

## GCSE Chemistry B (Twenty First Century Science) J258/02 Depth in chemistry (Foundation Tier)

**Question Set 25** 

1 Crude oil is a mixture of mainly alkanes.

All alkanes have this general formula:

## CnH2n+2

- (a) Explain how this formula shows that alkanes are hydrocarbons. [2]
- (b) Crude oil also contains a small amount of other molecules that are **not** alkanes. The table shows the formulae of three molecules in crude oil.

Formula of molecule
C <sub>8</sub> H <sub>18</sub>
C <sub>3</sub> H <sub>8</sub> S
C <sub>6</sub> H <sub>12</sub>

Put a ring around the correct answer to show whether each molecule **is** or **is not** an alkane.

Explain your answers.

Use the general formula for alkanes to help you.

C8H18 is / is not an alkane because....

C<sub>3</sub>H<sub>8</sub>S **is / is not** an alkane because ...

C6H12 is / is not an alkane because .....

[3]

## **Total Marks for Question Set 25: 5**



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