

NSAA 2019 Section 1

Model Solutions

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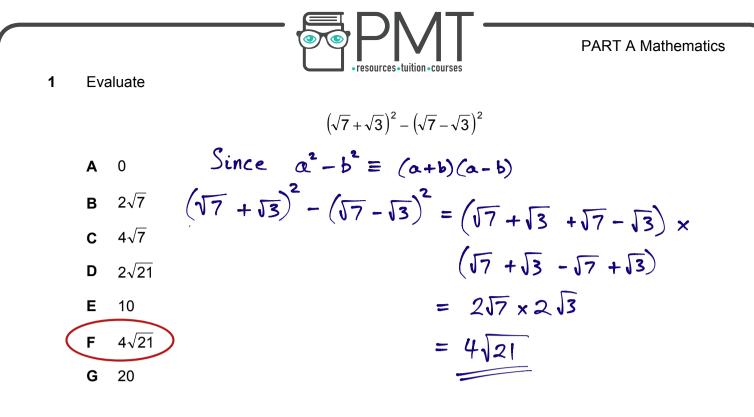


PART A Mathematics

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2 Find the complete set of values of *x* which satisfy the inequality

	$\frac{1}{2}(3x-2) - \frac{2}{3}(x-4) < x$
A <i>x</i> < –22	$\frac{3}{2}x - 1 - \frac{2}{3}x + \frac{8}{3} < x$
B $x > -22$.	2 3 3
C <i>x</i> < -2.5	$\frac{5}{6}x + \frac{5}{3} < x$
D x > -2.5	6 3
E <i>x</i> < 1.2	$\frac{x}{6} > \frac{5}{3}$
F <i>x</i> > 1.2	\sim $>$ 10
G <i>x</i> < 10	x > 10
H x > 10	

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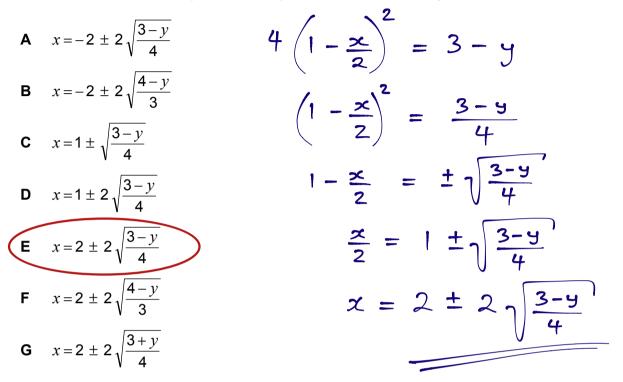


PART A Mathematics

3 The equation gives *y* in terms of *x*:

 $y = 3 - 4\left(1 - \frac{x}{2}\right)^2$

Which one of the following is a rearrangement for x in terms of y?



4 The resistance to the motion of a car is directly proportional to the square of the speed of the car. $R \propto v^2 \implies R = kv^2$

The car increases its speed by 20%.

What is the percentage increase in the resistance to the motion of the car?

Α	20%	When $V \mapsto 1.2v$
в	24%	$p \rightarrow 1 (10)^2$
С	44%	$R \mapsto k(1.2v)^2$
D	120%	$= 1.44 \mathrm{ku}^2$
Е	224%	$= \underbrace{1.44}_{R}$
F	240%	44% increase
G	400%	in R

PART A Mathematics resources tuition courses The equation of a curve is $y = px^2 + qx$ where *p* and *q* are constants. 5 The curve passes through the points (2,6) and (4,-4). What is the value of q - p? At (2,6): At (4,-4): Α 1 6 = 4p + 2q -4 = 16p + 4qВ 2 С 5 \Rightarrow 4p + q = -1 $\Rightarrow 2p + q = 3 - (1)$ 2 6 D 9 Е (2)-(1): 2p=-4 $\begin{cases} \Rightarrow q-p = 7+2 \\ = q \end{cases}$ F 16 =) p=-2, q=3-2p= 3-2(-2)=7Which of the following is a simplification of 6 $4 - \frac{x(3x+1)}{x^2(3x^2-2x-1)}$ In the denominator. $\mathbf{A} \quad \frac{12x^3 - 8x^2 - 7x - 1}{x(3x - 1)(x - 1)}$ $3x^2 - 2x - 1 \equiv (3x + 1)(x - 1)$ **B** $\frac{4x^2+4x-1}{x(x+1)}$ Considering the whole expression: **c** $\frac{4x^2+4x+1}{x(x+1)}$ $4 - \frac{x(3x+1)}{x^2(3x+1)(x-1)}$ **D** $\frac{4x^2-4x-1}{x(x-1)}$ $= 4 - \frac{1}{r/r-1}$ $E = \frac{4x^2 - 4x + 1}{4x^2 - 4x + 1}$ x(x-1) $\frac{12x^3 - 8x^2 - x + 1}{x(3x-1)(x-1)}$ $= \frac{4x(x-1) - 1}{x(x-1)}$ F $= \frac{4x^2 - 4x - 1}{x (x - 1)}$

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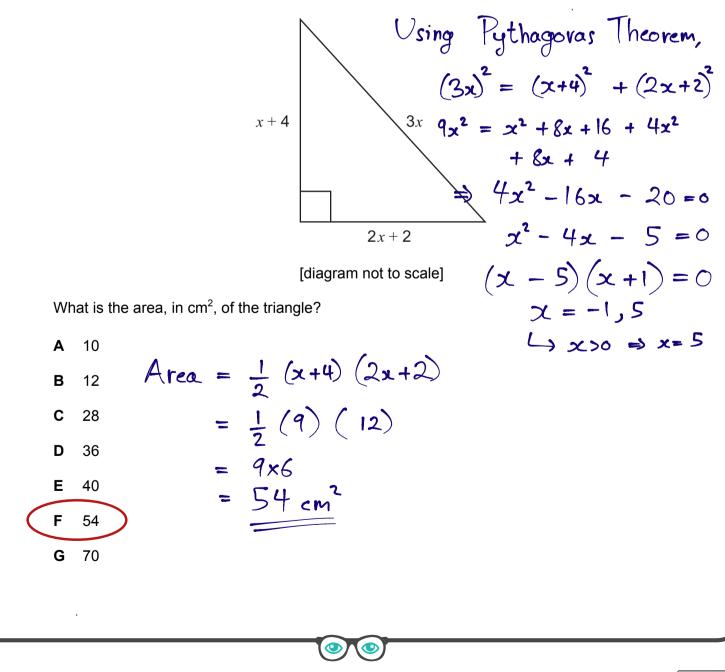
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PART A Mathematics

7 The ball for a garden game is a solid sphere of volume 192 cm³.

For the children's version of the game the ball is a solid sphere made of the same material, but the radius is reduced by 25%.

8 The diagram shows a right-angled triangle, with sides of length x + 4, 2x + 2 and 3x, all in cm.



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PART A Mathematics





$$9^{2x-1} \times \frac{1}{27^x} = 81^x$$

Convert all bases to 3

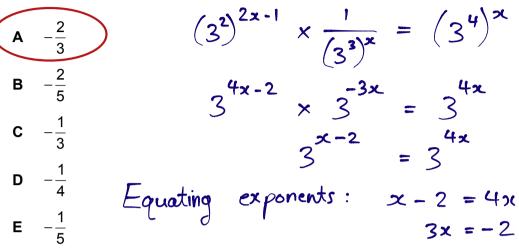
 $(3^2)^{2x-1} \times \frac{1}{(3^3)^x} = (3^4)^x$

 $3^{4x-2} \times 3^{-3x} = 3^{4x}$

 $3^{x-2} = 3^{4x}$

 $x = \frac{-2}{3}$

what is the value of x?



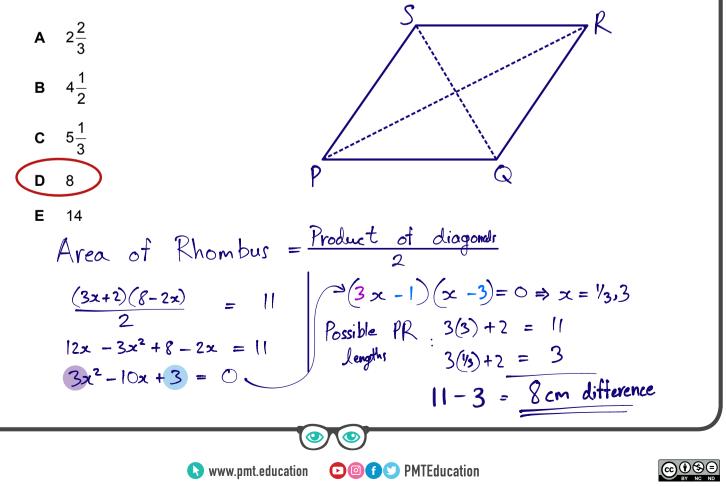
10 *PR* and *QS* are the diagonals of a rhombus *PQRS*.

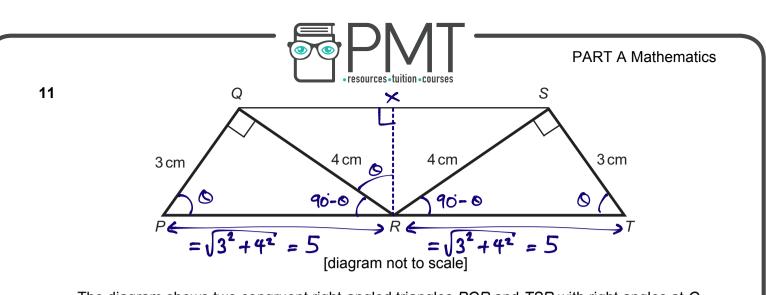
$$PR = (3x + 2) \,\mathrm{cm}$$

$$QS = (8 - 2x) \, \mathrm{cm}$$

The area of PQRS is 11 cm^2 .

What is the difference, in cm, between the two possible lengths of PR?





The diagram shows two congruent right-angled triangles *PQR* and *TSR* with right angles at *Q* and *S*, respectively.

PQ = TS = 3 cm

QR = SR = 4 cm

PRT is a straight line.

What is the length, in cm, of QS?

Α	4	$\angle QRS = 180' - (90' - 0) - (90' - 0)$
В	3√2	= 20
С	5.2	From DPQR,
D	$4\sqrt{2}$	$\sin 0 - 4$
E	6.4	$\sin \circ = \frac{4}{5}$
F	8.2	Let X be the midpoint of QS
G	10	$QX = QR \sin Q$
		$= 4 \times \frac{4}{5}$
		$= \frac{16}{5} = \frac{32}{10} = 3.2$ QS = 2QX
		QS = 2QX ¹⁰
		= 6.4

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		PART A Mathematics
12	The total of	three numbers p , q and r is 375 $p + q + r = 375$
	The ratio p :	$q is 5:7 \equiv 5 \times 4 : 7 \times 4 = 20:28$
	The ratio q :	$r is 4:11 \equiv 4 \times 7 : 11 \times 7 = 28 : 77$
	What is the	value of $p + r$? $\therefore p : q : r = 20 : 28 : 77$
	A 16	So if x is some number,
	B 60	p + q + r = 20x + 28x + 77x = 375
	C 97	Since the ratios tell us p,q and r will always be muliples of this ratio.
	D 125	$=) 125 \times = 375$
	E 144	x = 3
	F 231	$p = 20 \times 3 = 60$
(G 291	$\implies q = 28 \times 3 = 84 \therefore p + r = 60 + 231 \\ = 291$
	H 315	$r = 77 \times 3 = 231$

13 The straight line *P* has equation 3y - 2x = 12 and intercepts the *y*-axis at the point (0, *p*).

The straight line Q is parallel to P, passes through the point (6, -1) and intercepts the *y*-axis at the point (0, q).

What is the value of p - q?

Α -9 $P: \quad y = \frac{2}{3}x + 4$ $L_{y-intercept} = p = 4$ -7 В С 1 D 9 $Q: \quad q = \frac{2}{3}x + q$ Е 14 Sub in (6,-1) F 17 $-1 = \frac{2}{3} \times 6 + q \implies q = -5$ p-q = 4+5 = 9

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PART A Mathematics

14 The vertices of a rectangle have coordinates:

P(4,5) Q(4,8) R(10,8) S(10,5)

PQRS is transformed by a clockwise rotation of 90° about *P* followed by a reflection in the x-axis.

What are the coordinates of the final position of R?

A (-8,-10) Move the rectangule so P is on the origin
B (-7,-1)
C (-4,1) Pnew = (0,0)
D (-1,11) Rnew =
$$\binom{10}{8} - \binom{44}{5} = (6,3)$$

E (1,-11)
F (4,-1) Rotate Rnew 90° (2
G (7,1)
H (8,10) $\binom{3}{1+3} \times \ldots \times \binom{7}{1+3} = (3,-6)$
 $\binom{3}{1+3} \times \ldots \times \binom{7}{1+3} = (3,-6)$
Translate R'new by moving Rectangle so P is not origin,
 $\binom{3}{-6} + \binom{4}{5} = (7,1)$
ALTERNATIVE METHOD if you are familiar with Linear Transforms
R'fout = so P is origin Rect THEN Rotate THEN Like P was never at origin
= $\binom{4}{5} + \binom{9}{1-1} \times \binom{7}{1-1} = \binom{7}{1}$

PART A Mathematics

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15 Box A contains exactly 10 balls, of which 6 are red and 4 are blue.

Box B contains exactly 15 balls, of which 3 are red and 12 are blue.

All the balls are identical in every respect, apart from colour.

One of the two boxes is chosen at random by tossing two fair coins, as follows:

"If **both** coins show heads, box A is selected. Otherwise box B is selected."

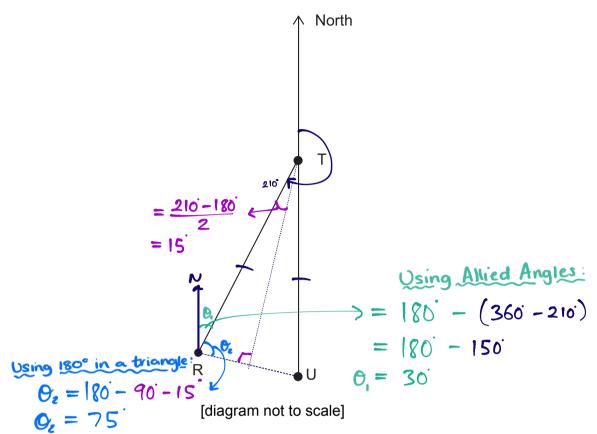
One ball is then randomly taken from the selected box.

What is the probability that a red ball is taken?

P(red) = P(red | A OR red | B) Α 400 3 = P(red | A) + P(red | B) В 25 3 = P(red AND A) + P(red AND B) С 10 = P(A) × P(red from A) + P(B) × P(red from B) 2 D 5 = $P(HH) \times P(red from A) + P(HH) \times P(red from B)$ 1 Ε 2 $=\frac{1}{4} \times \frac{6}{10} + \frac{3}{4} \times \frac{3}{15}$ 4 5 F $= \frac{6}{40} + \frac{9}{60}$ 323 G 400 $= \frac{9+9}{60}$ $=\frac{18}{60}=\frac{3}{10}$



16 Three towns Ryeton, Tonbridge and Uphampton are represented on the diagram by the points labelled R, T and U, respectively.

