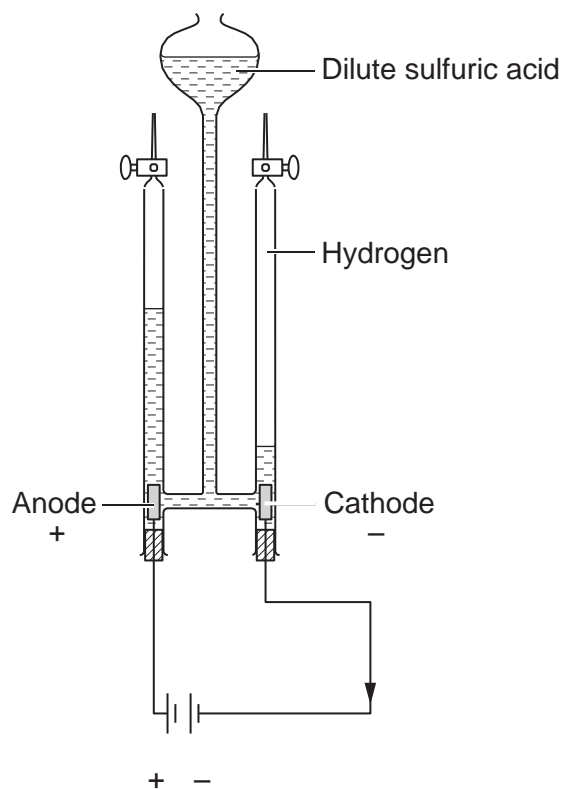


GCSE Chemistry A (Gateway Science)
J248/01 Chemistry A C1-C3 and C7 (Foundation Tier)

Question Set 18

1 A student electrolyses dilute sulfuric acid.



Hydrogen gas is made at the cathode.

The student measures the volume of hydrogen made at the cathode every 2 minutes for 10 minutes.

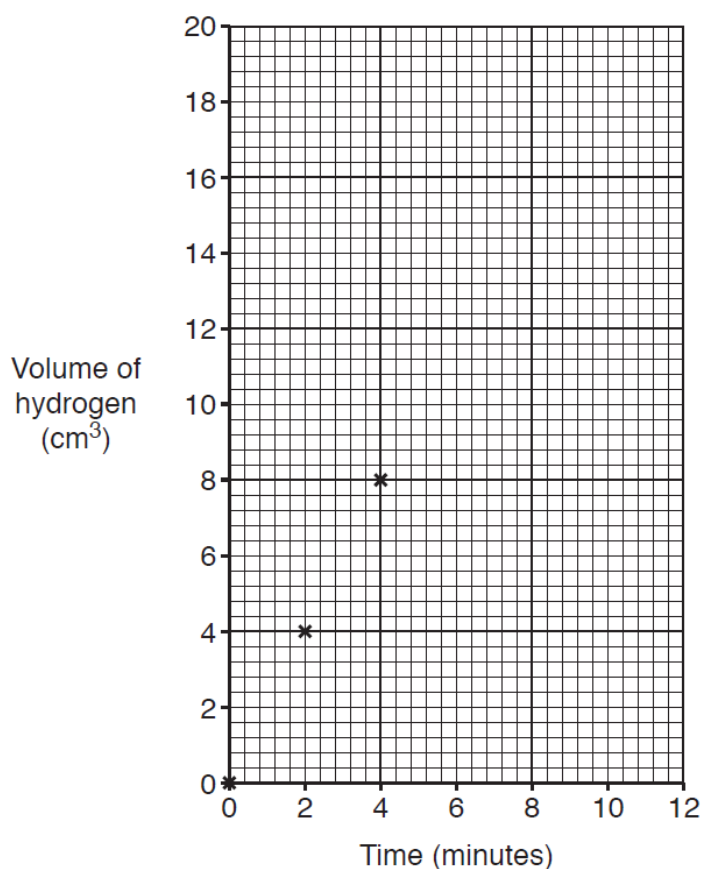
Look at his results.

Time (minutes)	Volume of hydrogen (cm ³)
0	0.0
2	4.0
4	8.0
6	14.0
8	16.0
10	20.0

(a) Plot the results on the grid. The first 3 points have been done for you.

Draw a line of best fit.

[2]



(b) One of the results is **anomalous**.

Circle the anomalous result on the graph.

[1]

(c) Sulfuric acid contains these particles.



Which particles are attracted to the **anode**?

[1]

(d) The student also investigates the electrolysis of some molten (liquid) salts.

Complete the table.

Molten salt	Formula	Product at cathode	Product at anode
Potassium chloride	KCl	Potassium
Lead iodide	PbI_2	Iodine

[2]

Total Marks for Question Set 18: 6



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