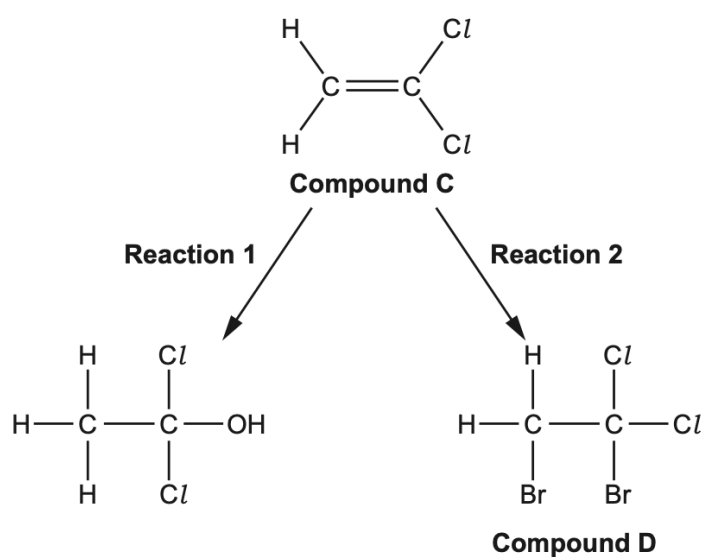


AS level Chemistry A

H032/02 Depth in chemistry

Question Set 12

1. Two reactions of compound **C** are shown in the flowchart below.



- (a) State the reagents and conditions for **reaction 1**. [1]
- (b) In **reaction 2**, compound **C** reacts with bromine to form compound **D**. [1]
- (i) Give the systematic name of compound **D**. [1]
- (ii) Outline the mechanism for **reaction 2**. [3]
- Include curly arrows, charges and relevant dipoles.
- (c) Compound **C** forms an addition polymer **E**. [3]
- (i) Write a balanced equation for this reaction. [2]
- Show displayed formulae.
- (ii) State **one** advantage and **one** disadvantage of using combustion as a method for the disposal of waste polymer **E**. [2]

Total Marks for Question Set 6: 9

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