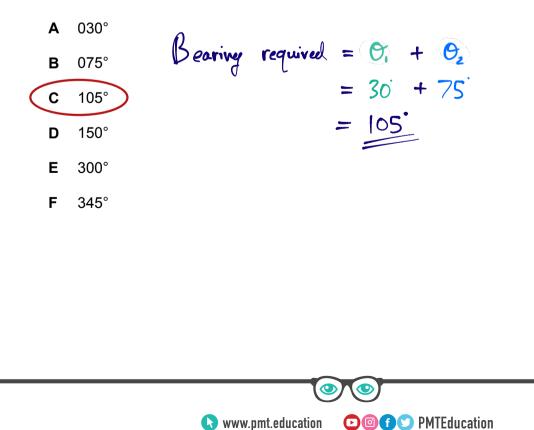


The distance from Tonbridge to Ryeton is the same as the distance from Tonbridge to Uphampton.

Uphampton is south of Tonbridge.

Ryeton is on a bearing of 210° from Tonbridge.

What is the bearing of Uphampton from Ryeton?





17 A list of five numbers has mean x, median y and range z.

 $x_{1}, x_{2}, x_{3}, x_{4}, x_{5}$

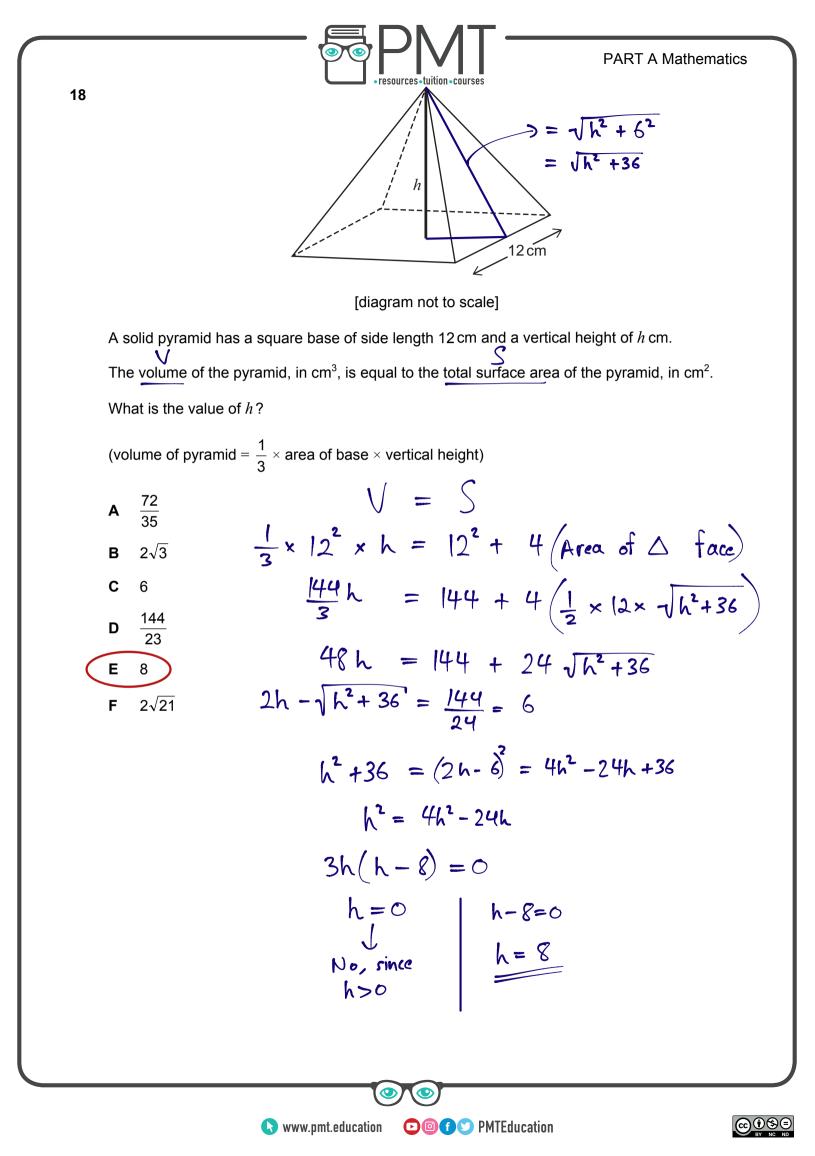
A sixth number is added to the list. This sixth number is greater than x.

Which of the following statements must be true?

- 1 The median of the six numbers cannot be one of the numbers in the list.
- 2 The mean of the six numbers is greater than *x*.
- **3** The range of the six numbers is greater than *z*.
- A none of them
- B 1 only
- C 2 only
 - D 3 only
 - E 1 and 2 only
 - F 1 and 3 only
 - G 2 and 3 only
 - H 1, 2 and 3

1) FALSE : It can it xe is bigger than all the others 2) TRUE . x => old mean, so x = will definitely raise the mean 3) FALSE: The range would only increase if x₆ was bigger than the largest of the original 5

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PART B Physics

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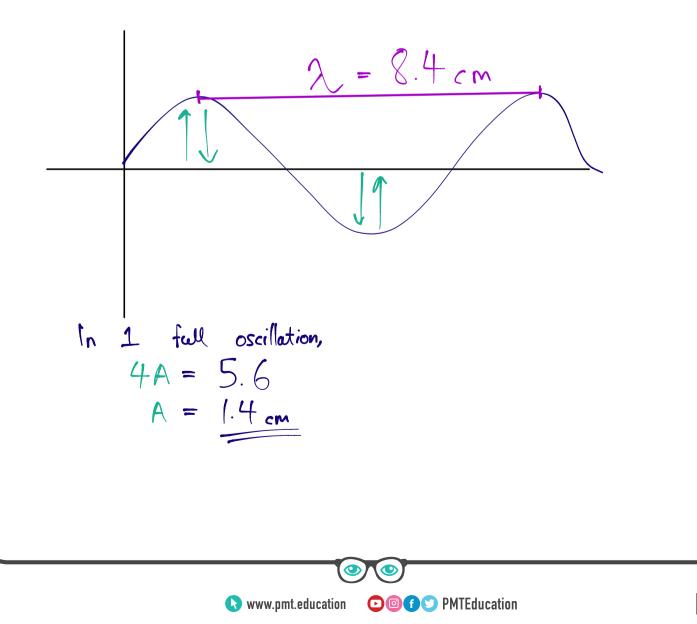
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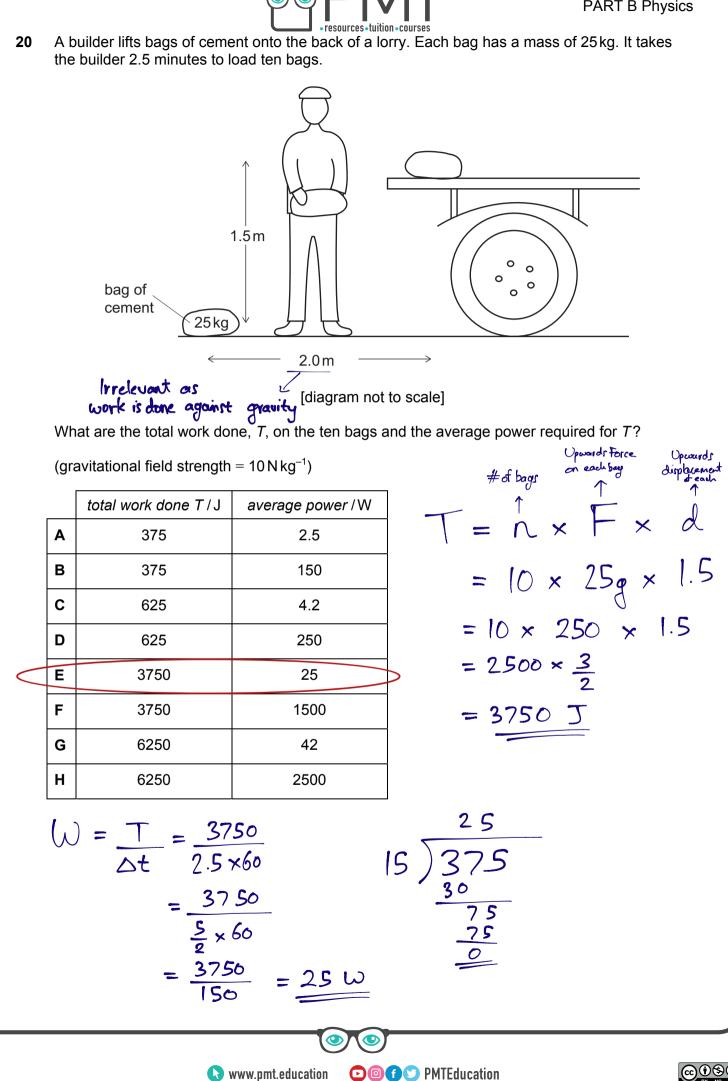
19 A transverse wave is travelling through a medium. The distance between successive wave peaks is 8.4 cm and the total distance travelled by a particle during one complete oscillation is 5.6 cm.

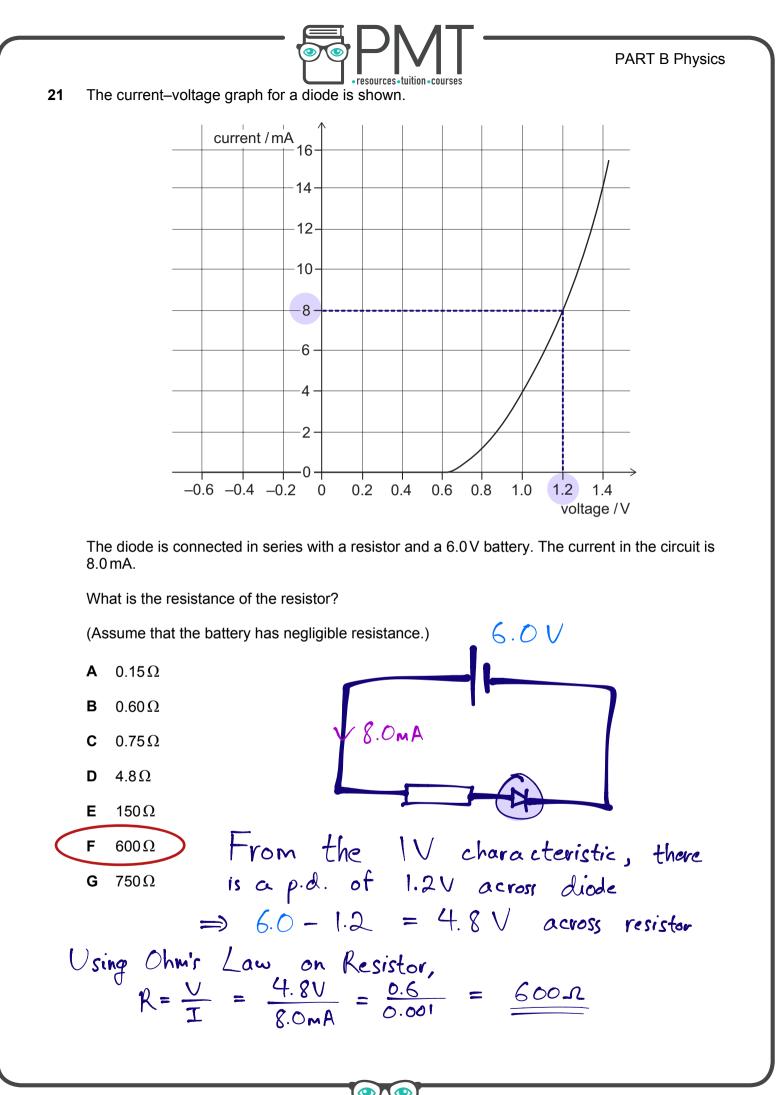
What is the amplitude and wavelength of the wave?

				_
		<i>amplitude</i> / cm	<i>wavelength</i> / cm	
	Α	1.4	4.2	
<	В	1.4	8.4	\triangleright
	С	2.1	2.8	
	D	2.1	5.6	
	Е	2.8	4.2	
	F	2.8	8.4	
	G	4.2	2.8	
	н	4.2	5.6	
				-



PART B Physics





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PART B Physics



$$\frac{\text{wavelength of P}}{\text{wavelength of Q}} = 1.0 \times 10^8 = \frac{f_{\alpha}}{f_r} \Rightarrow \frac{f_{\rho}}{f_{\alpha}} = 1.0 \times 10^8$$

Which row in the table shows the ratio of their speeds, the ratio of their frequencies, and identifies the possible natures of P and Q? 1. I fra ~ · MARC

		th EM Wower hour same spec	ed Q	has a higher frequency and must be MORE encryptic		
		speed of P speed of Q	frequency of P frequency of Q	nature of P	nature of Q	
<	Α	1.0	1.0×10^{-8}	microwave	X-ray	
	в	1.0	1.0 × 10 ⁻ °	microwave	radio wave	
	С	1.0	1.0 × 10 ⁸	infrared	ultraviolet	
	D 1.0		1.0 × 10 ⁸	visible light	infrared	
	Е	1.0 × 10 ⁸	1.0	gamma	X-ray	
	F	1.0×10 ⁸	1.0	gamma	infrared	
	G	1.0 × 10 ⁸	1.0×10^{16}	infrared	radio wave	
	н	1.0 × 10 ⁸	1.0 × 10 ¹⁶	visible light	ultraviolet	

23 A block of aluminium of mass 0.80 kg, initially at a temperature of -21 °C, is supplied with 54 000 J of thermal energy.

The specific heat capacity of aluminium is $900 \, \text{J kg}^{-1} \,^{\circ}\text{C}^{-1}$.

What is the final temperature of the block?

22

Nr =

fa

(Assume that there is no other transfer of energy between the block and the surroundings.)

A 27 °C	$Q = mc \Delta T$	
B 39 °C	$54000 = 0.80 \times 900 \times 07$	
C 54 °C	$\Delta T = \underline{54000}$ 720	
D 75°C	$= \frac{9 \times 600 \times 10}{9 \times 8 \times 10}$	
E 96 °C	9×8×10	
	= 75 °C	
	$T_{\text{final}} = T_{\text{initial}} + \Delta T$ $= -21 + 75 = 54^{\circ}C$	
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24 A light spring is used to support a range of loads.

The spring obeys Hooke's law. The system is in equilibrium.

Which of the following statements is/are correct?

- 1 The tension in the spring is directly proportional to the length of the spring.
- 2 The tension in the spring and the weight of the load it supports are a Newton's third law pair of forces.
- **3** When the extension of the spring is doubled, the energy stored in the spring increases by a factor of four.
- A none of them

В 1 only С 2 only D 3 only Е 1 and 2 only F 1 and 3 only G 2 and 3 only 1, 2 and 3 н False : Hooke's Law says Tension & EXTENSION False: Both those forces act on the same object (the load). The actual 3rd law pairs are: -> Weight of Load attracted to Earth -> Tension from spring pulling Lood up. Downwards force from load pulling the spring down -> Grav. attraction of Earth towards Load 3) True : Elastic Potential energy in spring = $U_{E} = \frac{1}{2}kx^{2}$ **D O** www.pmt.education



25 A water-tight cylinder with a thin, freely moving piston contains 2.0×10^{-3} m³ of trapped air at atmospheric pressure of 1.0×10^{5} Pa.

