



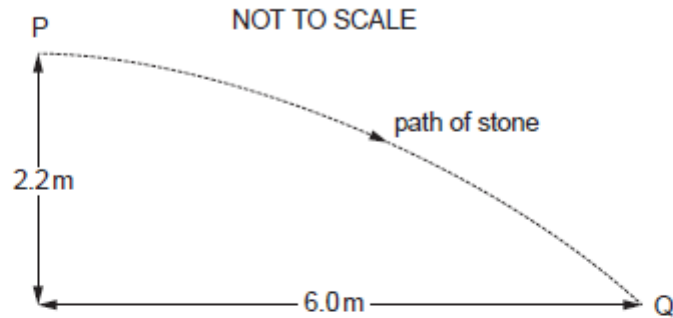
## **GCE PHYSICS**

S21-A420QS

### **Assessment Resource number 4**

### **Newtonian Physics Resource D**

1. (a) In an investigation of projectile motion, a student throws a stone. It is moving horizontally when it leaves his hand (at point P). It reaches the ground at point Q.



- (i) By analysing a video of the stone's flight, its horizontal velocity component,  $v_h$ , is found to be almost constant. Discuss whether or not this is to be expected. [2]

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- (ii) The approximate value of  $v_h$  obtained from the video was  $9.0 \text{ ms}^{-1}$ . Determine whether this value is consistent with the measured distances recorded in the diagram. Show your reasoning clearly. [3]

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- (b) Calculate the magnitude and direction of the stone's velocity just before it hits the ground. [4]

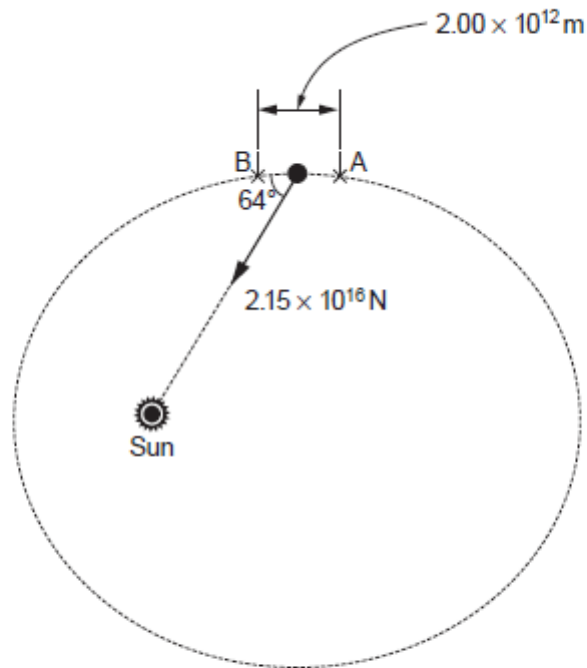
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2. The diagram shows the dwarf planet, Eris, at one point in its orbit.



(a) Explain why the *moment* (about the centre of the Sun) of the Sun's force on Eris is zero. [1]

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(b) Calculate the *work* done by the Sun's gravitational force on Eris as Eris moves from A to B. The mean values of the force and the angle at which it acts are shown on the diagram. [2]

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(c) Showing your reasoning clearly, determine whether your answer to (b) is consistent with these data:

Mass of Eris =  $1.66 \times 10^{22} \text{ kg}$

Speed of Eris at A =  $3460 \text{ m s}^{-1}$

Speed of Eris at B =  $3770 \text{ m s}^{-1}$

[3]

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3. (a) State the principle of conservation of momentum. [2]

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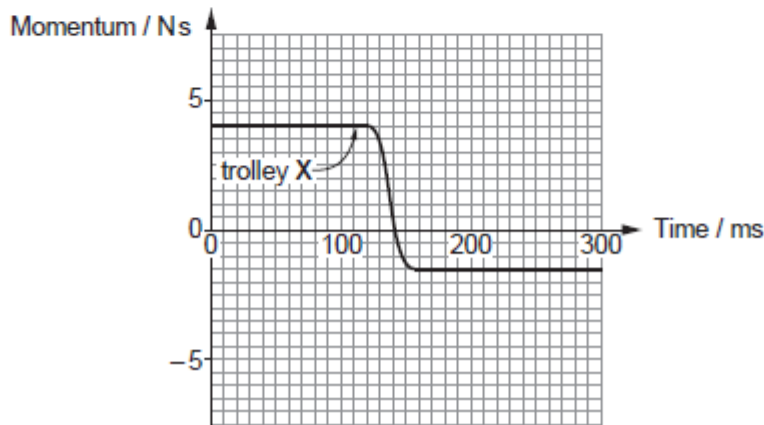
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- (b) A trolley, X, travels towards a stationary trolley, Y. See diagram.



The trolleys collide head-on. A momentum-time graph is given for trolley X.



- (i) Trolley Y has a mass of 2.4 kg. Determine its velocity after the collision. [3]

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- (ii) Using the same graph grid (opposite) carefully sketch a graph of Y's momentum between 0 and 300ms. [3]

- (iii) Use the momentum-time graph for X to estimate the mean force on X during the collision. [2]

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- (a) Vadim uses a ruler to measure the sides of a copper block. He records the measurements as:

length =  $50 \pm 1$  mm,      breadth =  $42 \pm 1$  mm,      height =  $36 \pm 1$  mm.

Using an electronic balance he measures the mass of the block as  $670.85 \pm 0.01$  g.

Use Vadim's data to answer the following.

- (i) Determine a value for the density of copper in  $\text{kg m}^{-3}$  and the **absolute uncertainty** in this value. [4]

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- (ii) Determine the number of atoms per  $\text{m}^3$  of copper. The uncertainty is **not** required. The atomic mass of copper is 63.5u. [2]

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- (b) (i) I. Calculate the number of molecules per  $\text{m}^3$  for a gas (assumed to be ideal) at a temperature of  $15^\circ\text{C}$  and a pressure of  $101\text{ kPa}$ . [3]

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- II. When asked why there are far fewer gas molecules per  $\text{m}^3$  than atoms per  $\text{m}^3$  in the copper block, a student replies, "Each molecule of the gas takes up much more space." Discuss whether or not he is right. [2]

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- (ii) I. Two gases have molecular masses  $m_{(1)}$  and  $m_{(2)}$ . Show clearly that when the gases are at the same temperature, the ratio of the rms speeds of their molecules is: [2]

$$\frac{c_{\text{rms}(1)}}{c_{\text{rms}(2)}} = \sqrt{\frac{m_{(2)}}{m_{(1)}}}$$

- II. Calculate the percentage difference in the rms speeds of nitrogen and oxygen molecules in the same sample of air. Take the percentage difference to be defined as:

$$\frac{\text{rms speed for nitrogen} - \text{rms speed for oxygen}}{\text{rms speed for oxygen}} \times 100\%$$

[Molecular mass for nitrogen =  $28.0\text{ u}$ . Molecular mass for oxygen =  $32.0\text{ u}$ .] [2]

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