

GCSE Chemistry B (Twenty First Century Science)

J258/03 Breadth in chemistry (Higher Tier)

Question Set 7

This is an equation for a reaction that occurs in a lightning flash.

N2 + O2 **⇒**2NO

1

Very high temperatures are needed.

(a)	(i)	Explain how you can tell that this equation refers to an equilibrium.	[1]
	(ii)	Use ideas about rates to explain what is happening when the reaction reaches dynamic equilibrium.	[2]
(b)		Scientists can use this reaction to make nitrogen compounds from gases in the air.	
	(i)	Suggest a use for these compounds.	[1]
	(ii)	The scientists discuss increasing the pressure on the reaction.	
		Describe and explain the effect on the equilibrium position.	[2]
(c)		There are several ways of making nitrogen compounds from nitrogen gas in industry.	
		Give two reasons why scientists may choose this reaction and one against.	
			[3]

Total Marks for Question Set 7:9



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