

GCE A level Chemistry A (H432) H432/02 Synthesis and analytical techniques

Question Set 23

- This question is about esters.
- (a) The structure of ester **A** is shown below.



- (i) What is the systematic name of ester A?
- (ii) In the boxes, draw the organic products for the reactions of the functional groups in ester **A** shown below.

+ $H^+(aq)$ $H^+(aq)$ Br Ester A $excess OH^-(aq)$ +

Each reaction forms two organic products.

(iii) Name the type of reactions of ester A shown in (ii).

[5]

[1]



The protons in ester **A** are in four different environments, labelled 1-4 on the structure below.



Complete the table to predict the **proton** NMR spectrum of ester **A**.

Proton environment	Chemical shift	Splitting pattern
1		
2		
3		
4		

(c) Compound **B** is a structural isomer of ester **A**.

- Compound **B** reacts with aqueous sodium carbonate.
- The ¹³C NMR spectrum of **B** has 4 peaks.

Draw a possible structure for compound **B**.

(d) A polyester is formed from 200 molecules of 4-hydroxybenzoic acid.

What is the relative molecular mass, M_r , of the polyester?

(e) A student intends to synthesise ester C.



(i)* Plan a two-stage synthesis to prepare 12.75 g of ester C starting from 2-methylpropanal, $(CH_3)_2$ CHCHO.

Assume the overall percentage yield of ester **C** from 2-methylpropanal is 40%.

In your answer include the mass of 2-methylpropanal required, reagents, conditions and equations where appropriate.

Purification details are **not** required.

[6]

[4]

[1]

[2]

(b)

(ii) The mass spectrum of ester **C** is shown below.



Suggest possible structures for the species responsible for peaks **Y** and **Z** in the mass spectrum.

[2]

Total Marks for Question Set 23: 22



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