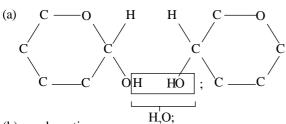
## **QUESTIONSHEET 1**



(b) condensation;

1

2

1

(c) hydrolysis/digestion;

(d) liver/muscles;

1

(e) mix equal volumes of solution and dilute hydrochloric acid;

boil in a water bath (for 2 minutes) to hydrolyse sucrose (to glucose and fructose);

mix with equal volumes of Benedict's reagent;

boil in water bath (for 2 minutes);

if brick red precipitate appears then sucrose is present;

mix equal volumes of solution and (dilute/5 %) sodium hydroxide;

run dilute/1% copper sulphate solution into the solution;

a purple ring at the interface indicates protein is present;

Max 6

TOTAL 11

# **QUESTIONSHEET 2**

(a) isomers;

1

(b) the position of -H and -OH groups on first carbon atom;

1

(c) leads to greater chemical variety/biochemical division of labour; alpha glucose/starch is respiratory; whereas beta glucose/cellulose is structural;

max 2

1

(d) glycosidic/condensation links;

#### ANSWERS & MARK SCHEMES

## **QUESTIONSHEET 3**

(a) (i) saturated means that the molecule contains the maximum number of hydrogen atoms; unsaturated means that the molecule contains fewer hydrogen atoms than it might/contains double bonds;

(ii) unsaturated fats have lower melting points than saturated fats; unsaturated fats form oils but saturated fats are solid:

(b) Any four of:

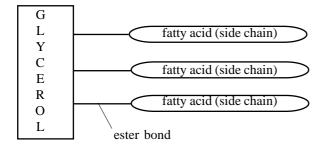
don't dissolve in water/body fluids/ therefore don't affect osmotic balance of cells/tissues/ have higher calorific value than carbohydrates/ can yield more energy per gramme on oxidation/ reference to other useful properties/buoyancy/insulation;;;;

4

1 mark for labelling the glycerol part of the molecule; 1 mark for labelling the fatty acids;

1 mark for labelling the bond;

(c) (i)



(ii) one fatty acid molecule would be replaced by phosphoric acid/phosphate;

TOTAL 12

2

2

3

1

2

2

## **QUESTIONSHEET 4**

(a) fatty acids joined to;

glycerol;

phosphate also attached;

reference to alcohol/choline attached to phosphate; (credit points on a diagram) max3

(b) (i)

(fatty acid) side chain polar head

Correct drawing; Correct labels;

(ii) polar heads mix with water, non polar tails do not; thus heads face water on both sides with tails to middle;

thus heads face water on both sides with ta

(c) Any 2 of:
waterproofing/protection qualified/cell membrane structure/insulation/give buoyancy;;

2

(d) Either emulsion test - take sample and add equal volume of ethanol and an equal volume of cold water; mix and if positive a white emulsion forms;

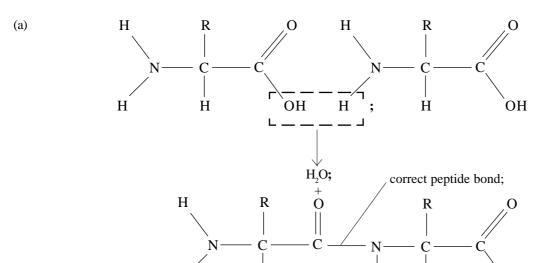
or Sudan III test - add a few drops of Sudan III to sample; red fat droplets appear if positive;

### ANSWERS & MARK SCHEMES

Н

Η

# **QUESTIONSHEET 5**



Η

(b) (i) rest of the molecule/side chain;

1

1

1

3

(ii) hydrogen/methyl group/any correct group;

Η

1

OH

(iii) peptide;

(c) their molecules contain both acidic and basic groups;

TOTAL 7

## **QUESTIONSHEET 6**

(a) secondary protein structure/beta-pleated sheet;

1

(b) many hydrogen bonds between polypeptide chains;

1

(c) polypeptide chain;

bends/folds extensively;

into a compact/globular structure;

3

(d) Any two of:

ionic/

hydrogen/

disulphide bridges;; (reject 'peptide' since these hold the primary structure together)

2

(e) haemoglobin/myoglobin/antibodies;

1

(f) add equal volume of (dilute/5%) potassium hydroxide solution to test solution; (down side of test tube) add a few drops of (1%) copper sulphate solution; presence of a purple ring at interface; which dissolves to form a purple solution on shaking, indicates protein;

max 3

#### ANSWERS & MARK SCHEMES

# **QUESTIONSHEET 7**

(a) A - phosphate;

B - pentose/ribose/deoxyribose; (not. sugar).

