

**GCSE Chemistry B (Twenty First Century Science)**  
**J258/01 Breadth in Chemistry (Foundation Tier)**

**Question Set 6**

1

A factory electrolyses sodium chloride solution to make useful products.

(a) The electrolysis produces chlorine.

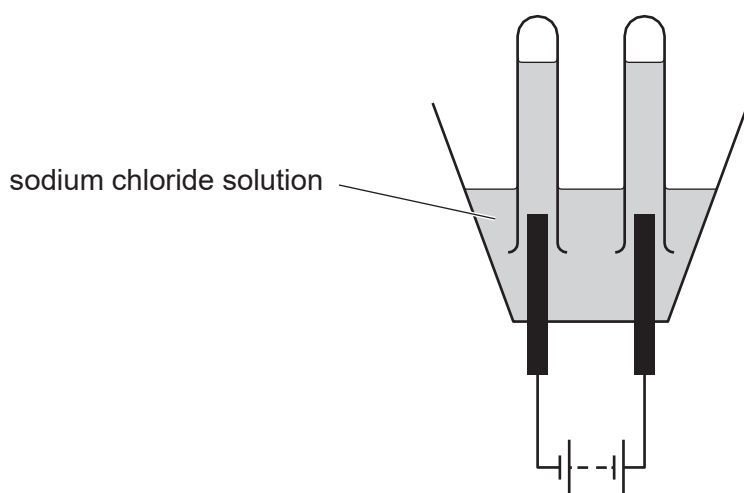
Complete the sentence by putting a **ring** around one word in each pair.

The **chlorine / chloride** ions are attracted to the **positive / negative** electrode, where they lose **electrons / protons**.

[3]

(b) Nina electrolyses a solution of sodium chloride.

She uses this apparatus.



Nina thinks that the experiment makes chlorine gas at one electrode.

She is not sure if the gas at the other electrode is hydrogen or oxygen.

Describe the tests Nina can do to identify chlorine, hydrogen and oxygen **and** the results she should expect.

[3]

**Total Marks for Question Set 6: 6**

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