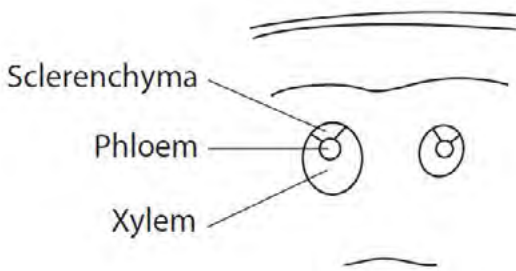
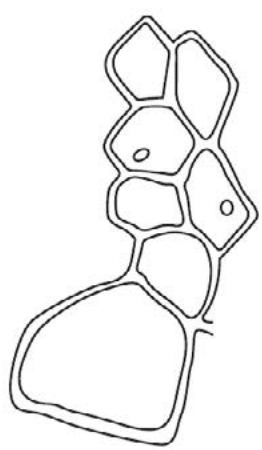


Unit 3 - Mark scheme

Question number	Answer	Mark
1(a)(i)	<p>A drawing showing the following features:</p> <ul style="list-style-type: none"> • neat lines and no cells drawn (1) • two vascular bundles drawn and three or four layers of tissue (1) • two tissues labelled (1) <p>Example of labelled drawing:</p> 	(3)

Question number	Answer	Mark
1(a)(ii)	<p>A drawing showing the following features:</p> <ul style="list-style-type: none"> • cells in the box drawn to the correct proportions (1) • nuclei drawn in the two cells (1) • cell walls shown correctly (1) <p>Example of drawing:</p> 	(3)

Question number	Answer	Mark
1(b)(i)	<p>A calculation showing the following steps:</p> <ul style="list-style-type: none"> • equivalent micrometer units and graticule units given (1) • 1 graticule unit calculated (1) <p>Example of calculation:</p> <p>6 stage micrometer units = 20 graticule units</p> <p>so 1 graticule unit = $60 \div 20 \mu\text{m} = 3 \mu\text{m}$</p>	(2)

Question number	Answer	Mark
1(b)(ii)	<p>An answer showing the following steps:</p> <ul style="list-style-type: none"> • reading diameter of A using scale (1) • actual diameter calculated (1) <p>Example of calculation:</p> <p>A is 24 units wide</p> <p>so actual diameter is $24 \times 3 = 72 \mu\text{m}$</p>	(2)

Question number	Answer	Mark
2(a)(i)	<ul style="list-style-type: none"> • concentration of sodium hydroxide solution 	(1)

Question number	Answer	Mark
2(a)(ii)	<p>An answer that includes any one of the following pairs:</p> <ul style="list-style-type: none"> • temperature of solution (1) • carry out in a thermostatically controlled water bath (1) <p>or</p> <ul style="list-style-type: none"> • length of time in solution (1) • start them all at same time / stopwatch (1) 	(2)

Question number	Answer	Mark
2(a)(iii)	<p>An answer that includes any five of the following points:</p> <ul style="list-style-type: none"> • a source material variable taken into account, e.g. length, width, age, mass, hydration level, part of plant extracted from (1) • environmental variable controlled, e.g. temperature, humidity (1) • named procedural variable controlled, e.g. size of masses used (1) • idea of adding masses until fibre breaks / measure the mass {that breaks the fibre / that the fibre can hold before breaking} (1) • repeat and find the {mean / average} (1) • reference to safety procedure (1) 	(5)

Question number	Answer	Mark
2(b)(i)	<p>An answer that includes the following points:</p> <ul style="list-style-type: none"> • for {Baobab, Okra and Kenaf / three of the plants} concentration should be 10% (1) • for Roselle it should be 25% (1) 	(2)

Question number	Answer	Mark																	
2(b)(ii)	<p>A table, drawn showing:</p> <ul style="list-style-type: none"> • suitable table drawn (1) • headings of sodium hydroxide concentration with units, tensile strength with units and the two species (1) • data correctly entered into it (1) <p>Example of table drawn:</p> <table border="1"> <thead> <tr> <th rowspan="2">Sodium hydroxide concentration/%</th> <th colspan="2">Tensile strength/a.u.</th> </tr> <tr> <th>Roselle</th> <th>Baobab</th> </tr> </thead> <tbody> <tr> <td>10</td> <td>0.40</td> <td>1.00</td> </tr> <tr> <td>15</td> <td>0.30</td> <td>0.63</td> </tr> <tr> <td>20</td> <td>0.59</td> <td>0.65</td> </tr> <tr> <td>25</td> <td>0.65</td> <td>0.60</td> </tr> </tbody> </table>	Sodium hydroxide concentration/%	Tensile strength/a.u.		Roselle	Baobab	10	0.40	1.00	15	0.30	0.63	20	0.59	0.65	25	0.65	0.60	(3)
Sodium hydroxide concentration/%	Tensile strength/a.u.																		
	Roselle	Baobab																	
10	0.40	1.00																	
15	0.30	0.63																	
20	0.59	0.65																	
25	0.65	0.60																	

