

A level Physics B

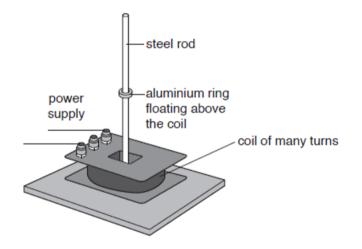
H557/01 Fundamentals of physics

Question Set 4 (Module 6 MCQs)

- Which of the following changes doubles the flux in a magnetic circuit?
 - 1 doubling the permeance
 - 2 doubling the current-turns
 - 3 halving the circuit length
 - 1, 2 and 3 are correct Α
 - В only 1 and 2 are correct
 - С only 2 and 3 are correct
 - D only 1 is correct

[1]

An aluminium ring is free to move on a steel rod. When the power supply is on, the ring floats.



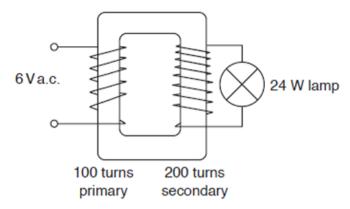
Which of the following is correct?

- An a.c. or d.c. power supply can be used. А
- в The induced current in the ring is in the same direction as the current in the coil.
- С The only purpose of the steel rod is to support the ring.
- When the ring is pushed down towards the coil more flux links it and the induced current D increases. [1]

2

1.

A 6 V a.c. supply is connected to the 100 turn primary coil of an ideal transformer. The 200 turn secondary coil runs a lamp which dissipates 24 W.

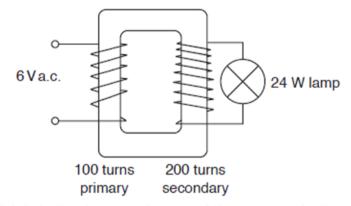


Which is the best estimate of the current in the secondary coil?

- $A \quad \frac{1}{4}A$
- **B** $\frac{1}{2}$ A
- **C** 2 A
- **D** 4 A

[1]

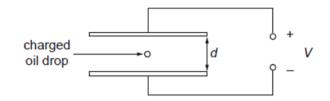
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An oil drop of mass *m* charged by one electron is balanced between two parallel horizontal metal plates. A potential difference V is applied between the plates as shown.

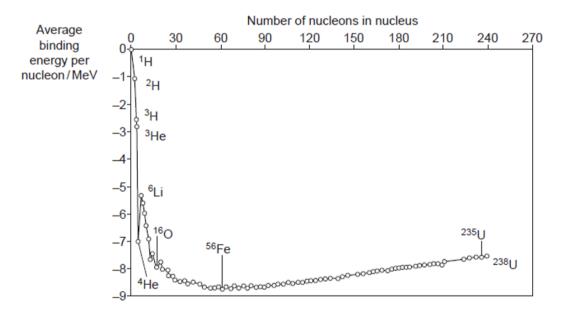


Which expression shows the balanced electrical and gravitational forces acting?

A
$$eVd = mg$$

B $\frac{eV}{d} = mg$
C $\frac{V}{ed} = mg$
D $\frac{dV}{e} = mg$

The graph shows how the binding energy per nucleon varies with the nucleon number for stable nuclei.



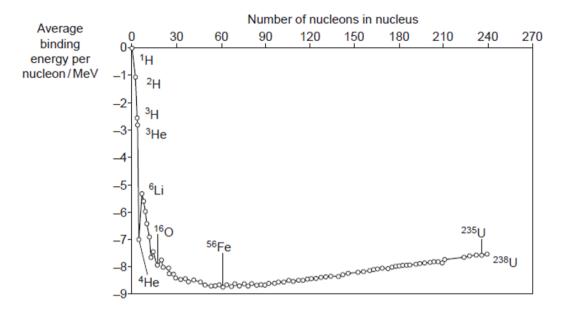
Which one of the following statements is correct?

- Α All unstable nuclei have less binding energy than stable nuclei.
- В ⁵⁶Fe requires less energy per nucleon than other stable nuclei to pull it apart into individual nucleons.
- Binding energy can be released in the fission of some heavy elements С
- D Binding energy is the energy released when a nucleus breaks down into individual nucleons.

6

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The graph shows how the binding energy per nucleon varies with the nucleon number for stable nuclei.



Which is the best estimate for the total binding energy for a nucleus of ¹⁶₈O (Oxygen)?

- **A** 10 pJ
- **Β** 20 pJ
- **C** 64 pJ
- **D** 128 pJ

[1]

Isotopes of a given element all have the same

- A proton number.
- B charge / mass ratio.
- C neutron number.
- D nucleon number.

²¹⁴₈₂Pb decays by a series of transformations to a final stable product. The particles emitted are: β , β , α , β , β , α . Which one of the isotopes below is the final product?

