

## A Level Physics A

H556/01 Modelling physics

**Question Set 10** 

- 1 (a) Define the internal energy of a substance. The Sum of the randymy distributed kinetic and pression energies associated with [1] the atoms or millicules which make up the Substance
  - (b) A block of paraffin wax is melting at a constant temperature of 52 °C. Use the behaviour of paraffin molecules to describe and explain the changes to the internal energy of the molecules of the paraffin wax as it melts.

[4]

- The KE will not change us it melts, since me temperature remains constant.
- However, the PE of the molecules will increase as it melts

## **Total Marks for Question Set 10: 5**

\_ Therefore, the internal energy will increase



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