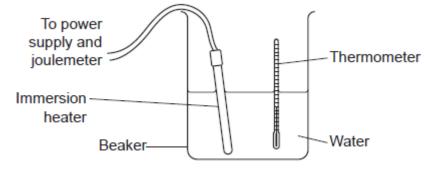


## GCSE Physics A (Gateway)

J249/01 Physics A P1-P4 and P9 (Foundation Tier)

**Question Set 3** 

A student completes an experiment to find the specific heat capacity of water.



He heats up 1kg of water, using an immersion heater. He measures the temperature rise and calculates the specific heat capacity of the water.

Attempt	Energy supplied (J)	Temperature rise (°C)	Specific heat capacity (J / kg °C)
1	10 000	2	5000
2	21000	4	5250
3	44 000	8	5500

(a) (i) Calculate the **mean** specific heat capacity.

Q=mCDT c is the specific heat capacity Q is energy supplied and DT is temperature rise since temperature did not rise as much as expected, c is higher than

actual value.

1

(ii) Write down **two** problems with this experiment **and** suggest how they could be solved.

Use the diagram and results table to help you.	
Problem 1 There is nothing covering	
the bearer	
Solution Use polystyrene cup instead of the beaker	
and use lid.	
Problem 2	
in water - thus heat lost through air	
Solution Ensure the water fevel is above the	
immersion heater [4]	J

## **Total Marks for Question Set 3: 10**



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