When the cylinder is submerged in water of constant density 1000 kg m^{-3} , the volume of air in the cylinder decreases to $4.0 \times 10^{-4} \text{ m}^{3}$.

The piston is at a depth h below the surface of the water and the water surface is open to the atmosphere.

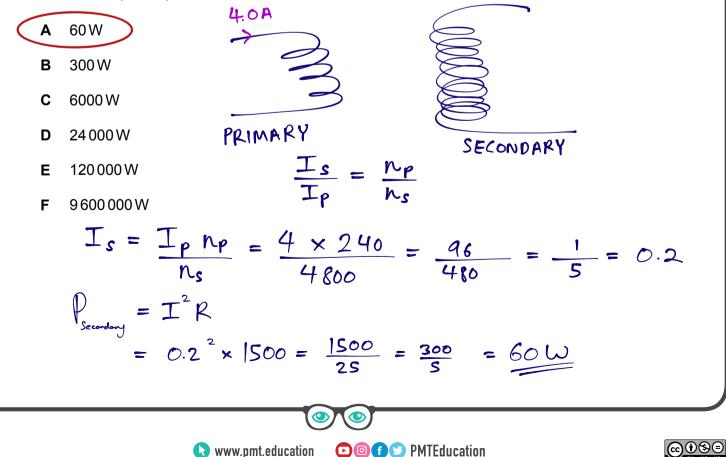
What is the depth h?

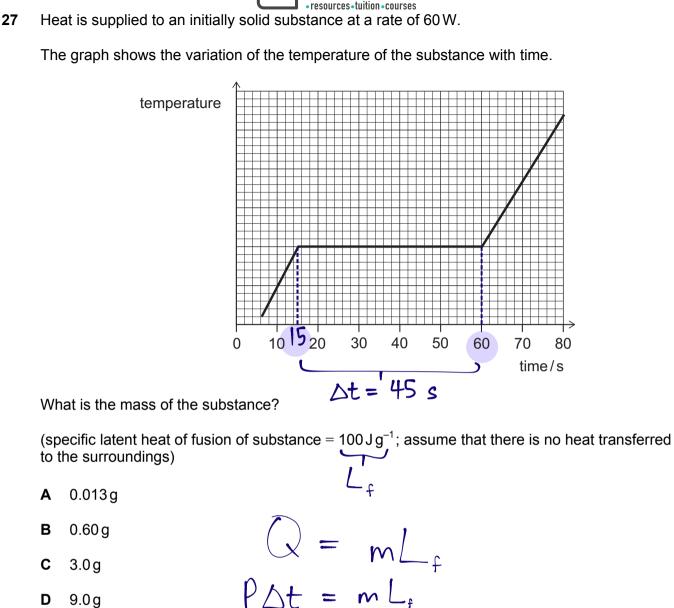
(gravitational field strength = 10 N kg^{-1} ; assume that the temperature of the air remains constant and that air is an ideal gas)

A	40 m	Using Boyle's Law:
В	50 m	$p_{1}V_{1} = p_{2}V_{2} \implies p_{2} = \frac{p_{1}V_{1}}{V_{2}} = \frac{1.0 \times 10^{5} \text{ Ra} \times 2.0 \times 10^{3} \text{ Ra}}{4.0 \times 10^{4} \text{ m}^{3}}$ $-\frac{2}{4} \times \frac{10^{2}}{10^{4}} = 5 \times 10^{5} \text{ Ra}$
С	60 m	$-\frac{2}{4} \times \frac{10^2}{10^4} = 5 \times 10^5 \text{ Pa}$
D	400 m	
Е	500 m	pressure due to water = increase in pressure $hpg = P_2 - P_1$
F	600 m	$h \times 1000 \times 10 = 5 \times 10^{5} - 1 \times 10^{5}$ $h = \frac{4 \times 10^{5}}{10^{4}} = \frac{40}{10} \text{ m}$

26 The secondary coil of an ideal, 100% efficient transformer is connected to a resistor by cables of total resistance 1500Ω . The current in the primary coil is 4.0 A. There are 240 turns in the primary coil and 4800 turns in the secondary coil.

What is the power produced as heat in the cables?





$$\Delta t = m L_{f}$$

$$m = \frac{P\Delta t}{L_{f}} = \frac{60 \times 45}{100}$$

$$= 27g$$

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27 g

36 g

Ε

F

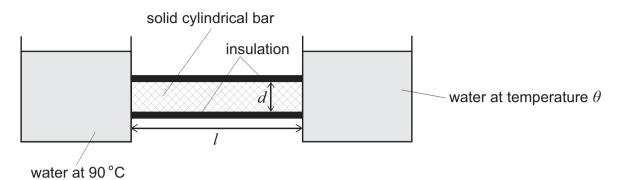
PART B Physics



28 Two tanks of water are connected by a solid cylindrical copper bar of length l and diameter d.

The bar is insulated.

One tank contains water at 90 °C and the other tank contains water at temperature θ .



For which of the following conditions is thermal energy conducted along the bar at the lowest rate?

		<i>l</i> / m	<i>d /</i> cm	θ/°C		
	Α	0.40	4.0	20		
	в	0.40	4.0	40		
	С	0.40	8.0	20		
	D	0.40	8.0	40		
	Е	0.80	4.0	20		
<	F	0.80	4.0	40		
	G	0.80	8.0	20		
	Н	0.80	8.0	40		
Sme	Bigger L ⇒ Longer distance for water to travel Slower Heat transfer Smaller d ⇒ Smaller cross section, Less hot water can flow through the tube each second Slower Heat transfer Larger () ⇒ Shallower Temperature gradient Slower Heat transfer					
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PART B Physics

resources • tu 29 A U-shaped permanent magnet rests on a balance.

A straight, horizontal wire of length 5.0 cm is fixed in position between the poles of the magnet, perpendicular to the horizontal magnetic field.

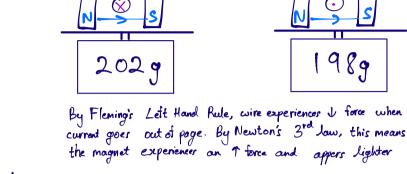
There is a current of 2.0 A in the wire and the reading on the balance is 202 g.

When the direction of the 2.0 A current is reversed, the reading changes to 198 g.

What is the strength of the magnetic field?

(gravitational field strength = 10 N kg^{-1})

- Α 0.020 T
- В 0.040 T
- С 0.20 T
- 0.40 T D
- Е 200 T
- 400 T F



2.00

Е	200 T	Since Force changer mass by ±20, the magnitude of
F	400 T	Since Force changer mass by $\pm 2g$, the magnitude of the force = 0.002 kg × 10N kg ⁻¹ = 0.02 N
		$F = B \downarrow I \implies B = \frac{0.02N}{0.05m \times 2.0A} = \frac{0.01}{0.0S} = \frac{0.2T}{0.2T}$
The	e radioactive isoto	pe X becomes the stable isotope Y after a succession of decays involving

30 only the emission of alpha and beta (β^{-}) particles.

During the decay of one nucleus from X to Y, a total of seven particles are emitted. It is known that more of these particles are alpha particles than beta particles.

The atomic number of X is Z and the mass number of X is A. $\stackrel{A}{=} X$

Which row in the table could give the atomic number and the mass number of Y?

					Each & emission decreases
		atomic number of Y	mass numb	er of Y	A here 4 and Z by 2
	Α	Z – 2	A – 12	E	A by 4 and Z by Q ach β -emission increases both
	В	Z – 5	A – 8		by I
<	o	Z – 8	A – 20		After n & emissions and m
	D	<i>Z</i> – 10	A – 24		β - emissions, $Z \longrightarrow Z + m - 2n$
	Е	Z – 11	A – 16		$\overrightarrow{A} \mapsto \overrightarrow{A} - 4n$
	n = 1	says $m+n$ 7-n $r \gg Z+7-$ $r \gg A-4n$			$n > m \implies n = 4, 5, 6$ when $n = 5$ $2 \mapsto 2 - 8$ (option C) $A \mapsto A - 20$ THE OTHER POSSIBILITIES AMENT OPTIONS
		() v	www.pmt.educatior		PMTEducation

PART B Physics

resources tuition courses 31 The kinetic energy of an object of mass 4.0 kg, travelling in a straight line, increases from 32 J to 200 J in 3.0 seconds due to a constant resultant force.

What is the value of this resultant force?

Α

В

С

2.0N

4.0N

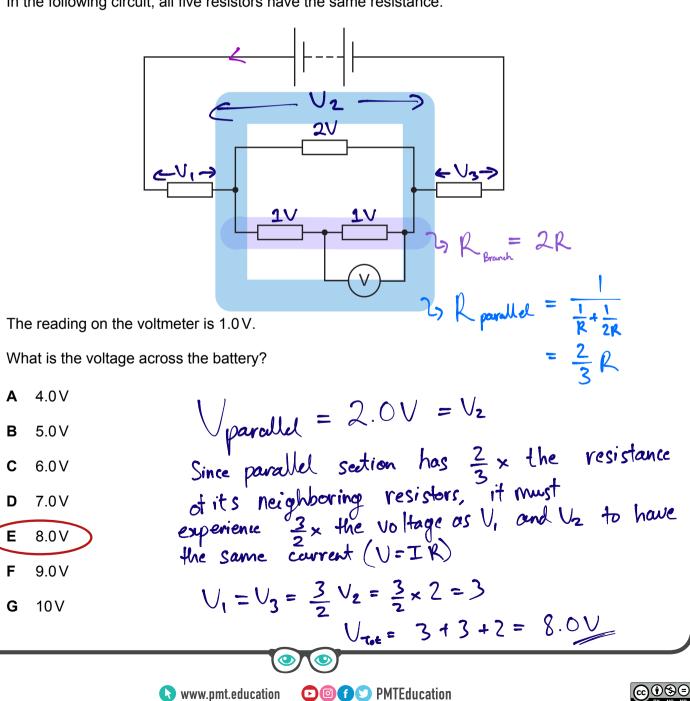
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Initially,

 $E_{K} = \frac{1}{2} m V_{i}^{2} = 32$ $V_{i} = \sqrt{\frac{32 \times 2}{4}} = 4 m s^{-1}$

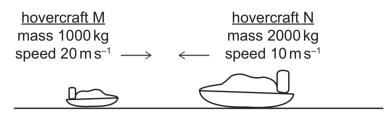
After 3.0 s, $E_{k} = \frac{1}{2}mv_{f}^{2} = 200$

- $V_{\rm f} = \sqrt{\frac{200 \times 2}{4}}$ = 10 ms⁻¹
- $\Delta v = 10 4 = 6$ Using Newton's 2nd Law, D 24 N Е 28 N $F = ma = m \Delta V$ F 56 N $=\frac{4\times6}{2}=\frac{8.0N}{2}$
- 32 In the following circuit, all five resistors have the same resistance.





33 Two hovercraft travel horizontally in opposite directions along the same straight line. The mass and speed of each hovercraft are shown in the diagram. Horizontal resistive forces acting on each hovercraft are negligible.

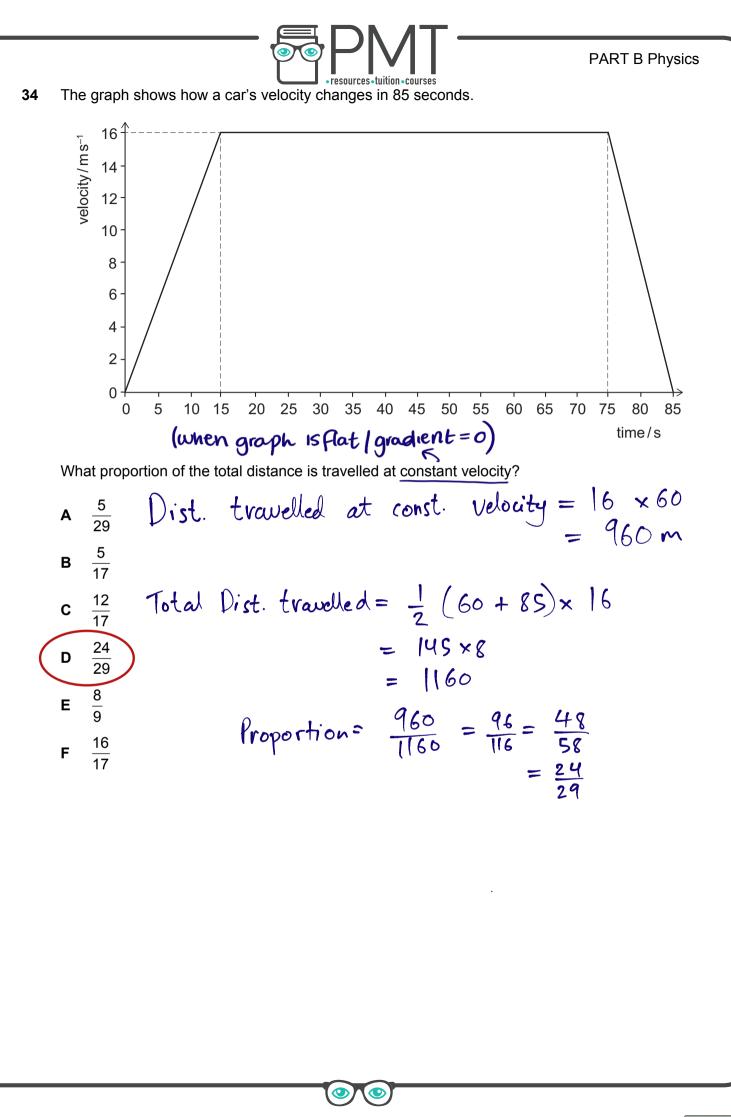


The hovercraft collide and stick together. The collision lasts for 0.10 s.