[1]

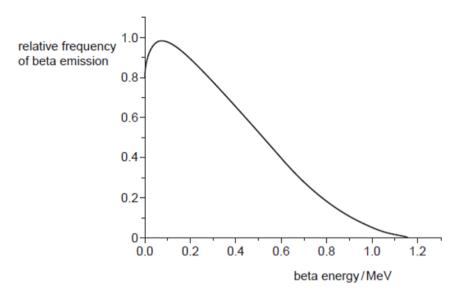
An electron is travelling at a speed of 3.1 x10⁵ m s⁻¹.

What is its kinetic energy in electronvolts?

- **A** 4.4 x 10⁻²⁰ eV
- **B** 8.8 x 10⁻⁷ eV
- **C** 0.27 eV
- **D** 500 eV

[1]

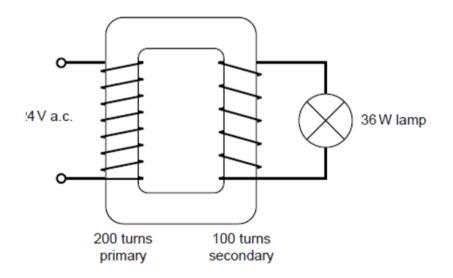
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Which statement is correct?

- A The maximum energy of beta emissions is about 0.1 MeV.
- **B** More than half the beta particles are emitted with less than half the maximum energy.
- **C** The "missing energy" in beta emissions is taken by the recoiling nucleus.
- **D** The most frequent energy of beta emission is about 1.16 MeV.

A transformer runs a 36.0W lamp from a 24.0V a.c. supply.

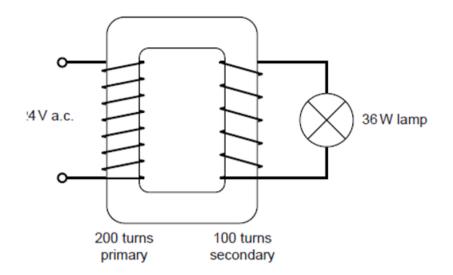


What is the current in the secondary coil?

- A 1.5A
- **B** 3.0A
- **C** 4.0A
- **D** 6.0A

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[1]

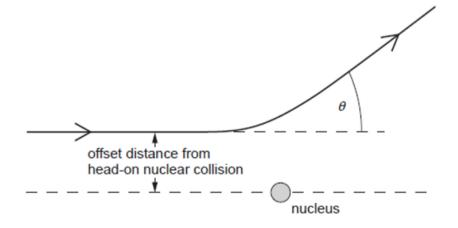


What is the best estimate for the rate of change of flux in the transformer core?

- A 0.12Wbs⁻¹
- B 0.12Tm⁻²s⁻¹
- C 12Wbs⁻¹
- D 12Tm⁻²s⁻¹

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An α particle approaches an atomic nucleus and is scattered through angle θ . The path of the particle is offset as shown.



Which change on its own would cause the scattering angle θ to increase?

- **A** Use an α particle with higher kinetic energy.
- **B** Have a smaller offset distance for the initial trajectory.
- **C** Use a target nucleus with a smaller charge.
- D Use a target nucleus with a smaller mass.

A particle of charge *q* and mass *m* travels in a vacuum in a region of constant magnetic flux density *B*. It moves at a constant speed *v* in a circle of radius *r*.

Which expression gives the time for the particle to complete one circle of its r

A
$$\frac{2\pi m}{Bqr}$$

B $\frac{2\pi mr}{Bq}$
C $\frac{2\pi m}{Bq}$
D $\frac{Bq}{2\pi m}$
[1]

A particle of charge q and mass m travels in a vacuum in a region of constant magnetic flux density B.

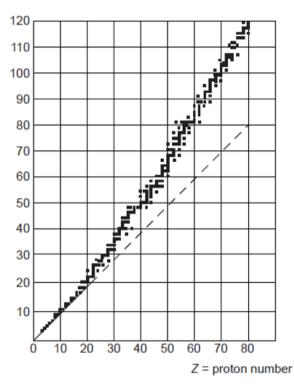
It moves at a constant speed v in a circle of radius r.

15

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A proton and an α particle, with the same velocity, enter the region of constant magnetic flux density.

What is the ratio of the radii of their paths, $\frac{r_{\alpha}}{r_{proton}}$? A $\frac{1}{4}$ B $\frac{1}{2}$ C 2 D 4



N = neutron number

Which statement about stable nuclei is correct?

Which statement about stable nuclei is correct?

- A For elements up to about Z = 50 $N \approx Z$.
- **B** Elements with $Z \approx 80$ have more protons than neutrons.
- **C** Greater $\frac{N}{Z}$ ratio is needed to hold larger nuclei together, because only neutrons take part in the strong nuclear force of attraction to balance electrostatic repulsion.
- **D** Greater $\frac{N}{Z}$ ratio is needed to hold larger nuclei together because only nearest neighbour nucleons take part in the strong nuclear force of attraction to balance electrostatic repulsion. [1]

The electric potential at a distance r from a positive point charge Q is 450V. The potential increases to 500V when the distance from the point charge decreases by 1.5m.

