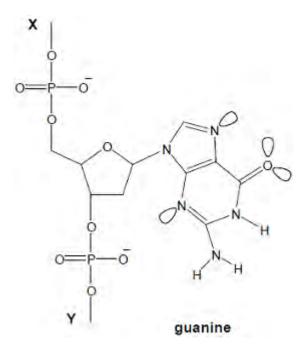


Figure 1

(a) The DNA chain continues with bonds at **X** and **Y**.

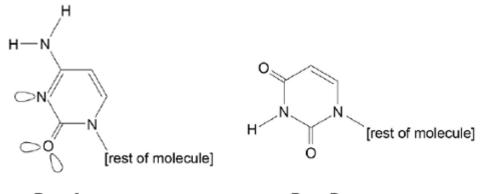
State the name of the sugar molecule that is attached to the bond at X.

(1)

(b) Messenger RNA is synthesised in cells in order to transfer information from DNA. The bases in one strand of DNA pair up with the bases used to synthesise RNA.

Figure 2 shows two bases used in RNA.

Figure 2



Base A

Base B

Suggest which of the bases **A** and **B** forms a pair with guanine in **Figure 1** when messenger RNA is synthesised.

Explain how the base that you have chosen forms a base pair with guanine.

••••••	 ••••••	

(4)

(c) Cisplatin works because one of the atoms on guanine can form a co-ordinate bond with platinum, replacing one of the ammonia or chloride ligands. Another atom on another guanine can also form a co-ordinate bond with the same platinum by replacing another ligand.

On Figure 1, draw a ring round an atom in guanine that is likely to bond to platinum.

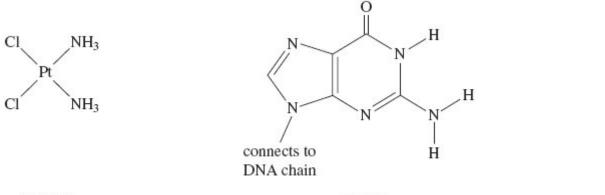
(1)

(d) An adverse effect of cisplatin is that it also prevents normal healthy cells from replicating.

Suggest **one** way in which cisplatin can be administered so that this side effect is minimised.

(1) (Total 7 marks)

Q2. The complex cisplatin acts as an anticancer drug by changing the properties of DNA when it reacts with guanine, a component of DNA.



cisplatin

guanine

When cisplatin is absorbed into the human body, it undergoes a ligand substitution reaction and one chloride ligand is replaced by a water molecule forming a complex ion **Q**.

(a) Write an equation for this substitution reaction to form the complex ion **Q**.

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(2)

(b) The complex ion **Q** can bond to guanine in two different ways.

The first way involves a hydrogen atom, from one of the ammonia ligands on Q, bonding to an atom in a guanine molecule. State the type of bond formed to guanine and identify an atom in guanine that could form a bond to this hydrogen atom.

Type of bond	
Atom in guanine	

(ii) The second way involves a ligand substitution reaction in which an atom in a guanine molecule bonds to platinum by displacing the water molecule from Q. State the type of bond formed between guanine and platinum when a water molecule is displaced and identify an atom in guanine that could bond to

	platinum in this way.	
	Type of bond	
	Atom in guanine	(2)
		(-)
()	State and explain one risk associated with the use of cisplatin as an anticancer drug.	
F	Risk	
E	Explanation	(2)
	(Total 8 ma	(2) rks)