Just before the collision, what is the total kinetic energy and the magnitude of the total momentum of the two hovercraft, and what is the magnitude of the average force acting horizontally on each hovercraft during the collision?

	total initial kinetic energy /kJ	<i>total initial momentum</i> ∕kgms ^{−1}	average force on each hovercraft / kN
Α	100	0	2.0
В	100	0	200
С	100	$4.0 imes 10^4$	2.0
D	100	$4.0 imes 10^4$	200
Е	300	0	2.0
F	300	0	200
G	300	$4.0 imes 10^4$	2.0
н	300	4.0×10^4	200

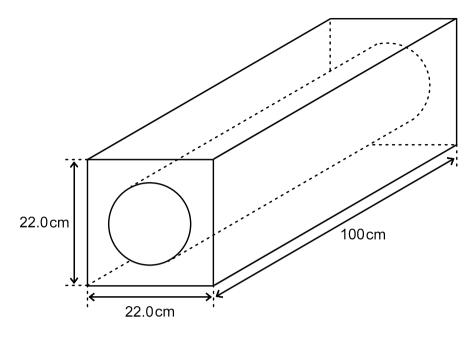
Total KE = $\frac{1}{2}m_{\mu}V_{\mu}^{2}$ + $\frac{1}{2}m_{\nu}V_{\nu}^{2}$ = $\frac{1}{2} \times 1000 \times 20^{2}$ + $\frac{1}{2} \times 2000 \times 10^{2}$ before collision = $\frac{1}{2}m_{\mu}V_{\mu}^{2}$ + $\frac{1}{2}m_{\nu}V_{\nu}^{2}$ = $\frac{1}{2} \times 1000 \times 20^{2}$ + $\frac{1}{2} \times 2000 \times 10^{2}$ = 300 kJTotal Initial = 1000x20 + 2000 x-10 momentum = 0 kgms⁻¹ For a single Hoveverent M, $F = \frac{\Delta p}{\Delta t} = \frac{20000}{0.10} = 2.0 \times 10^{5} N$ = 200 kN🕟 www.pmt.education 🛛 🖸 🕤 🕤 PMTEducation



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35 A block is designed with a cylindrical channel to accommodate a hot-water pipe. The block is 100 cm long and it has a square cross-section of side 22.0 cm with a cylindrical hole in the middle, as shown in the diagram:



[diagram not to scale]

The diameter of the cylindrical hole is 14.0 cm and the density of the material from which the block is made is $0.100 \, \text{g cm}^{-3}$.

What is the mass of the block?

(take
$$\pi$$
 to be $\frac{22}{7}$)
Volume of block = Volume of - Volume of Cylinder
= $22 \times 22 \times 100 - \pi \times 7^2 \times 100$
= $100 (22 \times 22 - 22 \times 7^2)$
D 33.0 kg
= $100 (22 \times 22 - 22 \times 7)$
= 132 kg
= $100 (22 \times 22 - 22 \times 7)$
= 132 kg
= 2200×15
G 1320 kg
= 2200×15
G 1320 kg
= 33000 cm^3
Density x volume
= Mass
Mass of block = $33000 \text{ cm}^3 \times 0.1 \text{ gcm}^3$
= 3.3 kg



36 A sample initially contains equal numbers of atoms of a radioactive isotope X and a stable isotope Y.

Isotope X has a half-life of 3 years and decays in a single stage to the stable isotope Y.

What is the ratio

number of atoms of X : number of atoms of Y

in the sample 6 years later?

A The sample contains only isotope Y.

B 1:7

1:4

D 1:3

С

E 7:4

Initially, $X : Y \equiv X : X = 1:1$ After 3 years (One half life) $\frac{X}{2} : Y + \frac{X}{2} \equiv \frac{X}{2} : \frac{3}{2}X = 1:3$ After 3 more years, (The 2nd half life) $\frac{X}{4} : Y + \frac{X}{2} + \frac{X}{4} \equiv \frac{X}{4} : \frac{7}{4}X = \frac{1:7}{4}$



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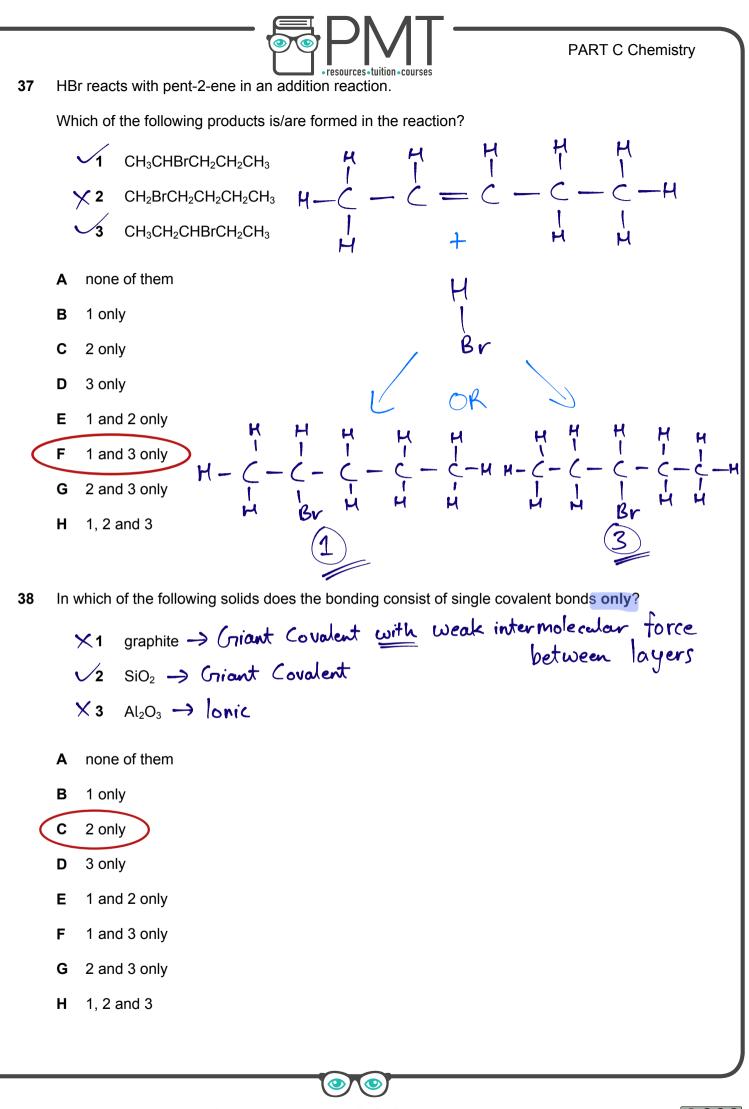


PART C Chemistry



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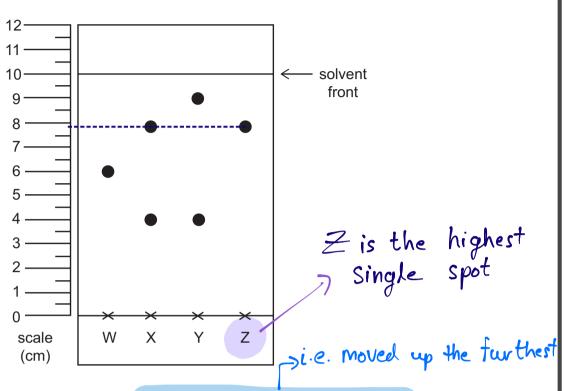


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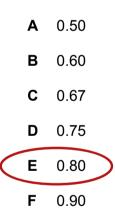
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PART C Chemistry

39 Four samples, labelled W, X, Y and Z, were investigated using paper chromatography with a solvent that caused any mixtures present to be fully separated. The results are shown in the chromatogram.



What is the R_f value of the spot with the strongest attraction to the mobile phase relative to the stationary phase **and** that is from a sample containing only one substance?



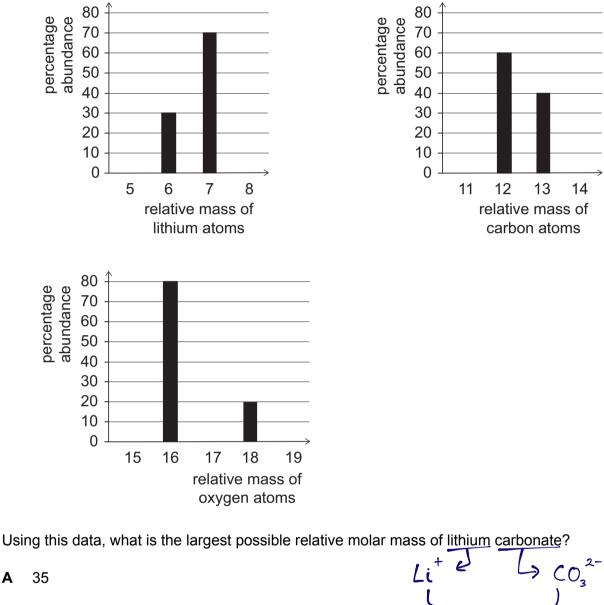
produced only one spot

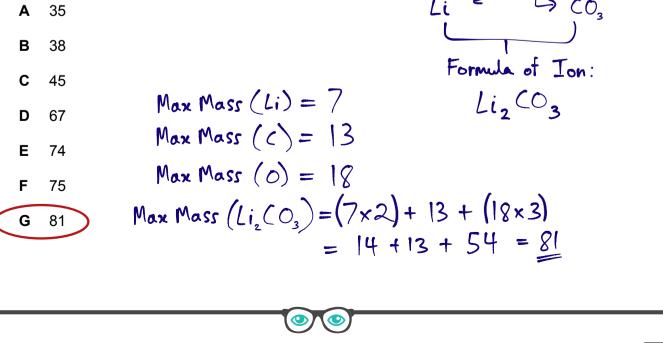
 $R_{f}(z) \approx \frac{8}{10} = 0.8$



40 A mass spectrometer is a device that can measure the mass of isotopes. It shows this data as a spectrum, giving both the relative mass and the percentage abundance of each isotope.

The charts indicate the relative mass and percentage abundance for lithium atoms, carbon atoms and oxygen atoms found in a sample taken from a nuclear reactor.

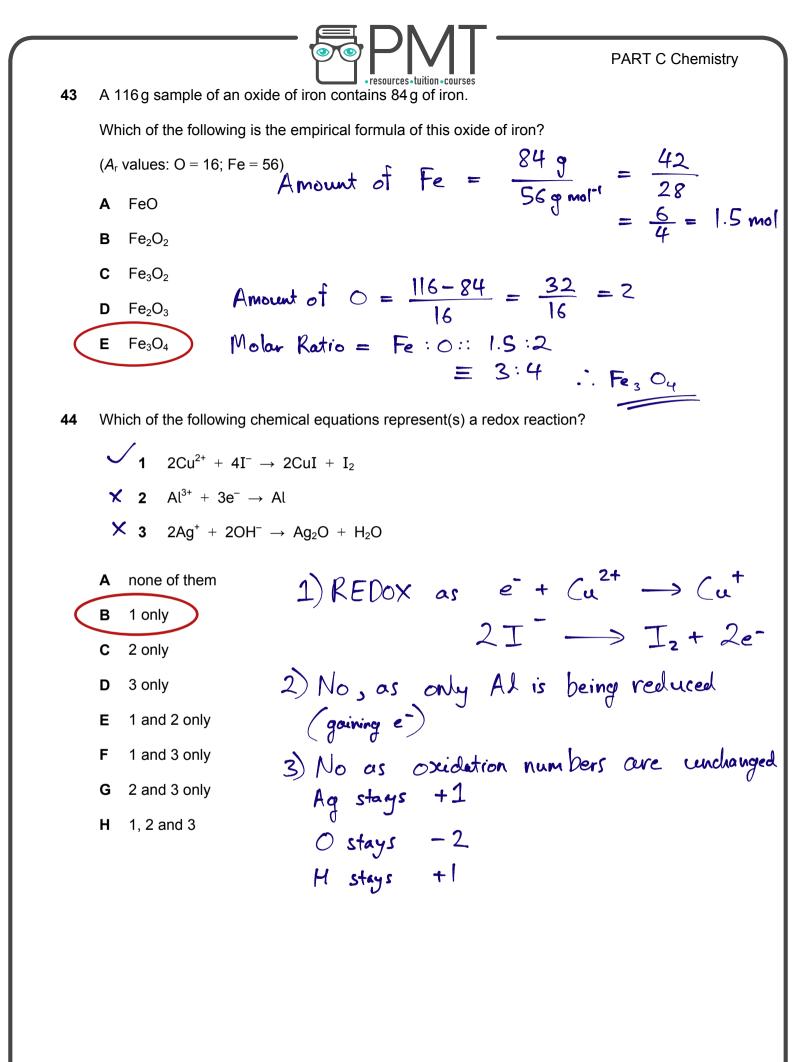




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			- 🗟 PN	PART C Chemistry				
41	The	e following information a	bout metals labelled P,	on «courses				
P and S MORE rea	an ctiu	Metals P and S of	ctrolysis, but not by reaction with carbon.					
thon car	bon	Metals Q and R	can be extracted by rea	action with carbon. \Rightarrow Q and R are LESS				
S MORE		Metal S forms positive ions more readily than metal P. Veactive than carbon						
reactive +	han	ρ. Metal R reacts w	rith the oxide of metal C	$R \implies R more reactive than Q$				
What is the order of reactivity of these four metals, starting with the most reactive?								
	Α	P, S, Q, R	P, S > Carb	on AND Q.R 2 Carbon				
	В	P, S, R, Q						
	С	Q, R, P, S	$\Rightarrow P, S > Q$, K				
	D	Q, R, S, P	Since S>P	and R>Q				
	Е	R, Q, P, S	=> S>P	> R > Q				
	F	R, Q, S, P	J					
	G	S, P, Q, R	MOST	LEAST				
	Н	S, P, R, Q						
42 A simple ion of an element with atomic number <i>x</i> has a mass number of $(2x + 2)$.								
The ion has a charge of -2 .								
How many protons, neutrons and electrons are present in this ion?								
.		protons	neutrons	electrons				
	Α	<i>x</i> – 2	<i>x</i> + 4	x – 2				
	В	<i>x</i> – 2	<i>x</i> + 4	<i>x</i>				
	С	<i>x</i> – 2	x + 4	<i>x</i> + 2				

[protons	neutrons	electrons						
Α	<i>x</i> – 2	<i>x</i> + 4	x – 2						
В	<i>x</i> – 2	<i>x</i> + 4	x						
С	<i>x</i> – 2	<i>x</i> + 4	x + 2						
D	x	<i>x</i> + 2	x – 2						
E	x	<i>x</i> + 2	x						
F	x	<i>x</i> + 2	<i>x</i> + 2						
# # # e	# protons = atomic number = x # neutrons = Mass number - atomic number = $x + 2$ } F # electrons = atomic number - charge = $x + 2$ }								
S www.pmt.education S www.pmt.education S www.pmt.education									



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D O

	PART C Chemistry
45	X is an anhydrous salt of iron containing one type of cation and one type of anion.
	An aqueous solution of X gives a white precipitate when aqueous barium chloride is added in the presence of hydrochloric acid. $\rightarrow +ve$ Sul phate Test
	On adding aqueous sodium hydroxide to an aqueous solution of X, a brown precipitate formed immediately. $free Fe^{2t}$ Test
	The relative atomic mass of iron is 56, and its atomic number is 26.
	What is the relative molar mass of X?
	(A _r values: C = 12; N = 14; O = 16; S = 32; Cl = 35.5; Br = 80)
	A 127 $Fe^{3+} SOu^{2-}$
	B 152
	$c_{162.5}$ $F_{e_2}(SOu)_3$
	D 208 $M_r(Fe_2(SO_4)_3) = 2 \times 56 + 3 \times (32 + 16 \times 4)$
	E 264 $ ' r(Fe_2(50u)_3) = 2\times 56 + 5\times (32+16\times 9)$
	$F 272 = 112 + 3 \times 96$
	$G_{360} = 112 + 288$
\langle	H 400 = 400
46	In the electrolysis of dilute sulfuric acid, hydrogen gas is formed at the negative electrode (cathode) and oxygen gas is formed at the positive electrode (anode).
	If 100 g of hydrogen gas is formed in the electrolysis of dilute sulfuric acid, what mass of oxygen gas is also formed? (A values is also formed? At cathode, $2H^+ + 2e^- \rightarrow H_2$ [REDUCTION]
	$(A_r \text{ values: } H = 1; O = 16)$
	A 50g At anode, $40H^- \rightarrow 2H_20 + O_2 + 4e^- [OXIDATION]$
	B 100g Multiply cathode equil by 2 to balance e
	c_{200g} Amount of $H_0 = 100 g$ T_{-1}
(C 200g Amount of $H_2 = \frac{100 \text{ g}}{2 \text{ g mol}^{-1}} = 50 \text{ mol}$
	E 1600 g
	1/2 mol of O2 produced for every mole of H2
	\Rightarrow 50 mol of O2 produced \equiv 50 x 32 x 1/2
	= 800g

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PART C Chemistry

47 0.005 mol of a chloride of element X was dissolved in water and then reacted with excess silver nitrate solution to form a precipitate of silver chloride, AgCl. This precipitate is the only product of this reaction that contains chlorine.

