

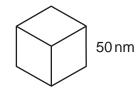
GCSE Chemistry A (Gateway Science) J248/03 C1-C3 and C7 Higher (Higher Tier)

Question Set 5

1 A new sun cream has been developed using zinc oxide nanoparticles.

The small particles provide better protection from the sun and they do not leave white marks on the skin.

- (a) Explain **one** possible risk of using nanoparticles in sun cream. [1]
- **(b)** A cube-shaped nanoparticle has sides of length 50 nm.



Calculate the surface area to volume ratio for this nanoparticle.

Use the equation: ratio = surface area ÷ volume

(c) (i) Scientists compare the size of nanoparticles to the sizes of other small objects.

Look at the table.

| Object | Diameter (nm) | | | | | | | |
|-------------------------|---------------|--|--|--|--|--|--|--|
| Gold atom | 0.14 | | | | | | | |
| Water molecule | 0.27 | | | | | | | |
| DNA strand | 2.5 | | | | | | | |
| Zinc oxide nanoparticle | 32 | | | | | | | |
| Red blood cell | 7000 | | | | | | | |
| Human hair | 100 000 | | | | | | | |

The diameter of a DNA strand is 2.5 nm.

Explain why DNA is a nanoparticle but a water molecule is **not** a nanoparticle.

| (ii) | · | | | | | | | | | | | |
|------------|--|-----|--|--|--|--|--|--|--|--|--|--|
| | Give your answer to 2 significant figures. | | | | | | | | | | | |
| | Number of nanoparticles = | [2] | | | | | | | | | | |
| Total Mark | | | | | | | | | | | | |

The Periodic Table of the Elements

| 1 | 0) | 18 | 2 Hp | helium 4.0 | 10 | Ne | neon 20.2 | 18 | Ar | argon 39.9 | 36 | 궃 | krypton 83.8 | 54 | Xe | xenon 131.3 | 98 | 몺 | radon | | | |
|--|-----|-------------------|------------|-----------------|----|----|------------------|------------------|-------------------|--------------------|-----------------|--------------------|-------------------|-------------------|-----------------|--------------------|---------|-------------|-------------------|-----|--------|---------------|
| Car A | (7) | • | | 17 | 6 | щ | fluorine 19.0 | 17 | 23 | chlorine 35.5 | 35 | Ŗ | bromine 79.9 | 53 | - | lodine 126.9 | 85 | At | astatine | | | |
| Carry Carr | (9) | | | 16 | œ | 0 | oxygen 16.0 | 16 | တ | sulfur 32.1 | 34 | Se | selenium 79.0 | 52 | Te | tellurium 127.6 | 84 | Po | polonium | 116 | ۲ | livermorium |
| Canada C | (2) | | | 15 | 7 | z | nitrogen 14.0 | 15 | ۵ | phosphorus 31.0 | 33 | As | arsenic 74.9 | 51 | Sb | antimony 121.8 | 83 | ö | bismuth 209.0 | | | |
| C 2 | (4) | | | 14 | 9 | ပ | carbon 12.0 | 14 | S | slicon 28.1 | 32 | Ĝ | germanium 72.6 | 20 | Sn | tin 118.7 | 82 | Рь | le ad 207.2 | 114 | F1 | flerovium |
| C 2 Fe A Fe Fe A Fe A Fe A A A A A A A A A | (3) | | | 13 | 2 | ω | boron 10.8 | 13 | Αl | aluminium 27.0 | 31 | Ga | gallium 69.7 | 49 | 딥 | indium 114.8 | 81 | 11 | thallium 204.4 | | | |
| Canonic number Symbol Sy | | | | 12 | 30 | Zn | zinc 65.4 | 48 | ర | cadmium 112.4 | 80 | Нg | mercury 200.6 | 112 | ၁ | copernicium | | | | | | |
| Carry Symbol Sy | | | | | | 7 | 29 | ರ | ооррег 63.5 | 47 | Ag | siliver 107.9 | 6/ | Αn | gold 197.0 | 111 | Rg | roentgenium | | | | |
| Ca | | | | 10 | 28 | Z | nickel 58.7 | 46 | Pd | palladium 106.4 | 28 | ቷ | platinum 195.1 | 110 | Ds | darmstadfium | | | | | | |
| Key Symbol Felative atomic mass Symbol Symbol Symbol Felative atomic mass Symbol | | | | 6 | 27 | ပိ | oobalt 58.9 | 45 | R | modium 102.9 | 22 | = | iridium 192.2 | 109 | Ĭ | meitnerium | | | | | | |
| Key atomic number Symbol Felative atomic mass 4 | | | | | | œ | 26 | Fe | lron 55.8 | 44 | Ru | ruthenium 101.1 | 9/ | Os | 08mium 190.2 | 108 | Нs | hassium | | | | |
| Key atomic numb Symbol Symbol Symbol Symbol 4 | | | | | | 7 | 25 | Mn | manganese 54.9 | 43 | ည | technetium | 22 | Re | menium 186.2 | 107 | 뮵 | bohrium | | | | |
| 2 | | | Jer. | mass | | | | | | 9 | 24 | ပ် | chromium 52.0 | 42 | Mo | molybdenum 95.9 | 74 | > | ungsten 183.8 | 106 | Sg | seaborgium |
| 2 | | name /e atomic | | | 2 | 23 | > | vanadium 50.9 | 41 | qN | niobium 92.9 | 73 | Та | tantalum 180.9 | 105 | Op | dubnium | | | | | |
| 2 4 4 4 Be beryllium 9.0 112 Mgg magnesium 24.3 20 Ca calcium 40.1 38 Sr strontium 87.6 Ba bartum 137.3 88 Ra radum | | | ato | relativ | | | | | | 4 | 22 | F | ttanium 47.9 | 40 | Zr | arconium 91.2 | 72 | Ξ | hafinium 178.5 | 104 | ₹ | rufherfordium |
| | ' | | | | | | | | | က | 21 | သွ | scandium 45.0 | 39 | > | yttrium 88.9 | 1 | 57-71 | lanthanoids | 3 | 89-103 | actinol ds |
| (1) 1 H hydrogen 1.0 3 Li lithium 6.9 11 Na sodium 23.0 19 K K P Potassium 39.1 37 R B D R B S S S S S S S S S S S S S S S S S S | (2) | _ | | 2 | 4 | Be | beryllium 9.0 | 12 | Mg | magnesium 24.3 | 20 | Sa | calcium 40.1 | 38 | S | strontium 87.6 | 26 | Ba | 137.3 | 88 | Ra | radium |
| | Ð | - | ← I | hydrogen 1.0 | က | : | lithium 6.9 | 11 | Na | sodium 23.0 | 19 | ¥ | potassium 39.1 | 37 | Rb | rubidium 85.5 | 22 | S | caesium 132.9 | 87 | ቷ | francium |



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