

A Level Physics A H556/01 Modelling physics

Question Set 9

 1 (a) Fig. 21.1 shows some of the energy levels of electrons in hydrogen gas atoms. The energy levels are labelled A, B, C and D.



Fig. 21.1 (not to scale)

- (i) Explain why the energy levels are negative. [1] Electrons are bound to the nucleus, so require energy to leave the nucleus.
- (ii) An electron makes a transition (jump) from level C to level A.

2 Calculate the wavelength in nm of the photon absorbed by this electron.

Light from a distant galaxy is passed through a diffraction grating. Fig. 21.2 shows the part of the spectrum of light that shows a strong hydrogen-alpha emission line.



Fig. 21.2

- (i) State how an emission line is produced. [1]
 Electrons drop Le a lower energy level and emit EM radiation.
 (ii) State on adjustment that applied to the experimental emergence that
- (ii) State an adjustment that could be made to the experimental arrangement that would space the emission lines more widely.
 (ألى رف ورسيل من عليه من المناطقة (1)
- (iii) In the laboratory, the wavelength of the hydrogen-alpha emission line is 656.3 nm. Use Fig. 21.2 to determine the recession velocity of the galaxy.

[1]

(iv) Suggest why hydrogen spectral lines play an important role in determining red shift of galaxies.

Hydrogen is very abundant in stors.

- (c) Light from a similar star is viewed in a galaxy further away. The star is part of a pair of stars which orbit a common centre of mass.
 Describe and explain how the equivalent spectrum might appear. [3]
 - Less intende as the light has traductiled former
 - Als it is further away, we would expect it to have a larger recessional relacity. There will there for be more reasoning and the warding the will be largen
 - These could be a periodic shipt in X due to the orbit.

Total Marks for Question Set 9: 14

(b)



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