After filtering, washing and drying, the mass of the precipitate was recorded to be 1.435 g.

Which of the following could be the formula of the chloride of X?

(Mr value: AgCl = 143.5)
A
$$X_5Cl$$
 amount of Ag($l = \frac{1.435}{143.5} = 0.01 \text{ mol}$
B X_2Cl 0.005 mol of the Chloride contains 0.01 mol of
C XCl CL^- ions
D $XCl_2 \implies$ Formula of the compound must contain Cl_2
E XCl_5 $\frac{0.01}{0.005} = 2$ XCl_2

48 A chemical equation that represents the reaction of phosphorus with concentrated nitric acid is:

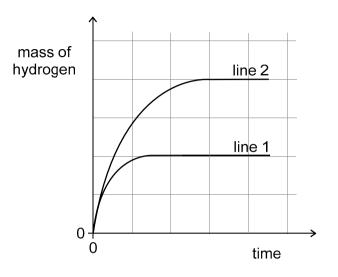
$$P_4 + wHNO_3 + H_2O \rightarrow xH_3PO_4 + yNO + zNO_2$$

What is the value of the sum
$$w + x + y + z$$
?
A 24
B 28
Comparing P ratio $\Rightarrow x = 4$
B 28
This means there are 12 mol of H atoms on RHS
 $\Rightarrow w = 12 - 2 = 10$
D 32
Comparing N ratio $\Rightarrow w = y + z = 10$
E 36
 $\therefore x + y + w + z = 4 + 10 + 10 = 24$



A 2.40 g lump of magnesium was added to 500 cm³ of a 2.00 mol dm⁻³ solution of HCl in a 49 conical flask that was on an electronic balance. The neck of the flask was plugged with cotton wool, and the decrease in mass of the flask and its contents was recorded at regular intervals.

The mass of the hydrogen released (equal to the mass loss recorded) was plotted against time. The result is line 1 on the graph.



Which of the following experiments performed under the same conditions would give line 2?

 $(A_r \text{ value: Mg} = 24.0)$

So

a 2.40 g lump of magnesium added to 500 cm^3 of 2.00 mol dm⁻³ H₂SO₄ Α

2.40 g of magnesium powder added to 500 cm^3 of 2.00 mol dm⁻³ HCl В

a 2.40 g lump of magnesium added to 1000 cm^3 of 2.00 mol dm⁻³ HCl С

a 4.80 g lump of magnesium added to 500 cm³ of 2.00 mol dm⁻³ HCl D

4.80 g of magnesium powder added to $500 \, \text{cm}^3$ of 2.00 mol dm⁻³ HCl Ε

$$Mg + 2H(1 \longrightarrow Mg(l_2 + H_2))$$

For line 1:
Amount of
$$Mg = \frac{2.4}{24} = 0.1 \text{ mol}$$
 Amount of $H(l = 0.5 \times 2 + 1.0 \text{ mol}) = 1.0 \text{ mol}$
 $\therefore Mg$ is the limiting reagent in this case.



50 A technician needs to separate three liquids (X, Y and Z) which have been accidentally mixed together. None of the liquids react with each other.

liquid	Х	Y	Z
boiling point / °C	65	51	100
<i>density</i> /g cm ⁻³	0.79	0.68	1.00

X and Z are miscible, but Y is immiscible with both X and Z.

The technician uses a separating funnel to separate the upper and lower layers of the mixture.

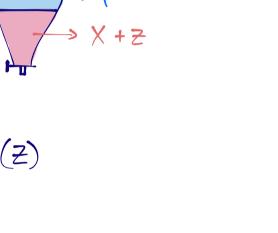
What should the technician do next to maximise separation of the three liquids?

- A distil the lower layer at 51 °C
- B distil the lower layer at 65 °C
- **C** distil the lower layer at 100 °C
- D distil the upper layer at 51 °C
- E distil the upper layer at 65 °C
- F distil the upper layer at 100 °C

Density (Y) < Density (X), Density (Z)

⇒ Upper layer is Y

 \therefore Must Distil the LOWER layer at <u>65°C</u> to obtain X as distillate.



> Separating Funnel

Since B.P. of X is lower, it boils off first.

	PIVII	PART C Chemistry
0,	n excess of the metal is added to	$1.0 \text{dm}^3 \text{of a } 1.0 \text{mol dm}^{-3}$ = 1.0 mol
copper added to sulfuric acid		
iron added to hydrochloric aci	d	
magnesium added to sulfuric	acid	
zinc added to hydrochloric aci	d	
roduce the largest, and the smallest	, theoretical mass of anhydrous	salt? Largest mass of salt
M_r values: CuSO ₄ = 160; FeCl ₂ = 12	Largest mass of salt occurs when salt has a high Mr. Similarly	
reaction that produces the largest mass of salt	reaction that produces the smallest mass of salt	for lowest
Cu and H₂SO₄	Fe and HCl	BUT
Cu and H ₂ SO ₄	Mg and H_2SO_4	
Fe and HCl	Zn and HCl	- Cu and Zn are quite unreactive Largest
Mg and H ₂ SO ₄	Fe and HCl	Largest
Mg and H ₂ SO ₄	Zn and HCl	fe (lz -> M/
Zn and HCl	Mg and H_2SO_4	$\therefore Fe(J_z \rightarrow M_r)$ $MgSO_4 \rightarrow Lowest$ Mr
	solution of the acid. copper added to sulfuric acid iron added to hydrochloric aci magnesium added to sulfuric zinc added to hydrochloric aci Which row in the following table ident broduce the largest, and the smallest M_r values: CuSO ₄ = 160; FeCl ₂ = 12 ⁻¹ reaction that produces the largest mass of salt Cu and H ₂ SO ₄ Cu and H ₂ SO ₄ Fe and HCl Mg and H ₂ SO ₄	copper added to sulfuric acidiron added to hydrochloric acidmagnesium added to sulfuric acidzinc added to hydrochloric acidWhich row in the following table identifies combinations of metal and a broduce the largest, and the smallest, theoretical mass of anhydrous M_r values: CuSO ₄ = 160; FeCl ₂ = 127; MgSO ₄ = 120; ZnCl ₂ = 136) $\overline{M_r}$ values: CuSO ₄ = 160; FeCl ₂ = 127; MgSO ₄ = 120; ZnCl ₂ = 136) $\overline{M_r}$ values: CuSO ₄ = 160; FeCl ₂ = 127; MgSO ₄ = 120; ZnCl ₂ = 136) $\overline{M_r}$ values: CuSO ₄ = 160; FeCl ₂ = 127; MgSO ₄ = 120; ZnCl ₂ = 136) $\overline{M_r}$ values: CuSO ₄ = 160; FeCl ₂ = 127; MgSO ₄ = 120; ZnCl ₂ = 136) $\overline{M_r}$ values: CuSO ₄ = 160; FeCl ₂ = 127; MgSO ₄ = 120; ZnCl ₂ = 136) $\overline{M_r}$ values: CuSO ₄ = 160; FeCl ₂ = 127; MgSO ₄ = 120; ZnCl ₂ = 136) $\overline{M_r}$ values: CuSO ₄ = 160; FeCl ₂ = 127; MgSO ₄ = 120; ZnCl ₂ = 136) $\overline{M_r}$ values: CuSO ₄ = 160; FeCl ₂ = 127; MgSO ₄ = 120; ZnCl ₂ = 136) $\overline{M_r}$ values: Cu and H ₂ SO ₄ $\overline{M_r}$ Cu and H ₂ SO ₄ $\overline{M_r}$ Cu and H ₂ SO ₄ $\overline{M_r}$ and HCl $\overline{M_r}$ and HCl $\overline{M_r}$ and H ₂ SO ₄

52 3.4 g of an impure sample of silicon tetrachloride is reacted with water. The mixture is then filtered and the resulting solution made up to 250 cm³.

 $SiCl_4(I) + 2H_2O(I) \rightarrow SiO_2(s) + 4HCl(aq)$

 12.5 cm^3 of this solution is neutralised exactly by 20.0 cm^3 of $0.100 \text{ mol dm}^{-3}$ sodium hydroxide. $= 0.02 \times 0.1 = 0.002 \text{ mol}$ What is the percentage purity of the silicon tetrachloride? of NOOH (M_r value: SiCl₄ = 170. Assume that the impurity does not react.) 0.002 mol of NaOH reacts with 0.002 mol HCL (1:1 radio) 12.5 cm³ contains 0.002 mol 1.7% Α В 2.5% $\int x 20$ $250 \text{ cm}^3 \text{ contains } 0.04 \text{ mol of } H(L \Longrightarrow \frac{0.04}{4} = \text{ of Si}(L_4) \text{ readed}$ С 10% 32% D Amount of "Si (Ly" used = $\frac{3.4}{170} = \frac{2 \times 1.7}{170} = 0.02 \text{ mol}$ 50% E $\frac{1}{2}$, purity = $\frac{0.01}{0.02}$ = $\frac{50!}{0.02}$ www.pmt.education **DOfS** PMTEducation

PART C Chemistry

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H '



$$N_2 \ + \ 3H_2 \ \rightarrow \ 2NH_3$$

 $N \equiv N + H - H - H - H$

H - H

 \rightarrow

What is the overall enthalpy change for the reaction?

(Bond enthalpies: $N \equiv N = 945 \text{ kJ mol}^{-1}$; $H - H = 435 \text{ kJ mol}^{-1}$; $N - H = 390 \text{ kJ mol}^{-1}$)

A
$$+90 \text{ kJ mol}^{-1}$$

53

- $-90 \, \text{kJ} \, \text{mol}^{-1}$ В
- С $+990 \, kJ \, mol^{-1}$
- $-990 \, \text{kJ} \, \text{mol}^{-1}$ D
- $+1080 \, kJ \, mol^{-1}$ Ε
- $-1080 \, \text{kJ} \, \text{mol}^{-1}$ F

$$F = -1080 \text{ kJ mol}^{-1}$$

$$Energy \text{ taken in} = \frac{Energy \text{ taken in}}{\text{to break bonds}} = \frac{Energy \text{ released}}{\text{to form bonds}}$$

$$= \left(\frac{1 \times 945 +}{3 \times 435} \right) - \left(\frac{6 \times 390}{6 \times 390} \right)$$

$$= \left(\frac{945 +}{1305} \right) - \left(\frac{2340}{2340} \right)$$

$$= 2250 - 2340$$

$$= -90 \text{ kJ mol}^{-1}$$

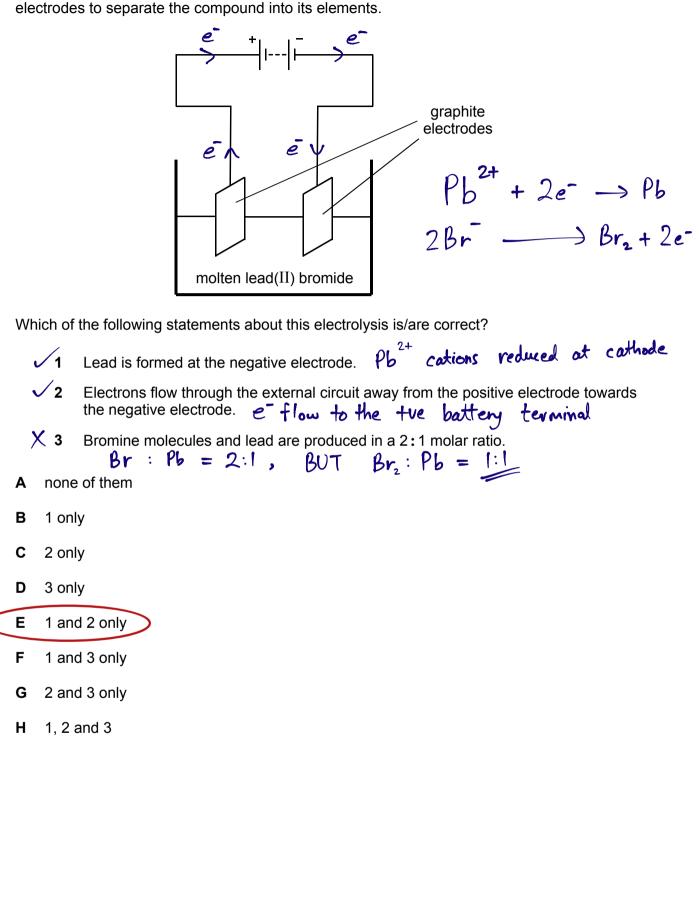
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PART C Chemistry

54 The diagram shows the electrolysis of molten lead(II) bromide, PbBr₂, using graphite electrodes to separate the compound into its elements.



