

GCSE Chemistry A (Gateway Science) J248/04 Chemistry A C4-C6 and C7 (Higher Tier)

Question Set 13

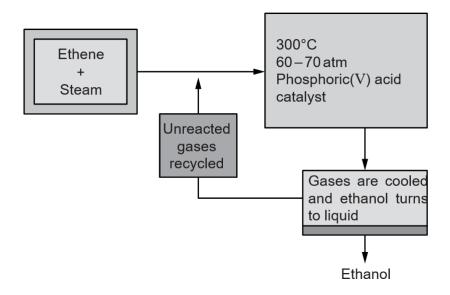
 1^* Ethanol is manufactured by reacting ethene, C_2H_4 , with steam.

The reaction is reversible and occurs in a closed system.

$$C_2H_4(g) + H_2O(g) = C_2H_5OH(g)$$
 $\Delta H = -45kJ \text{ mol}^{-1}$

Only 5% of the ethene is converted into ethanol at each pass through the reactor.

By removing the ethanol from the equilibrium mixture and recycling the ethene, it is possible to achieve an overall 95% conversion.



Explain why the conditions use for the reaction are chosen.

Total Marks for Question Set 14: 6

[6]

Resource Materials

0

(9 7 N N 14.00 114.00 114.00 115. (2) 4 5 B B boron 10.8 13 A 1 13 A 1 13 A 1 2 27.0 31 B Ga gallum 69.7 49 In In Indiam Indiam 1114.8 81 T 1 T 1 1 14.8 E 10.4 204.4 204.4 3 The Periodic Table of the Elements 30 Zn Zn Zn Znc 65.4 48 Cd Cd Cd Hg Hg Hg Hg Hg Conc.up 112.4 Conc.up 112.4 Conc.up Conc.up 112.4 Co 29 Cu copper 63.5 47 Ag silver 1107.9 79 T9 T9 T111 T111 Rg 9 27 27 Co cobalt 58.9 45 Rh rhodium 102.9 1r infetum 192.2 109 MR MR rhodium 192.2 109 MR MR methrerium methrerium 25 Mn nanganese 54.9 43 Tc 75 Re thenium 186.2 107 Bh bohrium Key atomic number Symbol name relative atomic mass 21 Sc Scandium Scandium 45.0 39 Y Y yttrium 88.9 89-103 (5)

2 He hellum hellum hellum 4.0 10 10 Ne neoral 20.2 20.2 18 Ar argon 39.9 36 Xr krypton 83.8 54 Xr krypton 83.8 86 Rn radon rad



OCR is committed to seeking permission to reproduce all third-party content that it uses in its assessment materials. OCR has attempted to identify and contact all copyright holders whose work is used in this paper. To avoid the issue of disclosure of answer-related information to candidates, all copyright acknowledgements are reproduced in the OCR Copyright Acknowledgements Booklet. This is produced for each series of examinations and is freely available to download from our public website (www.ocr.org.uk) after the live examination series.

If OCR has unwittingly failed to correctly acknowledge or clear any third-party content in this assessment material, OCR will be happy to correct its mistake at the earliest possible opportunity.

For queries or further information please contact The OCR Copyright Team, The Triangle Building, Shaftesbury Road, Cambridge CB2 8EA.

OCR is part of the Cambridge Assessment Group; Cambridge Assessment is the brand name of University of Cambridge Local Examinations Syndicate (UCLES), which is itself a department of the University of Cambridge