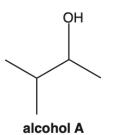


A level Chemistry A

H432/02 Synthesis and analytical techniques

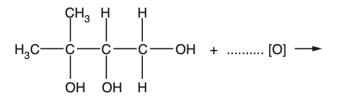
Question Set 14

- 2. This question is about reactions of organic compounds containing carbon, hydrogen and oxygen.
 - A chemist investigates two reactions of alcohol **A**, shown below. (a)



- What is the systematic name of alcohol A?
- (i) [1] What is the structural formula of alcohol A? (ii) [1] (iii) The chemist heats alcohol A with an acid catalyst to form a mixture containing two alkenes. Draw the structures of the **two** alkenes formed in this reaction. [2] (iv) The chemist heats alcohol A with sodium chloride and sulfuric acid. Construct a balanced equation for this reaction. Show structures for the organic compounds in your equation. [2] Compound B, shown below, is refluxed with excess acidified potassium (b) dichromate(VI) to form a single organic product.

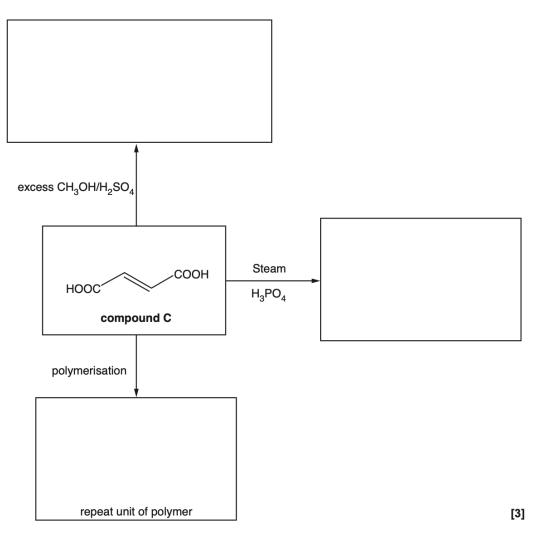
Complete the equation for this reaction.



compound B

[2]

The flowchart below shows some reactions of compound **C**.In the boxes, draw the organic products of these reactions.



[3]

Total Marks for Question Set 14: 11

(C)



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