

A Level Chemistry B (Salters) H433/02 Scientific literacy in chemistry

Question Set 11

1 Twaron[™] is a polymer used to make body armour.

The polymer strands are made using the reaction in Fig. 1.1.

$$C_1$$
 + H_2N - NH_2 - polymer strand of Twaron

Fig. 1.1

(a) Name the functional groups in compounds A and B.

A.....

B.....[2]

(b) Suggest the O=C-Cl bond angle in compound **A**.

Explain your answer. [3]

(c) Compound A can be made by the reaction in Fig. 1.2.

Fig. 1.2

Calculate the mass of compound **A** that can be made from 32 g of benzene-1,4-dicarboxylic acid if the yield is 67%.

Give your answer to the nearest whole number.

mass of compound
$$A = g$$
 [2]

(d) A synthetic route for making compound **B** is shown in **Fig. 1.3**.

Fig. 1.3

Use your chemical knowledge and the Data Sheet to suggest possible reagents for **steps** 1 and 2.

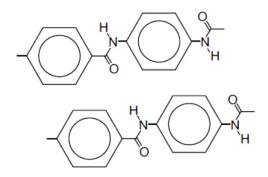
Step 1.....

Step 2

(e) (i) The polymer strands link together by intermolecular bonds when they are spun to form Twaron. This gives the fabric its tough quality.

Name the strongest intermolecular bonds that can form between the chains.

ii) Mark the positions of the intermolecular bonds by drawing dotted lines on the diagram

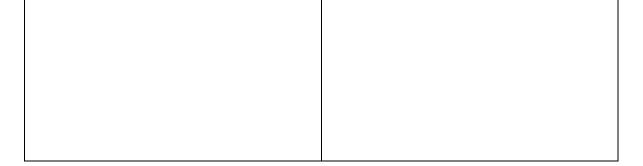


(f) (i) A Twaron polymer strand is hydrolysed.

below.

State the conditions that could be used to hydrolyse the polymer.

(ii) Draw the formulae of the **two** organic products of hydrolysis in the boxes.



[2]

[1]

[1]

[1]

Resource Materials

Question Set No: 11

The Periodic Table of the Elements

0	18	2 He	helium 4.0	10	Ne	20.2	18	Ar	argon 39.9	36	Ā	krypten 83,8	54	Xe	31.3	98	R	nadon			
6			17	6	ш	fluorine 19.0	17	CI	cliarine 35.5	35	ģ	79,9	53	-	iodine 126.9	85	Ą	asbaline			
(9)			16	00	0	000gen 16.0	16	s	32.1	34	Se	79.0	52	Te	127.6	84	8	polonium	116	۲۸	ivermonum
(2)			15	2	z	nitrogen 14.0	15	۵	shosphorus 31.0	33	As	arsenic 74.9	51	Sb	antimony 121.8	83	ā	209.0			
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(3)			13	2	В	10.8	13	Αĩ	aluminium 27.0	31	gg	9allum 69.7	49	드	indium 114.8	81	11	thellium 204.4			
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		Jec	mass						9	24	ប៉	chromium 52.0	42	W	таубалит 95.9	74	8	tungsben 183.8	106	Sg	seaborgum
	Key	atomic number Symbol	ve atomic						2	23	>	vanadium 50.9	41	g	niotium 92.9	73	Тa	180.9	105	음	dubnum
		atc	relati						4	22	F	tomium 47.9	40	Zr	ziroonium 91.2	72	Ì	hefhium 178.5	104	¥	nutherfordium
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(3)			2	4	Be	Benylfum 9.0	12	Mg	тврнея 24.3	20	ပီ	40.1	38	Š	strontium 87.6	26	Ba	137.3	88	Ra	Rigina
Ξ	-	- I	hydrogen 1.0	က	=	187ium 6.9	7	Na	23.0	19	¥	39.1	37	Rb	nbidium 85.5	22	ర	132.9	87	Ŧ	francium

Some useful organic reactions

1
$$R-Br + CN^- \longrightarrow R-CN + Br^-$$

2 R-CN
$$\xrightarrow{H^+ \text{ (aq)}}$$
 R-COOH

$$\begin{array}{c} & & & \\ & &$$



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