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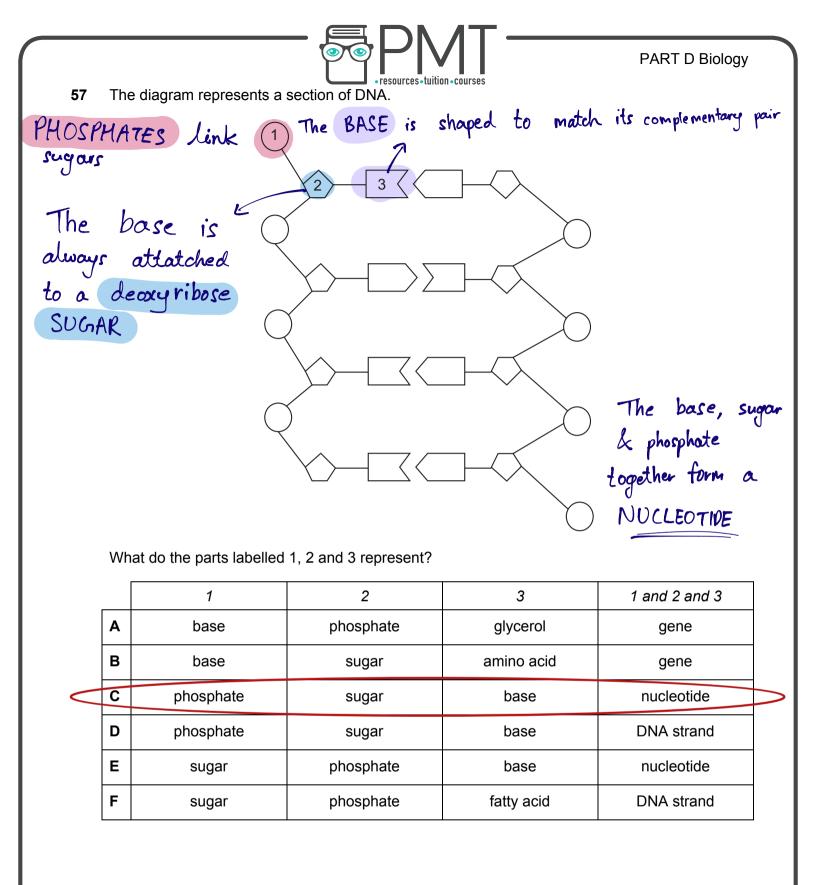
PART D Biology



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66	\ \ /bi	PART D Biology	
55		ich of the following statements is/are correct for both bacterial cells and sperm cells?	IFR
		X1 The cell can divide. Sperm cells are haploid and CAN'T DIVIDE FURTH	
		X3 The cell has a cell wall. Sperm cells DO NOT have a cell wall X4 The cell has a nucleus. Bacteria are prokaryotes	
		5 The cell can carry out respiration.	
	Α	2 only	
	в	4 only	
	С	1 and 2 only	
<	D	2 and 5 only	
	Е	3 and 4 only	
	F	3 and 5 only	
	G	1, 2 and 5 only	
	н	1, 3 and 4 only	
56	Whi	ich one of the following statements about cells or tissues is correct?	
	Α	Any adult stem cells can naturally give rise to all tissue. × Embryonic	
	в	Endocrine glands secrete enzymes into the bloodstream. X Hormones	
	с	Embryonic stem cells divide by meiosis to form all cell types. X Mitosis	
	D	Receptor cells send chemical impulses along neurones. × Electrical impulses	
\sim	Е	The stomach wall contains a tissue that enables movement of food.	
	-1	he stomach does both mechanical AND chemical digestion	L
	(he stomach abes both methanistic of	



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PART D Biology



58 Using the information in the table, which animal (A-F) is most at risk of extinction?

	animal	geographic range	habitat tolerance	population size	
	Α	restricted	broad	large	
	В	extensive	broad	large	
<	С	restricted	narrow	small	$\mathbf{>}$
	D	extensive	narrow	small	
	Е	restricted	broad	small	
	F	extensive	broad	small	

- -> restricted geographic range means there are Limited regions the species can travel to
- -> A narrow Habitat Tolerance means the species is less likely to accept new habitats if needed
- -> A small population size lowers the possibility of mutation / diversity within the species, hindering their ability to adapt to change.
- All these factors INCREASE risk of extinction.

D O



An ecologist used a $50 \text{ cm} \times 50 \text{ cm}$ square quadrat to estimate the number of meadow buttercups present in a field with an area of 50 m^2 . The quadrat was distributed randomly on ten 59 occasions in the field and the number of buttercups counted in each quadrat.

quadrat	number of buttercups
1	3
2	10
3	0
4	4
5	21
6	19
7	6
8	11
9	15
10	3

How many buttercups were there estimated to be in the $50 \, \text{m}^2$ field?

- 368 Α
- В 460
- С 920
- D 1840
 - Ε 4600
 - F 18400
 - G 45000

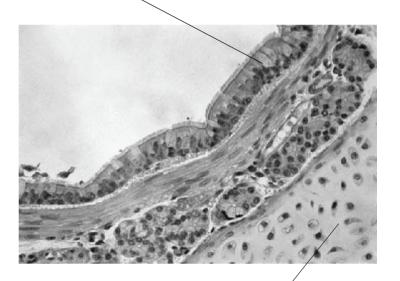
Average # of	$= \frac{3+10+0+4+21+19+6+11+15+3}{10}$
buttercups per quadret	
-	$=\frac{92}{10}=9.2$
Area of a quadrat =	$0.5 \mathrm{m} \times 0.5 \mathrm{m} = 0.25 \mathrm{m}^2$
# of quadrats in field =	$\frac{50 m^2}{0.25 m^2} = 200$
# of buttercups in field =	$200 \times 92 = 1840$
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60 The photomicrograph shows a section through the wall of a human bronchus, one of the tubes which carries air towards the lungs.

A student studying this structure annotated the photomicrograph by describing the cells found in two different layers in the wall of the bronchus.

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Two different types of cells are found in this single layer. Goblet cells, which synthesise and secrete the protein in mucus, and cells with cilia, which sweep mucus that has trapped dust and dirt away from the lungs.



A layer consisting of a group of similar cells. Each cell synthesises and secretes a protein that then surrounds the cells.

The student used these observations to write some conclusions.

Which of the following conclusions is/are correct?

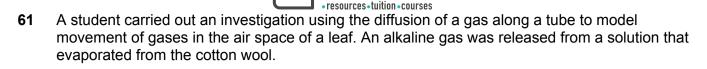
1 The two different layers are both tissues.

 \checkmark 2 The bronchus can be described as an organ.

 $\sqrt{3}$ Amino acids are found in the cytoplasm of cells in each layer.

none of them Α 2) TRUE: Tissues are a group of specialised cells with the same function В 1 only С 2 only 2) TRUE: Organs are groups of tissues (like layers 1 and 2) working together to do a D 3 only 1 and 2 only Ε specific job F 1 and 3 only G 1, 2 and 3 3) TRUE : Animal cell cytoplarms contain amino acids.

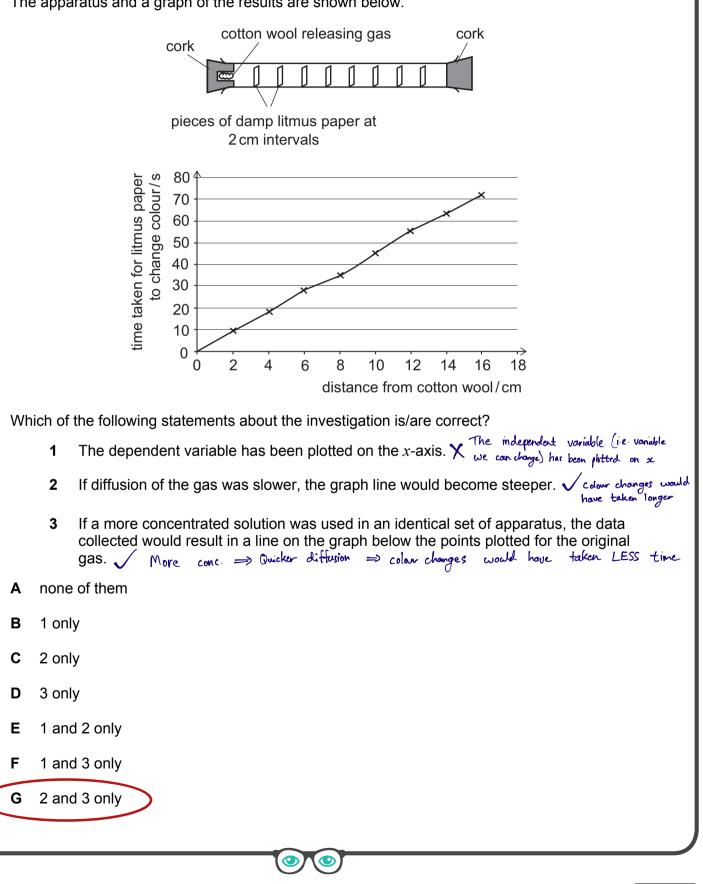
PART D Biology



As the gas diffused, it caused damp litmus paper to change colour.

The time was recorded when each piece of litmus paper changed colour.

The apparatus and a graph of the results are shown below.

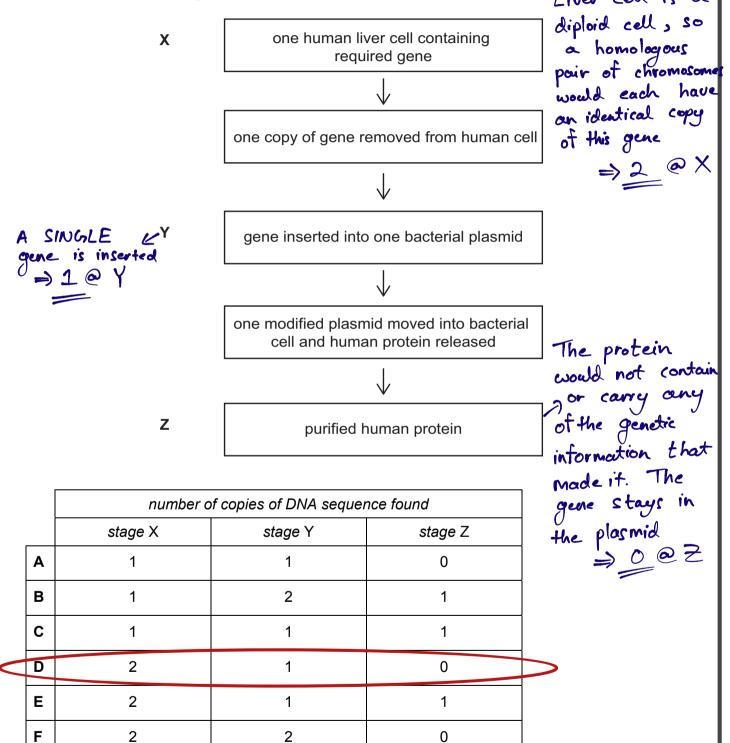




PART D Biology

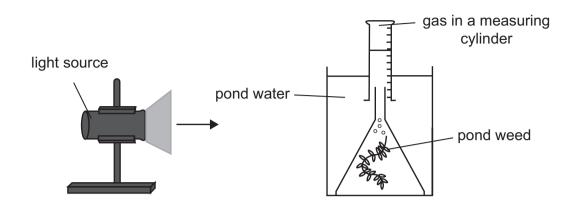
62 A bacterium was genetically modified by inserting a human gene into a plasmid. A plasmid is a small circle of DNA that can be used to transfer genes into bacterial cells. These modified bacteria then secrete the protein that the human gene codes for. The gene contains a sequence of bases that is not repeated anywhere else in its DNA and is not found in the DNA of other organisms. This sequence of bases is found in every allele of this gene.

Assuming that no mutations occur, how many copies of this base sequence would you expect to find at each of the stages X, Y and Z?





63 A student set up an experiment to measure the rate of photosynthesis, as shown in the diagram.



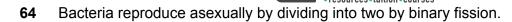
Data was collected and plotted on a graph.

If plotted, which of the following variables would give a gradient that is directly proportional to the rate of photosynthesis?

(All other variables were kept constant.)

		x-axis	y-axis		
	1	time	volume of CO ₂ released		
	2	time	number of gas bubbles released per minute		
	3	volume of oxygen released	time		
A	none o	f them 1) NO . 8	g radient = $(O_2 \text{ produced} \propto \text{Rate of} $ per second $\qquad \text{Respiration} $		
В	1 only		CO2 is TAKEN IN		
С	2 only		during photosynthesis		
D	3 only		a let let The solution		
E	1 and 2	2 only 2 NO · gra	2) NO. gradient = Bubbles produced ~ This includes per (minute) ² bubbles of O ₂ which is a product of photosynthesis that is oc rate of photosynthesis		
F	1 and 3	3 only			
G	2 and 3	3 only It is			
н	1, 2 an	d 3 that 1	is oc rate of photosynthesis		
	3) NO. gradient = 1 x rate of photosyn. Or persecond rate of photosyn. gradient would be inversely proportional to the rate of photosynthesis.				
		۷			
		🕟 www.pmt.education	$\bigcirc \bigcirc $		

PART D Biology



Which of the following statements is/are correct about binary fission in bacteria?

- \times 1 As it is a form of asexual reproduction, there cannot be any variation in the offspring.
- $\sqrt{2}$ It can lead to a repeated doubling in population size if there are no limiting factors.
- X 3 If binary fission occurs every 20 minutes, one bacterium would become 72² bacteria in 24 hours.
- 1) FALSE. Variation can occur through none of them Α 1 only random mutations. В 2) TRUE. The population grows exponentially С 2 only (y=2") without limiting factors D 3 only Ε 1 and 2 only 3) FALSE. 1 and 3 only F # of 20 mins in $24h = \frac{24h}{\sqrt{2}h} = 72$ G 2 and 3 only After 72 binary fission periods н 1, 2 and 3
- 65 Which of the following may stay the same when a mutation occurs in a human gene that codes for a protein?
 - \checkmark 1 the genotype of the organism's offspring
 - \checkmark 2 the phenotype of the organism
 - $\sqrt{3}$ the sequence of amino acids in the protein

Α none of them No guarantee the offspring will inherit the mutated gene CAN - MAPPEN 1 only В С 2 only D 3 only CAN Depends on the role of the protein 2. HAPPEN 1 and 2 only Е F 1 and 3 only G 2 and 3 only Silent mutations don't affect the 3. HAPPEN н 1, 2 and 3 Sequence www.pmt.education

66 A child is affected by a dominant genetic condition. All of his cells have the same genotype. The allele associated with this condition is not present in either of his parents.

Which of the following statements could explain this?

- \checkmark **1** A mutation occurred during meiosis in his father.
- \times 2 A random mutation occurred in his DNA after he was born.
- \times Both of his grandmothers had the condition.
- A none of them
- B 1 onlyC 2 only
- D 3 only
- E 1 and 2 only
- F 1 and 3 only
- G 2 and 3 only
- H 1, 2 and 3
- 1 TRUE. The zygote would have received this from the father's chromosomes. And all of the child's cells would have reflected this genotype by mitosis
 - 2. FALSE. He would have had some cells of the original genotype.
 - 3. FALSE. The parents would have also shown the condition if it were dominant



67 Phosphatase enzymes break the bond between a phosphate group and the rest of a molecule.

Phenolphthalein phosphate (PPP) is colourless and is a substrate for phosphatase enzymes.

- The phosphate group of PPP can be removed by the phosphatase enzyme to produce the indicator phenolphthalein.
- Phenolphthalein goes pink in alkaline conditions.

The SAME enzyme – Five tubes were set up with equal concentrations of PPP and a plant phosphatase enzyme.

different ph in each test trube Each tube was at a different pH. The tubes were incubated at the same temperature. After 10 minutes Na₂CO₃ was then added to each tube until a pH of 9.5 was reached. The colour was observed immediately and then again after another 5 minutes.

	tube	1	2	3	4	5
	pH of the initial reaction	3.2	4.2	5.2	8.2	9.2
A Stronger pink means	colour immediately after Na ₂ CO ₃ addition	pale pink	pink	dark pink	colourless	colourless
more indicator I was produced from the initial	– colour after another 5 minutes	pale pink	pink	dark pink	colourless	colourless

reaction

Which of the following statements is/are correct?

 \checkmark 1 Na₂CO₃ solution inhibited the activity of the phosphatase enzyme.

× 2 More substrate was produced in conditions of high acidity compared to low acidity.

