

GCSE Chemistry B (Twenty First Century Science)

J258/03 Breadth in chemistry (Higher Tier)

Question Set 29

		Amir is given a solution of potassium sulfate, which can be used as a fertiliser.	
(a)		He carries out a flame test on the solution.	
	(i)	Describe how Amir does the flame test.	[3]
	(ii)	What colour does Amir see when he carries out the flame test on potassium sulfate?	[4]
			[1]
(b)		Write a balanced symbol equation for the reaction of potassium sulfate (K2SO4) solution with barium chloride (BaC <i>l</i> 2) solution.	
		Show state symbols.	[3]
(c)		Amir has a sample of another fertiliser, Fertiliser E .	
		He looks at the emission spectra of Fertiliser E and potassium sulfate:	
		Fertiliser E	
		Potassium sulfate	
		400 Wavelength (nm) 700	
	(i)	What can Amir conclude about Fertiliser E from these emission spectra?	[1]
	(ii)	Compounds can be analysed using a flame test or an emission spectrum.	
		State one advantage of using an emission spectrum to analyse compounds.	[1]

Total Marks for Question Set 29: 9

1



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