

GCSE Physics A (Gateway) J249/04 Physics A P5-P8 and P9 (Higher Tier)

Question Set 26

Multiple Choice Questions

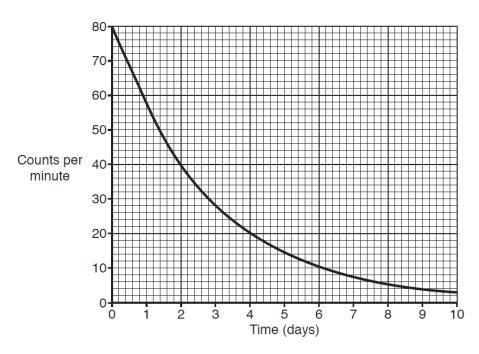
P6: Radioactivity

1	Whi	ch sta	n statement describes nuclear fusion?							
	Α	A helium nucleus joins with a hydrogen nucleus to form an alpha particle.								
	В	Two helium nuclei join to form a hydrogen nucleus.								
	С	Two hydrogen nuclei join to form a helium nucleus.								
	D	Uranium nuclei split and produce high energy neutrons causing a chain reaction.								
	You	r ansv	wer				[1]			
2	An element has more than one isotope.									
	Which row correctly describes the atoms of all isotopes of this element?									
				nbers of ctrons	Numbers of protons	Numbers of neutrons				
		Α	di	fferent	different	different				
		В	5	same	different	different				
		С	S	same	same	different				
		D	S	same	different	same				
	You	r ansv	wer				[1]			
3	Radium-226 is the most abundant isotope of radium.									
	Its n	Its nuclear mass is 226 and its nucleus contains 138 neutrons.								
	Which row is correct for another isotope of radium?									
	A nuclear mass 226; 137 neutrons									
		B nuclear mass 226; 139 neutrons								
			С	nuclear mass	s 227; 138 neutrons					
		D nuclear mass 227; 139 neutrons								
	You	r ansv	wer				[1]			

4	Radium-226, ²²⁶ Ra, decays to become radon-222, ²²² Rn.							
	What is emitted when a nucleus of radium-226 decays?							
	A A beta particle							
	B An alpha particle							
	C Four neutrons							
	D Four protons							
	Your answer	[1]						
5	A radioactive source has a half-life of 80 s.							
	How long will it take for ⅓ of the source to decay?							
	A 10 s							
	B 70 s							
	C 240 s							
	D 640 s							
	Your answer	[1]						
6	Beta radiation is used to check the thickness of thin aluminium foil at a factory.							
	Why is beta radiation used?							
	A All electromagnetic radiation is reflected by aluminium foil.							
	B Beta radiation will not pass through aluminium foil.							
	C Beta radiation will partially pass through aluminium foil.							
	D Beta radiation is reflected by aluminium foil.							
	·							
	Your answer	[1]						

7

A teacher measures the radiation from a radioactive source for 10 days.



What is the half-life of this radioactive source?

- A 1 day
- B 2 days
- C 4 days
- **D** 5 days

Your answer

[1]

8

An alpha particle collides with an atom to produce a positive ion.

What happens to the atom for it to become a positive ion?

- A It loses an electron from inside the nucleus.
- **B** It loses an electron from outside the nucleus.
- **C** It loses a neutron from inside the nucleus.
- **D** It loses a proton from outside the nucleus.

Your answer

9 The table gives some information about four radioactive isotopes. Which isotope is the best to use as a medical tracer? Half life **Radiation emitted** Α 6 hours alpha В 6 hours gamma C 6 minutes gamma D 6 years beta Your answer [1] 10 Which statement is true for isotopes of the same element? $N_{\rm p}$ = number of protons and $N_{\rm n}$ = number of neutrons. A $N_{\rm p} = N_{\rm n}$ $N_{\rm p}$ is the same but $N_{\rm n}$ is different

Your answer

 $N_{\rm p}$ is always greater than $N_{\rm n}$

The total $(N_{\rm p}$ + $N_{\rm n})$ is always the same

[1]

Total Marks for Question Set 2: 10

D

Equations in physics

 $(\text{final velocity})^2 - (\text{initial velocity})^2 = 2 \times \text{acceleration} \times \text{distance}$

change in thermal energy = mass × specific heat capacity × change in temperature

thermal energy for a change in state = mass × specific latent heat

energy transferred in stretching = $0.5 \times \text{spring constant} \times (\text{extension})^2$

potential difference across primary coil × current in primary coil = potential difference across secondary coil × current in secondary coil

Higher tier only -

force on a conductor (at right angles to a magnetic field) carrying a current = magnetic flux density × current × length



OCR is committed to seeking permission to reproduce all third-party content that it uses in its assessment materials. OCR has attempted to identify and contact all copyright holders whose work is used in this paper. To avoid the issue of disclosure of answer-related information to candidates, all copyright acknowledgements are reproduced in the OCR Copyright Acknowledgements Booklet. This is produced for each series of examinations and is freely available to download from our public website (www.ocr.org.uk) after the live examination series.

If OCR has unwittingly failed to correctly acknowledge or clear any third-party content in this assessment material, OCR will be happy to correct its mistake at the earliest possible opportunity.

For queries or further information please contact The OCR Copyright Team, The Triangle Building, Shaftesbury Road, Cambridge CB2 8EA.

OCR is part of the Cambridge Assessment Group; Cambridge Assessment is the brand name of University of Cambridge Local Examinations Syndicate (UCLES), which is itself a department of the University of Cambridge