What is the value of r?

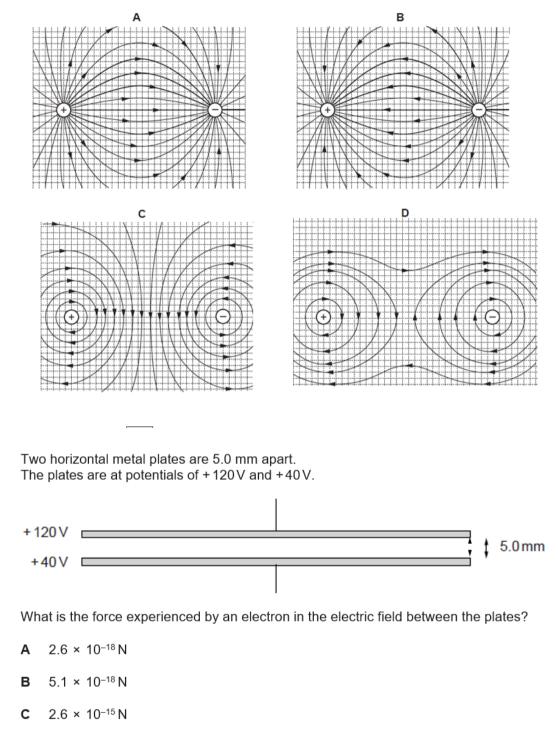
A 1.5m

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- **B** 7.9 m
- C 15m
- **D** 29 m

A positive and a negative charge of equal magnitude are placed near each other.

Which diagram best represents the electric field of the two charges?



D 5.1 × 10⁻¹⁵ N

[1]

[1]

Protons consist of quarks.

The 'up' anti-quark has a charge of $-\frac{2e}{3}$ and the 'down' anti-quark has a charge of $+\frac{1e}{3}$, where *e* is the charge on an electron.

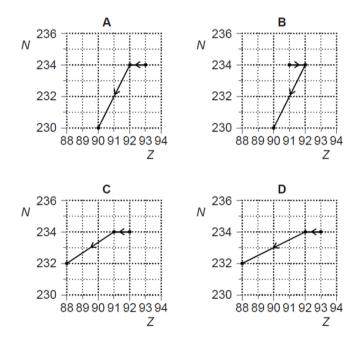
What does an anti-proton contain?

	Up anti-quarks	Down anti-quarks
Α	0	3
В	1	1
С	1	2
D	2	1

[1]

A radioactive nucleus is formed by β -decay. This nucleus then decays by α -emission.

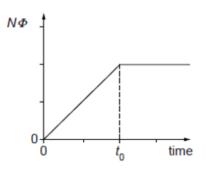
Which graph of nucleon number *N* plotted against proton number *Z* shows the β -decay followed by the α -emission?



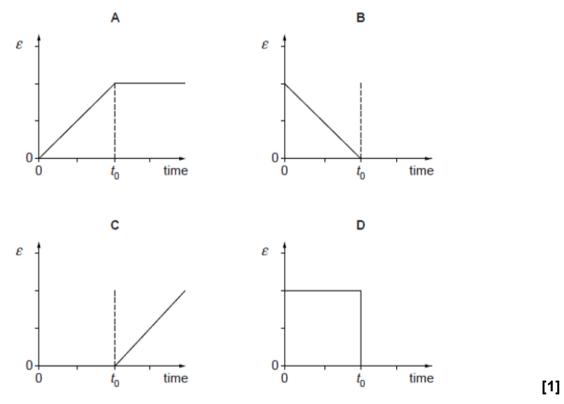
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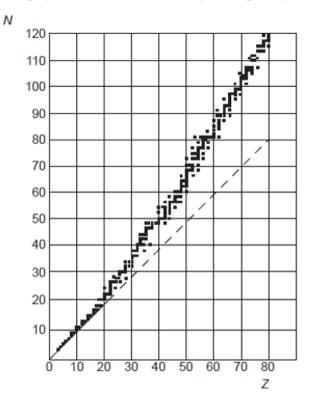
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The graph shows how the flux linkage $N\Phi$ through a coil changes with time when the coil is moved into a magnetic field.



Which of the following graphs shows the magnitude of the induced e.m.f. $\pmb{\epsilon}$ in the coil over the same time period?





Which statements about stable nuclei are correct?

- 1 Nuclei of elements with Z > 20 have more protons than neutrons.
- 2 For the nuclei of lighter elements $N \approx Z$.
- 3 Greater $\frac{N}{Z}$ ratio is needed to hold larger nuclei together, because only nearest neighbour nucleons take part in the strong nuclear force of attraction to balance electrostatic repulsion.
- A 1, 2 and 3 are correct
- B only 1 and 2 are correct
- C only 2 and 3 are correct
- D only 1 is correct

[1]

Total Marks for Question Set 4: 24



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