Question number	Answer	Mark
2(c)(i)	An answer that includes the following comparative points: <ul style="list-style-type: none"> • both {peak at 8% (sodium hydroxide) / go down above 8% (sodium hydroxide) / rise at 25% (sodium hydroxide)} (1) • 0°C peaks at 30% (sodium hydroxide) but at 20°C {reaches a plateau / flattens off at 30% (sodium hydroxide)} (1) 	(2)

Question number	Answer	Additional guidance	Mark
2(c)(ii)	A description that includes the following points: <ul style="list-style-type: none"> • an increase in temperature leads to a reduction in degree of swelling (1) • this effect is not linear (1) 	Accept reverse argument	(2)

Question number	Answer	Mark
3(a)(i)	An answer showing the following steps: <ul style="list-style-type: none"> • change in mass calculated (1) • percentage change calculated (1) <p>Example of calculation:</p> $2.75 - 1.92 = -0.83 \text{ g}$ $(-0.83 \div 2.75) \times 100 = -30.18\%$	(2)

Question number	Answer	Mark												
3(a)(ii)	<p>A graph showing the following features:</p> <ul style="list-style-type: none"> • A axes correct (x - concentration of sucrose solution, y - percentage change in mass) (1) • L axes correctly labelled, and with units mol dm^{-3} and % (1) • P correct plotting (1) • S points joined with straight lines (1) <p>Example of graph:</p> <table border="1"> <caption>Data points from the example graph</caption> <thead> <tr> <th>Concentration of sucrose solution / mol dm^{-3}</th> <th>Percentage change in mas (%)</th> </tr> </thead> <tbody> <tr> <td>0</td> <td>30</td> </tr> <tr> <td>0.2</td> <td>8</td> </tr> <tr> <td>0.4</td> <td>-2</td> </tr> <tr> <td>0.6</td> <td>-15</td> </tr> <tr> <td>0.8</td> <td>-25</td> </tr> </tbody> </table>	Concentration of sucrose solution / mol dm^{-3}	Percentage change in mas (%)	0	30	0.2	8	0.4	-2	0.6	-15	0.8	-25	(4)
Concentration of sucrose solution / mol dm^{-3}	Percentage change in mas (%)													
0	30													
0.2	8													
0.4	-2													
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Question number	Answer	Additional guidance	Mark
3(a)(iii)	<p>An answer showing the following steps:</p> <ul style="list-style-type: none"> • 0.35 (1) • mol dm^{-3} (1) 	Allow ecf from graph	(2)

Question number	Answer	Mark
3(a)(iv)	<p>An explanation that includes the following points:</p> <ul style="list-style-type: none"> • there is no (net) change in the quantity of water in the carrot tissue / the water potential in the carrot equals the water potential of the solution (1) • {because / therefore} the rate of water gain is equal to its loss (1) 	(2)

Question number	Answer	Mark
3(b)(i)	<p>Any two from the following:</p> <ul style="list-style-type: none"> • surface area / volume / age / variety / storage conditions / source (of beetroot) / same {wavelength / filter} (1) 	(2)

Question number	Answer	Mark
3(b)(ii)	<ul style="list-style-type: none"> • intensity of the red colour 	(1)

Question number	Answer	Mark
3(b)(iii)	<p>A graph showing the following features:</p> <ul style="list-style-type: none"> axes correctly labelled with units (x temperature/°C and y intensity of the red colour/a.u.) (1) scales correctly labelled (1) standard deviations (SDs) all correctly plotted (1) <p>Example graph:</p>	(3)

Question number	Answer	Additional guidance	Mark
3(b)(iv)	<p>An answer that includes the following points:</p> <ul style="list-style-type: none"> {cells / membranes / eq} damaged (by cutting up of pieces) (1) so pigment could leak out of {vacuoles / cells} (1) 	Accept reference to condensation on cuvette at low temperature, leads to absorption of some of the light	(2)

Question number	Answer	Mark
3(b)(v)	<p>An answer that includes any five of the following points:</p> <ul style="list-style-type: none">• (overall) the intensity of the red colour increases as temperature increases (1)• but from 0°C to 40°C, the SDs overlap so no significant effect (1)• at 50°C the mean is higher than that at 40°C but SDs overlap so there is no case for saying the difference is significant (1)• at 60°C the mean is higher than that at 50°C and the SDs do not overlap so this difference can be regarded as significant (1)• at 70°C the mean degree of redness falls from that at 60°C but the SDs overlap so temperatures above 60°C appear to have no further effect (1)• there are no data above 70°C so cannot say what any further rise in temperature would cause (1)	(5)