

## AS Level Physics A H156/01 Breadth in Physics

**Question Set 22** 

1	(a)		State <b>one</b> piece of evidence for the wave-like behaviour of electrons.	
	(b)	(i)	In an electron-gun, each electron is accelerated to a maximum kinetic energy of 210 eV. Show that the final speed of each electron is about $9 \times 10^6  \text{m}  \text{s}^{-1}$ .	[1]
		(ii)	Calculate the de Broglie wavelength $\lambda$ of each electron.	[3]
			$\lambda = m$	
	(c)		Electromagnetic waves interact with matter as photons.	[2]
			Explain the photoelectric effect using ideas of <b>photons</b> , <b>conservation of energy</b> and <b>work function</b> .	
Tot	Total Marks for Question Set 22: 10			



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