C - <u>nitrogenous</u> base/pyrimidine/purine;

D- nucleotide;

(b) (i) bases that will join together;

by hydrogen bonds; A to T and G to C;

max 2

4

(ii) RNA has uracil instead of thymine;

RNA has ribose sugar instead of deoxyribose;

RNA is single stranded instead of double stranded;

3

TOTAL 9

## **QUESTIONSHEET 8**

(a)

	monosaccharide	disaccharide	polysaccharide
ribose	✓	х	<b>x</b> ;
glucose	✓	x	<b>x</b> ;
maltose	X	✓	<i>x</i> ;
starch	х	x	<b>√</b> ;
lactose	x	✓	<i>x</i> ;
glycogen	х	х	<b>√</b> ;
cellulose	х	x	<b>√</b> ;

7

(b) starch contains  $\alpha$ -glucose, cellulose contains  $\beta$ -glucose; starch linked by  $\alpha$ - glycosidic links, cellulose by  $\beta$ -glycosidic links; starch may contain branched chains, cellulose is unbranched;

max 2

#### ANSWERS & MARK SCHEMES

# **QUESTIONSHEET 9**

- (a) (i) S = deoxyribose; P = phosphate; G = guanine; C = cytosine; T = thymine; A = adenine;
- 6

- (ii) G-C: hydrogen;
  - S-S: phosphate bridge/bond;

2

(b) (i) supply energy;

for all energy-requiring reactions/synthesis/anabolism;

(ii) co-enzyme;

hydrogen acceptor in respiration;

(iii) responsible for transfer of acetyl units; from glycolysis to Krebs cycle;

6

TOTAL 14

## **QUESTIONSHEET 10**

(a) disaccharide/maltose;

(c) condensation;

1

(b) 1,4-glycosidic/alpha link;

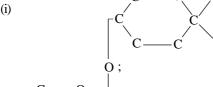
1

1

(d) respiration/energy substrate;

1

(e) (i)



1

- $C \longrightarrow C$
- (ii) (1, 4-)glycosidic beta link;

TOTAL 6

1

## **QUESTIONSHEET 11**

(a) protein;

1

(b) Any three of:

mix together equal volumes of solution and Benedicts reagent/

boil in a waterbath for (2 minutes)/

presence of brick red precipitate indicates glucose;;;

3

(c) use same volumes of test and standard solutions;

add same volume of glucose oxidase solution to each;

incubate for a standard/stated time at a suitable/stated temperature;

add same volume of peroxide and indicator;

read intensity of colour in photometer;

calculate answer using formula Concentration of test

 $\frac{\text{Concentration of test}}{\text{Concentration of standard}} = \frac{\text{Reading of Test}}{\text{Reading of Standard}}$ 

max 4

### ANSWERS & MARK SCHEMES

# **QUESTIONSHEET 12**

(a) A - (beta)-glucose;

B - (1,4) glycosidic beta link;

(b) condensation/removal of water;

(c) main component plant cell walls;

	SUBSTANCE		
	Starch	Protein	DNA
Only contains C, H, and O	✓	x	x
Contains nitrogen	X	✓	1
Positive when boiled with Benedicts reagent	X	х	х
Hydrolysed to smaller units during digestion	✓	✓	✓

X

TOTAL 9

5

# **QUESTIONSHEET 13**

Contains uracil

(a) A & B;

X

X

(b) (hydrophobic) tails;
of phospholipid molecules;
2

(c) (i) glycocalyx;

