

GCSE Chemistry A (Gateway Science)

J248/02 C4-C6 and C7 Foundation (Foundation Tier)

Question Set 24

1 This question is about hydrocarbons.

Name of alkane	Molecular formula	Structure
Methane	CH ₄	Н Н—С—Н Н
Ethane	C ₂ H ₆	н н н—с—с—н н н
Butane	C ₄ H ₁₀	Н Н Н Н - С-С-С-С-Н Н Н Н Н

The table shows some information about alkanes.

- (a) Complete the table.
- (b) Ethane is a **saturated hydrocarbon**.

Explain why ethane is called both a hydrocarbon and saturated.

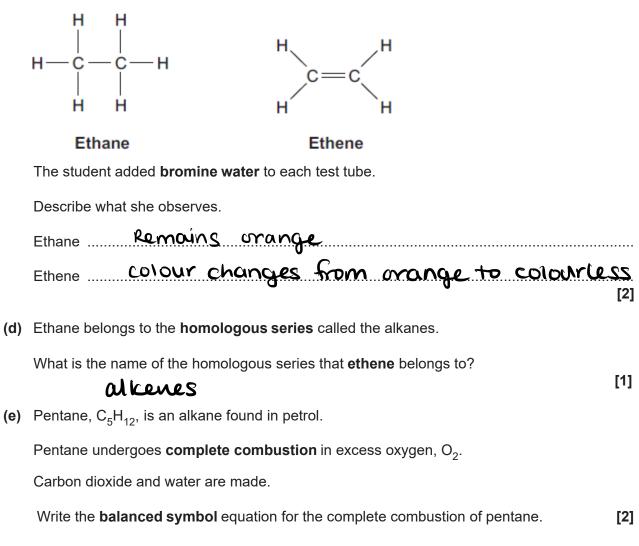
Because it is only composed of hydrogen and carbon. - thus called hydrocarbon

It is also saturated because no carbon form double bond / ethane contains only single bonds between carbon atoms.

[3]

[2]

(c) A student has two test tubes. One contains **ethane** and one contains **ethene**.



 $C_{S}H_{12} + 80_{2} \rightarrow SCO_{2} + 6H_{2}O$

Total Marks for Question Set 24: 10



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