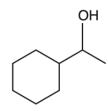


GCE A level Chemistry A (H432) H432/02 Synthesis and analytical techniques

MCQ Question Set 3
4.1 Basic concepts and hydrocarbons

Multiple Choice Questions

1. How can the molecule below be described?

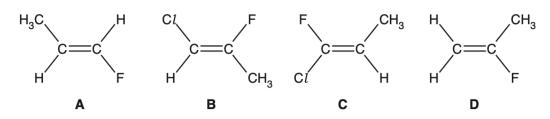


- A Aromatic and alicyclic
- **B** Aliphatic and unsaturated
- C Aromatic and unsaturated
- D Alicyclic and saturated

Your answer

[1]

2. Which molecule is a Z-isomer?



Your answer

3. What is the molecular formula of the compound below?

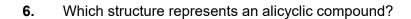


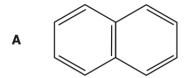
- **A** C₇H₁₀
- **B** C_7H_{12}
- C C₇H_{4,4}
- $D C_7H_{16}$

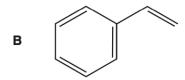
Your answer

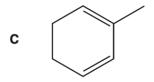
[1]

4.	Ethane r	reacts with chlorine by radical substitution to form chloroethane.			
	Which ra	adical(s) is/are present in the mechanism?			
		1 H•			
		2 Cl•			
		$3 C_2H_5^{\bullet}$			
	Α	1, 2 and 3			
	В	Only 1 and 2			
	С	Only 2 and 3			
	D	Only 1			
	Your answer				
5.	Which c	ompound(s) is a/are structural isomer(s) of C ₆ H ₁₂ O ₂ ?			
		1 hexanoic acid			
		2 ethyl butanoate			
		3 propyl propanoate			
	Α	1, 2 and 3			
	В	Only 1 and 2			
	С	Only 2 and 3			
	D	Only 1			
	Yo	our answer	[1]		











Your answer

[1]

- 7. Which alkane has the highest boiling point?
 - A CH₃(CH₂)₅CH₃
 - $\mathbf{B} \quad (\mathrm{CH_3})_3 \mathrm{CCH}(\mathrm{CH_3})_2$
 - C CH₃(CH₂)₃CH(CH₃)₂
 - $\mathbf{D} \quad (\mathrm{CH_3})_2\mathrm{CHCH_2CH(CH_3)_2}$

Your answer

[1]

8.	Butane reacts with	chlorine in	the	presence	of	ultraviolet	radiation	to	form	а	mixture	of
	organic products.											

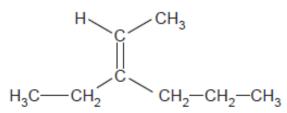
Which equation shows a propagation step in the mechanism for this reaction?

- A $Cl_2 \rightarrow {}^{\bullet}Cl + {}^{\bullet}Cl$
- $\textbf{B} \quad \bullet \textbf{C} l + \bullet \textbf{C}_4 \textbf{H}_8 \textbf{C} l \rightarrow \textbf{C}_4 \textbf{H}_8 \textbf{C} l_2$
- $\textbf{C} \quad \text{C}_4\text{H}_9\text{C}l \ + \ ^{\bullet}\text{C}l \ \rightarrow \ \text{C}_4\text{H}_8\text{C}l_2 \ + \ ^{\bullet}\text{H}$
- $\textbf{D} \quad {}^{\bullet}\textbf{C}l \; + \; \textbf{C}_{4}\textbf{H}_{9}\textbf{C}l \; \rightarrow \; {}^{\bullet}\textbf{C}_{4}\textbf{H}_{8}\textbf{C}l \; + \; \textbf{H}\textbf{C}l$

Your answer



9. What is the name of the compound below?



- A 3-Propylpent-2-ene
- **B** 3-Propylpent-3-ene
- C 3-Ethylhex-2-ene
- **D** 4-Ethylhex-4-ene

Your answer	
-------------	--

Which term correctly describes this stereoisomer?

A cis-

10.

- B trans-
- C E-
- D *Z*-

Your answer	
-------------	--

[1]

[1]

[1]

11. Which types of bonds are broken and formed in the reaction of ethene and bromine?

	Types of bond broken	Types of bond formed
Α	σ	π
В	π	σ
С	σ and π	π
D	σ and π	σ

Your answer			

[1]

Total Marks for Question Set 3: 11



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