

A Level Physics A

H556/02 Exploring physics

Question Set 22

The nuclear reaction below shows how the isotope of fluorine-18 $\binom{18}{9}$ F) is made from the isotope of oxygen-18 $\binom{18}{8}$ O).

$$^{18}_{8}O + ^{1}_{1}p \rightarrow ^{18}_{9}F + ^{1}_{0}n + \gamma$$

The oxygen-18 nucleus is **stationary** and the proton has kinetic energy of $0.25 \times 10^{-11} \, \text{J}$. The binding energy of the $^{18}_{8}\text{O}$ nucleus is $2.24 \times 10^{-11} \, \text{J}$ and the binding energy of the $^{18}_{9}\text{F}$ nucleus is $2.20 \times 10^{-11} \, \text{J}$. The proton and the neutron have zero binding energy.

(i) Explain why a high-speed proton is necessary to trigger the nuclear reaction shown above.

The proton is repelled by the nucleus, but a high speed one ranger close. [2]

(ii) Estimate the minimum wavelength λ of the gamma ray photon (γ).

(iii) Fluorine-18 is a positron emitter.

Name a medical imaging technique that uses fluorine-18 and state one benefit of the technique.

(b) Describe how the components of a computerised axial tomography (CAT) scanner can produce high-quality images of the internal structures of a patient.

Total Marks for Question Set 22: 11



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