

AQA Biology A-Level Required Practical 7

Use of chromatography to investigate the pigments isolated from leaves of different plants, e.g. leaves from shade-tolerant and shade-intolerant plants or leaves of different colours.

▶ Image: PMTEducation

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Chromatography is used to **separate** out different components in a sample. In this experiment, the **photosynthetic pigments** of a plant sample are separated into **bands of colour** by paper chromatography. The rates of migration of individual pigments will depend on their **solubility**, **mass** and **affinity to the paper**.

Equipment list

- Filter paper
- Leaf sample
- Distilled water
- Pestle and mortar
- Pencil
- Ruler
- Capillary tube
- Chromatography solvent
- Acetone

Method

- 1. Draw a straight line in pencil approximately **1cm** above the bottom of the filter paper being used. Do not use a pen as the ink will obscure the results.
- 2. Cut a section of leaf and place it in a **mortar**. Add 20 drops of **acetone** and use the **pestle** to grind up the leaf sample and release the **pigments**.
- 3. Use a **capillary tube** to extract some of the **pigment** and blot it onto the centre of the pencil line you have drawn.
- Suspend the paper in the solvent so that the level of the liquid does not lie above the pencil line and leave the paper until the solvent has run up the paper to near the top.
- 5. Remove the paper from the solvent and draw a pencil line marking where the solvent moved up to. The pigment should have separated out and there should be different spots on the paper at different heights above the pencil line.
- Calculate the Rf value for each spot (distance travelled by solute/distance travelled by solvent). Always measure to the centre of each spot.



Risk Assessment

Hazard	Risk	Safety Precaution	In emergency	Risk Level
Biohazard	Allergies; soil bacteria; contamination	Wash hands after use	Seek assistance	Low
Chromatography solvent	Flammable; causes irritation to eyes and skin; harmful by inhalation	Avoid contact with solvent; wear eye protection; keep solvent in fume cupboard; make sure room is well ventilated; keep away from naked flame	Wash from skin and eyes using cold water; put out fires; seek medical assistance	Low
Acetone	Flammable; causes irritation to eyes; inhalation may lead to dizziness and drowsiness	Avoid contact; wear eye protection; make sure room is well ventilated; keep away from naked flame	Wash from skin and eyes using cold water; put out fires; seek medical assistance	Low

Conclusion

 Rf values should be compared to the Rf known values in a database to identify pigment. When looking at the databases, ensure that they are for paper chromatography and use the same solvent as these variables will make results differ.

Factors that affect the rate of mobility

- 1. Affinity- pigments have different affinities to the chromatography paper; those with lower affinities will travel further up the paper.
- 2. Solubility- pigments that are more soluble travel faster up the paper and will end up closer to the top at the solvent front.

▶ Image: PMTEducation

Pigments that travel further up the paper will have a higher Rf value.