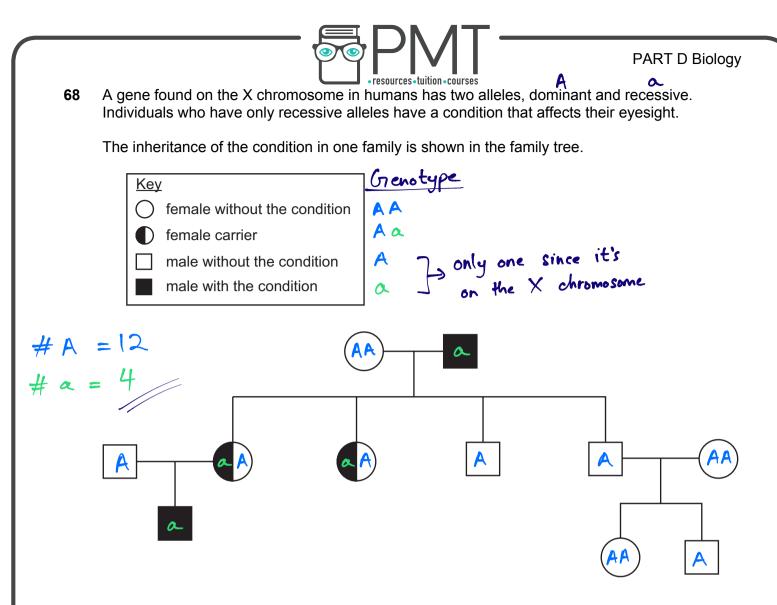
 \checkmark 3 The optimum pH for this plant phosphatase must be 5.2.

A none of them

1) TRUE as the results show no phenolphthalein was produced in alkaline conditions 1 only B С 2 only D 3 only 2) FALSE. More phenolph thalein was produced in low acidity (as shown from indicator colour) 1 and 2 only Ε F 1 and 3 only 2 and 3 only 3) FALSE. The optimum pH is certainly NEAR 5.2, but more measurements would be needed to determine its exact value. G 1, 2 and 3 н

D O





If one living skin cell from each member of this family were analysed, how many of the following alleles would be found?

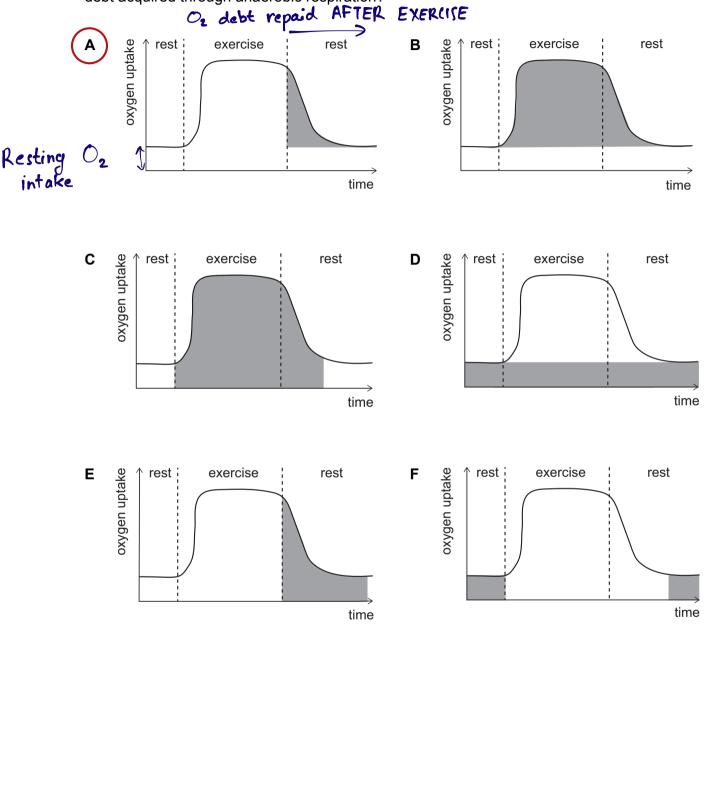
		number of copies of the recessive allele	number of copies of the dominant allele	
	Α	2	7	
	В	2	12	
	С	2	14	
V	D	4	7	
	E	4	12	
	F	4	14	
	G	6	7	
	H	6	12	

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PART D Biology

69 The graphs below show the oxygen uptake before, during and after a period of strenuous physical exercise.

Which shaded area correctly represents the additional oxygen taken in to repay the oxygen debt acquired through anaerobic respiration?



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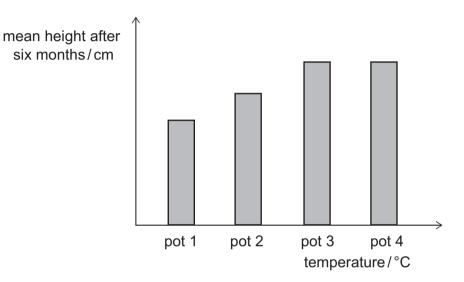


70 An investigation was carried out using clones of one plant.

Twenty plants of the same initial height were selected and divided into four equal groups. Each group was grown for six months and their environments were controlled as follows:

- kept at a different temperature to each other
- grown in a pot with an equal mass of soil with the same nutrients
- watered with an equal volume of water
- kept in the same light intensity

After six months, the height of the plants was measured. The mean height of the plants in each pot was calculated. The results are shown in the chart.



Which of the following statements could explain the results?

- 1 The difference in mean height between plants in pot 1 and plants in pot 2 could be due to the environment.
- 2 The mean height of plants in pot 4 equals that of plants in pot 3 because of another environmental factor in addition to temperature.
- 3 The mean height of plants in pot 3 equals that of plants in pot 4 because of the genotype of the plants.

Α	none of them	1) TRUE. The independent variable (Temperature)
В	1 only	1) TRUE: The independent variable (Temperature) 15 a part of the environment
С	2 only	
D	3 only	2) TRUE: Another Variable may be acting as a Limiting factor preventing additional growth
Е	1 and 2 only	
F	1 and 3 only	3) TRUE: They are identically affected by the same limiting factor as they're clones. (eg.
G	2 and 3 only	Same limiting tactor as they're clones. (eg.
Н	1, 2 and 3	one does not tend to grow taller with the same nutrients due to variation)



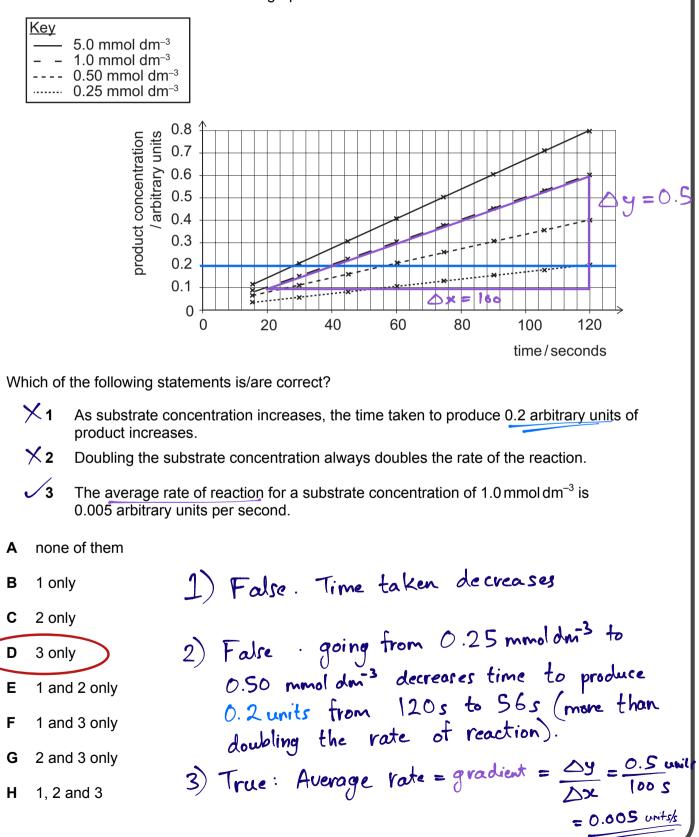


An investigation was carried out on the effect of substrate concentration on an enzyme-controlled reaction.

Four different concentrations of substrate were tested. In each case, the concentration of product was measured at regular intervals following the introduction of the substrate.

All the other variables were kept constant.

The results obtained are shown on the graph.



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72 The sequence of a coding section of DNA is shown below. $1^{4} 2^{4} 3^{4} 5^{4} 6^{4}$

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The number of each type of amino acid coded for by this sequence of DNA bases is shown in the table.

Each amino- acid is made	- amino acid coded for	number of this type of amino acid present
from 3 base pairs	Arg	1
	Leu	6
	Pro	2
	Trp	5
	Val	3

What is the correct sequence of amino acids coded for by this sequence of DNA?

A Leu-Trp-Trp-Leu-Arg-Pro-Pro-Leu-Leu-Val-Leu-Val-Val-Trp-Trp-Trp

C Leu-Val-Trp-Trp-Leu-Arg-Pro-Pro-Leu-Leu-Trp-Leu-Val-Val-Trp-Trp

X D Leu-Trp-Trp-Leu-Arg-Pro-Pro-Leu-Leu-Val-Leu-Val-Val-Trp-Trp-Trp-Leu

K Leu-Trp-Trp-Leu-Arg-Pro-Pro-Leu-Leu-Trp-Leu-Trp-Val-Val-Val-Leu

F Leu-Leu-Trp-Trp-Leu-Arg-Pro-Pro-Leu-Leu-Trp-Leu-Val-Val-Trp-Trp-Val

G Leu-Val-Trp-Trp-Leu-Leu-Pro-Pro-Leu-Leu-Trp-Leu-Val-Val-Trp-Trp-Leu

(H) Leu-Val-Trp-Trp-Leu-Arg-Pro-Pro-Leu-Leu-Trp-Leu-Val-Val-Trp-Trp-Leu

To solve this at a GCSE level, eliminate the wrong options -> The 2nd and 3rd triplets are NOT identical. So the 2nd and 3 amino acid CANT be the same. 1st and last triplets ARE identical. So the 1st and last -) The aminoacids MUST be the same -> The 5th and 6th triplets are NOT identical. So the 5th and 6 amino acid CANT be the same.

D O



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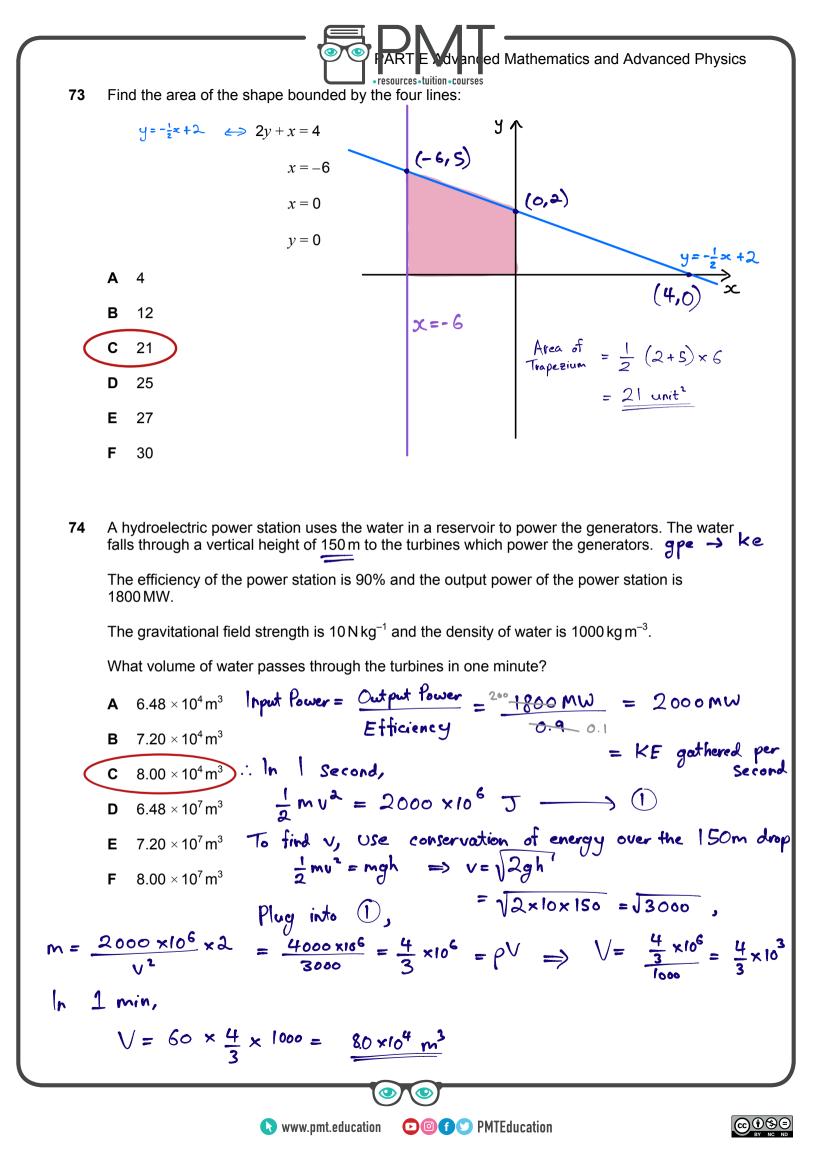


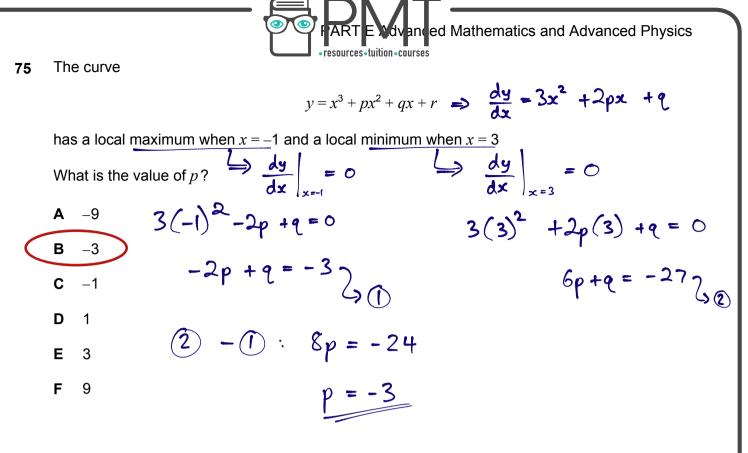
PART E Advanced Mathematics and Advanced Physics



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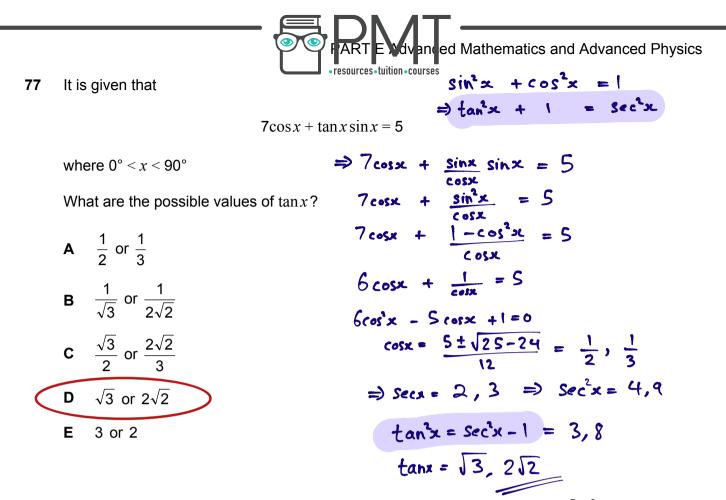
76 A car P of mass 1000 kg is travelling north at 30 m s^{-1} along a straight, horizontal road when it hits another car Q which is directly ahead of P and travelling in the same direction. Car Q has a mass of 500 kg and is travelling at 20 m s^{-1} .

