

GCSE Chemistry A (Gateway Science) J248/03 C1-C3 and C7 Higher (Higher Tier)

Question Set 27

- 1 Zinc nitrate can be made by reacting zinc oxide with nitric acid, HNO₃.
 - (a) Write a **balanced symbol** equation for this reaction. [2] (b) A student suggests this method for preparing zinc nitrate. 1. Measure 50 cm^3 of dilute nitric acid into a beaker. 2. Add one spatula measure of zinc oxide. 3. Heat the mixture until crystals of zinc nitrate made. Her method will **not** make a pure dry sample of zinc nitrate. What improvements should she make to the method to make sure that: • the reaction is complete • the zinc nitrate can be separated from the nitric acid and the zinc oxide? Explain your answer. [4] (c) Describe why this reaction is a neutralisation reaction. [2]

Total Marks for Question Set 27: 8

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The Periodic Table of the Elements



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