

GCSE Physics B (Twenty First Century Science)

J259/01 Breadth in Physics (Foundation Tier)

Question Set 23

Ali uses a hot water bottle to keep warm.



(a) He uses a kettle to heat 1.1 kg of water from 20 °C to 90 °C. Ali then pours the hot water into the hot water bottle.

The specific heat capacity of water is 4200 J/kg/°C.

Calculate the change in internal energy in heating the water.

Use the equation:

1

change in internal energy = mass × specific heat capacity × change in temperature

Give your answer to **2** significant figures.

Change in internal energy =	J	[3]	l
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(b) The kettle transfers energy electrically.

The resistance of the kettle is 20Ω .

The electric current in the kettle is 11A.

Calculate the power of the kettle.

 (c) Ali decides to use a heat pack instead of a hot water bottle.

A heat pack is a bag containing seeds, such as rice or wheat. It is heated in a microwave oven.



heat pack

Ali has two heat packs, one containing rice, and one containing wheat. He wants to investigate which heat pack will stay warm for longer.

- (i) Suggest **two** pieces of measuring apparatus he will need to use in his investigation.
- (ii) Suggest one control variable for Ali's investigation.

[1]

[2]

Total Marks for Question Set 23: 9



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