The collision lasts for 0.20 s and immediately after the collision car Q is moving north at $30 \,\mathrm{m\,s^{-1}}$.

What is the speed of P immediately after the collision and what is the size of the average resultant force that acts on Q during the collision?

(Assume that no external forces act on the cars during the collision.)

	Γ	speed of $P/m s^{-1}$	average force on Q / N	Before:		
	Α	20	25000	$Q \bullet p_{q} = 500 \times 20 = 10000 \text{ kgms}'$		
	В	20	50 000	$\rho = 1000 \times 30 = 30000 \text{ kgms}^{-1}$		
	С	20	100 000	P P Total Initial momentum = 40 000 kgms		
	D	20	125 000	After:		
\triangleleft	E	25	25000	p = 500 x 30 = 15000 kgms'		
	F	25	50 000			
	G	25	100 000	P By conservation of momentum, $P_p = 40000 - 15000 = 25000$		
	Η	25	125 000	= 1000 kg x 2 Sms		
Change in momentum of $Q = \Delta p_{Q} = 15000 - 10000$ = 5000 kgms ⁻¹ => $F = \frac{5000}{0.20} = 25000 N$ = $F \Delta t$						
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78 A metal wire of length 0.50 m has a uniform cross-sectional area of 4.0×10^{-7} m². There is a current of 4.0 A in the wire.

What is the potential difference across the ends of the wire?

(resistivity of the metal = $1.6 \times 10^{-7} \Omega m$)

- **A** 0.05∨
- **B** 0.20 V
- **C** 0.80 V
- **D** 3.2 V
- **E** 5.0 V
- **F** 20V

=	0.4 × 0.5	11	0.20
V = IR			
$= 4.0 \times 0.20$			
$=\frac{4}{5}=\frac{0.8V}{=}$			

 $R = \frac{PL}{A} = \frac{1.6 \times 10^{-7} \times 0.50}{4.0 \times 10^{-7}}$

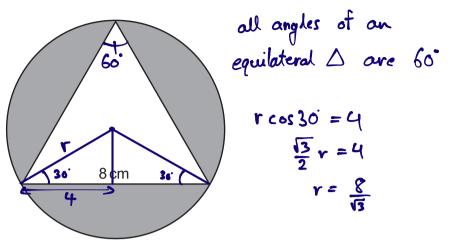
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Advanced Mathematics and Advanced Physics

79 An equilateral triangle of side 8 cm is drawn so that its vertices lie on the circumference of a circle, as shown in the diagram.

ART

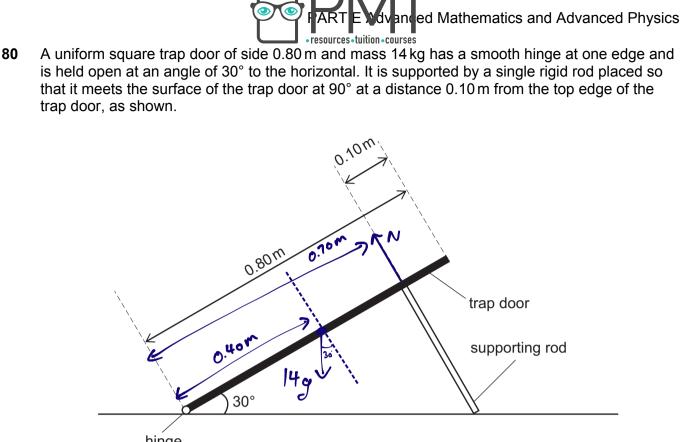


What is the total of the three areas shaded in the diagram, in cm²?

A $8(2\pi-3)$	= Area of circle - Area of \triangle
B $24(\pi - \sqrt{3})$	= Area of circle - Area of \triangle = $\pi \left(\frac{8}{\sqrt{3}}\right)^2 - \frac{1}{2} \times 8 \times 8 \times \sin 60^\circ$
C $48(4\pi - \sqrt{3})$	$= \frac{64}{2}\pi - 16\sqrt{3}$
D $\frac{16}{3}(4\pi - 6 - 3\sqrt{3})$	
E $\frac{16}{3}(4\pi - 3\sqrt{3})$	$=\frac{16}{3}\left(4\pi-3\sqrt{3}\right)$

▶ Image: PMTEducation





hinge

What is the normal contact force exerted on the trap door by the rod?

(gravitational field strength = 10 N kg^{-1})

Α	40 N	Taking Moments about the Hinge,
В	35√ <u>3</u> N	Sum of 5 = Sum of C
С	40√3N	$0.70 \text{ N} = 0.40 \times 149 \cos 30^{\circ}$
D	80 N	Ŷ
Е	80√3 N	$N = \frac{4}{7} \times \frac{14 \times 10 \times \sqrt{3}}{2}$
F	$280\frac{\sqrt{3}}{3}N$	$= 40\sqrt{3}$

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▶ Image: PMTEducation

Which one of the following is the real solution of the equation 81

$$3 \times 5^{2x+1} - 5^{x} - 2 = 0$$
Let $y = 5^{x}$

$$\Rightarrow 3 \times 5y^{2} - y - 2 = 0$$

$$15y^{2} - y - 2 = 0$$

$$15y^{2} - y - 2 = 0$$

$$y = \frac{1 \pm \sqrt{1 + 120}}{30}$$

$$y = \frac{12}{30}, \frac{-10}{30} = \frac{2}{5}, \frac{-1}{3}$$
Since $y = 5^{x} > 0$ for all real x ,
$$y = 5^{x} = \frac{2}{5}$$

$$x = \log_{5}(\frac{5}{2})$$

$$y = 12x + 12x +$$

()

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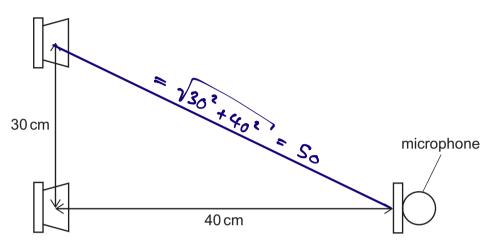
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82 Two small loudspeakers are placed side by side 30 cm apart.

They are connected to the same signal generator so that they emit sound of frequency 400 Hz in phase with one another.

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The sounds both reach a microphone placed 40 cm directly in front of one of the two loudspeakers as shown.



What is the phase difference between waves from the loudspeakers as they arrive at the microphone?

```
(speed of sound = 320 \text{ m s}^{-1})
                             \lambda = \frac{V}{f} = \frac{320}{400} = 0.8 \,\mathrm{m}
Α
     30°
В
     36°
                             Path difference between = 50-40
the 2 waves = 10 cm
С
     45°
D
     60°
                                                                           =\frac{1}{R}^{th} of a wavelength
Е
     72°
                      ... Phase difference is \frac{1}{8} of a cycle out of sync,
As phase difference is an angular measure:
F
     90°
G
     120°
                                            =\frac{1}{8}\times 360^{\circ}
                                            = 45'
```

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83 For a particular function f(x), it is given that:

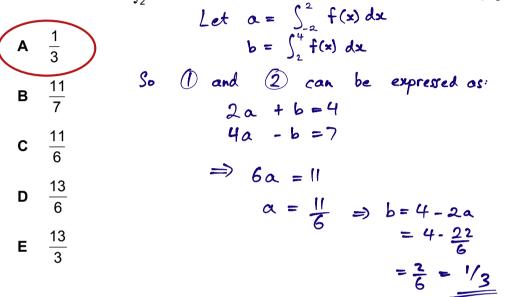
$$\int_{-2}^{2} 2f(x) dx + \int_{2}^{4} f(x) dx = 4$$

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and also:

$$\begin{array}{c}
\textcircled{(x)} & & & \\
\textcircled{(x)} & & \\
\textcircled{(x)} & & \\
\swarrow & & \\
& \\
\blacksquare & \\$$

Find the value of $\int_{2}^{4} f(x) dx$



84 An astronaut on the Moon throws a ball vertically upwards. The ball has a mass of 2.0 g and is thrown upwards at 80 m s^{-1} .

What is the maximum height gained by the ball?

(gravitational field strength close to the Moon's surface = $1.6 \,\mathrm{N \, kg^{-1}}$)

	Α	25 m	$v^2 = u^2 + 2\alpha s$	S = ? u = 80
	В	50 m	$0 = 80^2 + 2 \times 1.6 \times 5$	$\mathbf{v} = 0$
	С	320 m	s = 6400	$\alpha = 1.6$
<	D	2000 m	3.2 - 2×32×100	
	Е	3200 m	$= \frac{2 \times 32 \times 100}{32 \times 0.1}$	
	F	4000 m	= 2000 M	

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85 Given that

$$f(x) = \int_0^x (3+2t)^7 dt$$

what is the coefficient of x^4 in the expansion of f(x) in powers of x? The x^4 term in f(x) only comes from the t^3 term in $(3+2t)^7$ Α 70 В 162 2835 4 С $\binom{7}{3}$ $\binom{3}{3}$ $\binom{2}{2}$ $\binom{2}{2}$ 3024 D $= \frac{7!}{(7-3)! 3!} \times 8! \times 8t^{3}$ Е 5670 F 15120 $= \frac{7 \times 6 \times 5}{3 \times 2 \times 1} \times 81 \times 8t^{3}$ G 22680 $= 35 \times 81 \times 8t^{3}$ when integrated, $= \frac{35 \times 81 \times 8}{4} = \frac{4}{70 \times 81}$ = 5670 t⁴ Plug in bounds $= 5670x^2 - 5670(0)^4$ $= 5670x^{2}$

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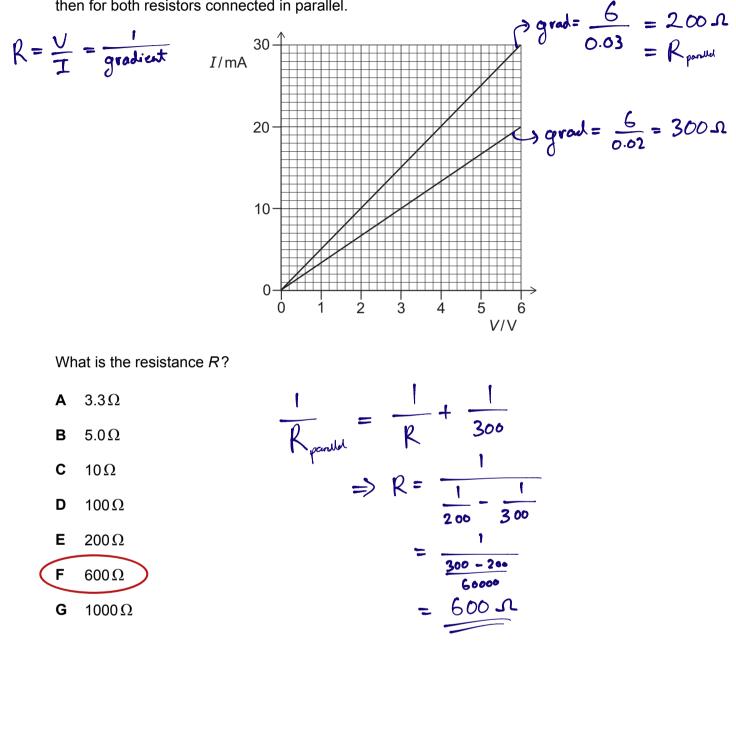
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86 A student has one 300Ω resistor and another resistor of resistance *R*.

The student plots a graph of current *I* against potential difference *V* for the 300Ω resistor and then for both resistors connected in parallel.

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▶ Image: Contraction PMTEducation



E Advanced Mathematics and Advanced Physics

resources tuition courses 87 The three internal angles in a triangle are α , β and θ , and

$$3\tan \alpha - 2\sin \beta = 2 \longrightarrow (1)$$
$$5\tan \alpha + 6\sin \beta = 8 \longrightarrow (2)$$

What is the value of θ in degrees?

What is the value of
$$\beta$$
 in degrees?
A 15
B 45
C 75
P lug tand =1 into (D),
E 135
Since $\alpha + \beta + 0 = 180$;
 $\beta = 180 - 45^{\circ} - 30^{\circ} = 105^{\circ}$

ART

88 A light, vertical, copper wire of length 2.4 m and uniform cross-sectional area
$$2.0 \times 10^{-6} \text{ m}^2$$
 supports a load of mass 4.0 kg.

The Young modulus of copper is 1.2×10^{11} Pa. = What is the strain energy in the wire? $=\frac{1}{2} k x^2$

(gravitational field strength = 10 N kg^{-1} ; assume that the wire obeys Hooke's law and that the cross-sectional area remains constant)

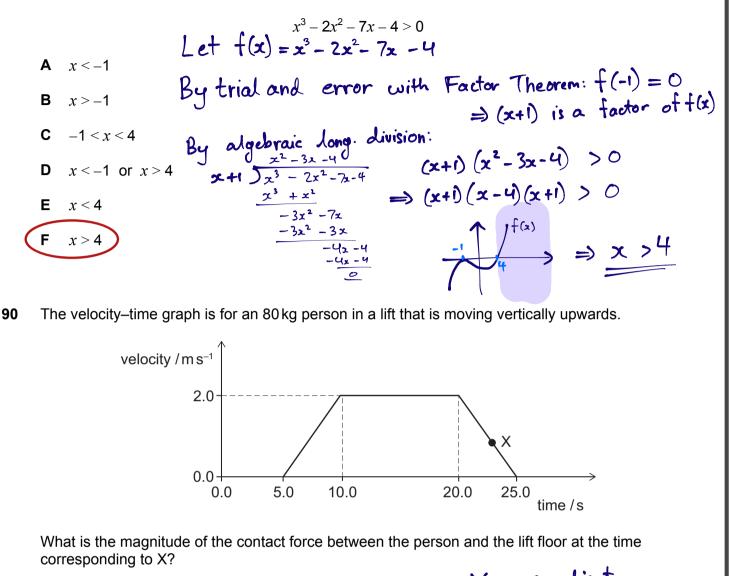
A
$$8.0 \times 10^{-5}$$
 Strain = $\mathcal{E} = \frac{\text{Stress}}{\text{Young}}$.
B 1.7×10^{-4} J
C 4.0×10^{-4} J
D 8.0×10^{-3} J
E 4.0×10^{-2} J
F 1.6×10^{-2} J
Extension = Strain \times Original Length
 $x = \frac{10^{-3}}{6} \times 2.4$
 $= 4 \times 10^{-4}$ m
Strain Energy = $\frac{1}{2}$ kx²
 $= \frac{1}{2}$ Fx = $\frac{1}{2}$ x $4 \times 10 \times 4 \times 10^{-4}$
 $= 8 \times 10^{-3}$ J
(2) (2)
(3) www.pmt.education

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 (\mathbf{c})

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89 Find the complete set of values of *x* for which



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