(ii) polysaccharide/glycoprotein/carbohydrate; 1

(ii) any two of: aids cell recognition/cell adherence/receptor sites for hormones/antibodies;; 2

#### ANSWERS & MARK SCHEMES

## **QUESTIONSHEET 14**

(a) X - nucleic acid/DNA/RNA/ATP/nucleotide;

Y - protein/amino acid/polypeptide;

Z - cellulose/hemicellulose;

(b) amylose; amylopectin; amylose is an unbranched chain but amylopectin is branched; 2

1

(c) (i) forms covalent/sulphur bonds; holding adjacent polypeptides together/contributes to secondary/tertiary structure;

(ii) increases protein's stability to pH change; increases protein's stability to temperature change; (allow 1 mark if just refer to 'strong bonding')

TOTAL 10

3

2

2

3

2

3

# **QUESTIONSHEET 15**

(a) X - phospholipid;

 $Y\hbox{-}extrinsic/external/surface protein;}\\$ 

Z - intrinsic/integral/internal protein;

(b) Y - cell recognition/support glycocalyx/has receptor groups;

Z - facilitated diffusion/active transport;

(c)

,	Phospholipid	Protein	Carbohydrate		
Act as enzymes	Х	✓	X		
Allows passage of water soluble substances	x	✓	x		
Involved in cell recognition	х	✓	✓		

#### ANSWERS & MARK SCHEMES

## **QUESTIONSHEET 16**

(a) A: phosphate; B: glycerol;

2

(b) circle round



1

(c) Precursors for prostaglandins/hormones/other fatty acids/essential for growth/provide energy/ATP synthesis;

1

(d) (i) vitamin A:

night blindness/hyperkeratosis/xerophthalmia;

failure to synthesise retinol/rhodopsin/drying/hardening/fracturing of epithelial tissues;

2

(ii) vitamin D:

rickets in children/osteomalacia in adults;

failure to regulate calcium/phosphate metabolism/absorption/mobilisation from bone/bent long bones/bow legs/ broken bones in adults;

TOTAL 8

2

# **QUESTIONSHEET 17**

a) (i) α helix/secondary structure/fibrous structure; 1

Any three of: (ii)

hydrogen bonds/ionic bonds/sulphur bonds/peptide bonds;;;

max3

(b) (i) tertiary structure/globular structure;

1

(ii) quaternary structure;

1

(c) consists of 4 polypeptide chains/ $2\alpha$  chains and  $2\beta$  chains;

tightly folded globular structure/compact;

to fit in red blood cells;

each chain contains a haem group which contains iron;

each haem group can reversibly bind with an oxygen molecule;

max 4

TOTAL 10

## **QUESTIONSHEET 18**

(a) alpha-;

glycosidic;

(b) liver/muscles;

2

1

(c) many ends/exposed/terminal glucose units;

allows rapid release of glucose/rapid digestion of molecule/rapid mobilisation in respiration; make molecule compact so much can be stored in a small space;

max 2

(d) glucoses linked by beta-glycosidic links;

molecules are unbranched;

2

### AS 2

## **BIOMOLECULES**

### ANSWERS & MARK SCHEMES

## **QUESTIONSHEET 19**

(a)				
(4)	Macromolecule	Composition		
	RNA/DNA/nucleotides;			
		amino acids;		
		glycerol;		
		alpha-glucose;		
		beta-glucose;		

5

(b) sequence of amino acids;

governs distribution of bonding regions;

which influence the position of hydrogen bonds/ionic bonds/disulphide bridges;

max 2

1

3

TOTAL 7

## **QUESTIONSHEET 20**

deoxyribose/pentose; nitrogenous; hydrogen; cytosine; adenine; complementary; purine; thymine/cytosine; double helix; ten; TOTAL 10

# **QUESTIONSHEET 21**

(a) (i) on the single chromosome strand/in a plasmid;

(ii) in the nucleus/on every chromosome; 1

(iii) a nucleotide;

(iv) A - adenine; T = thymine; C = cytosine; G = guanine; P = phosphate; D = deoxyribose;

(b) C pairs with G thus these must add up to 54%;

thus A and T must form remaining 46%;

in equimolecular proportions thus thymine = 23%;