

GCSE Physics A (Gateway) J249/04 Physics A P5-P8 and P9 (Higher Tier)

Question Set 26

Multiple Choice Questions

P6: Radioactivity

| 1 | · | ∆/hich | statement | describes | nuclear | fucion' | 2 |
|---|---------------------------------------|----------|-----------|-----------|---------|---------|---|
| | \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ | VVIIICII | Statement | describes | nuclear | IUSION | • |

- A helium nucleus joins with a hydrogen nucleus to form an alpha particle.
- **B** Two helium nuclei join to form a hydrogen nucleus.
- C Two hydrogen nuclei join to form a helium nucleus. ← fusion
- **D** Uranium nuclei split and produce high energy neutrons causing a chain reaction. \leftarrow f \sim \sim

| ., | | |
|------|--------|--|
| Your | answer | |



[1]

2 An element has more than one isotope.

Which row correctly describes the atoms of all isotopes of this element?

| | Numbers of electrons | Numbers of protons | Numbers of neutrons |
|---|----------------------|--------------------|---------------------|
| Α | different | different | different |
| В | same | different | different |
| С | same | same | different |
| D | same | different | same |

Your answer



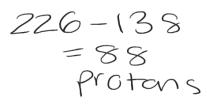
[1]

Radium-226 is the most abundant isotope of radium.

Its nuclear mass is 226 and its nucleus contains 138 neutrons.

Which row is correct for another isotope of radium?

- A nuclear mass 226; 137 neutrons ★
- B nuclear mass 226; 139 neutrons X
- C nuclear mass 227; 138 neutrons ★
- D nuclear mass 227; 139 neutrons



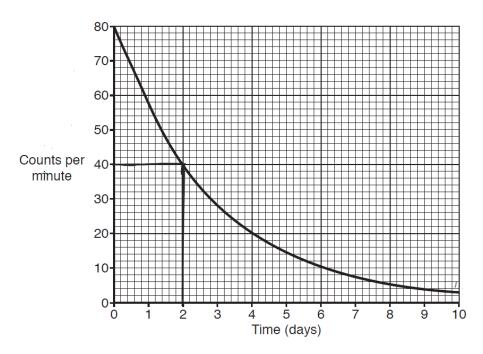
Your answer



| 4 | adium-226, ²²⁶ Ra, decays to become radon-222, ²²² Rn. | | |
|--------|---|------|--|
| | What is emitted when a nucleus of radium-226 decays? | | |
| | A A beta particle | | |
| | B An alpha particle | | |
| | C Four neutrons | | |
| | D Four protons | | |
| | Your answer B | r41 | |
| _ | let no of atoms be | 80 U | |
| 5 | A radioactive source has a half-life of 80 s. | | |
| | A radioactive source has a half-life of 80 s. How long will it take for $\frac{1}{8}$ of the source to decay? A 10 s Time no of atoms 100 LEFT. | C | |
| | A 10s Time no of atoms 100 18ft | | |
| | B 70s 80 400 | | |
| | C 240 s 160 200 240 100 | | |
| | D 640 s | | |
| | Your answer C | [1] | |
| 6 | Beta radiation is used to check the thickness of thin aluminium foil at a factory. | | |
| | Why is beta radiation used? | | |
| | A All electromagnetic radiation is reflected by aluminium foil. | | |
| | B Beta radiation will not pass through aluminium foil. | | |
| | C Beta radiation will partially pass through aluminium foil. | | |
| | D Beta radiation is reflected by aluminium foil. | | |
| | <u> </u> | | |
| | Your answer | r41 | |
| D | | [1] | |
| Dera | radiation can penetrate aluminium foil but the | | |
| TVIIVI | aluminium foil but the | | |
| mount | of penetration will vans | | |

surficiently as thickness changes.

A teacher measures the radiation from a radioactive source for 10 days.



What is the half-life of this radioactive source?

- A 1 day
- B 2 days
- C 4 days
- **D** 5 days

Your answer



8 An alpha particle collides with an atom to produce a positive ion.

What happens to the atom for it to become a positive ion?

- A It loses an electron from inside the nucleus.
- **B** It loses an electron from outside the nucleus.
- **C** It loses a neutron from inside the nucleus.
- **D** It loses a proton from outside the nucleus.

Your answer



[1]

9 The table gives some information about four radioactive isotopes.

Which isotope is the best to use as a medical tracer?

| | Half life | Radiation emitted |
|---|-----------|-------------------|
| Α | 6 hours | alpha |
| В | 6 hours | gamma |
| С | 6 minutes | gamma |
| D | 6 years | beta |

Your answer



10 Which statement is **true** for isotopes of the same element?

 $N_{\rm p}$ = number of protons and $N_{\rm n}$ = number of neutrons.

- $\mathbf{A} \qquad N_{\mathrm{p}} = N_{\mathrm{n}}$
- **B** $N_{\rm p}$ is the same but $N_{\rm n}$ is different
- **C** $N_{\rm p}$ is always greater than $N_{\rm n}$
- **D** The total $(N_p + N_n)$ is always the same

Your answer



Total Marks for Question Set 2: 10

[1]

[1]

Equations in physics

 $(\text{final velocity})^2 - (\text{initial velocity})^2 = 2 \times \text{acceleration} \times \text{distance}$

change in thermal energy = mass × specific heat capacity × change in temperature

thermal energy for a change in state = mass × specific latent heat

energy transferred in stretching = $0.5 \times \text{spring constant} \times (\text{extension})^2$

potential difference across primary coil × current in primary coil = potential difference across secondary coil × current in secondary coil

Higher tier only -

force on a conductor (at right angles to a magnetic field) carrying a current = magnetic flux density × current × length



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