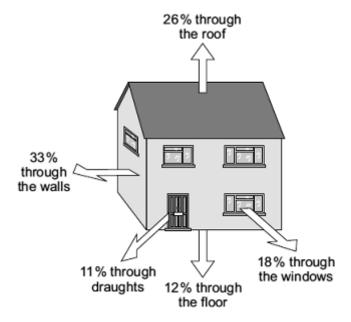
Q1. The diagram shows where heat is lost from a house that is **not** insulated.



(a) (i)	Through which part of the house is most heat lost?		
		(1)	

 (1)

(b) A homeowner wants to reduce her energy bills and make her home more energy efficient. The table shows five ways this could be done. The table also shows how much money each way would save the homeowner each year.

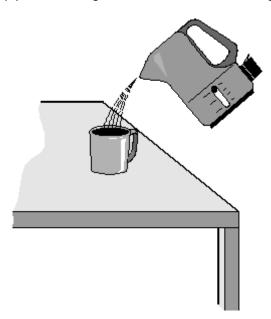
	Cost	Money saved each year
Installing loft insulation	£175	£60
Fitting draught-proofing	£45	£20
Installing cavity wall insulation	£300	£80
Adding a hot water tank jacket	£15	£20

Using energy efficient light bulbs		£60	£30		
(i)	Which one of the five way energy bill the most?	ys of reducing ene	rgy bills would redu	uce the yearly	
					(1)
(ii)	This year the homeowner efficiency of her home.	has only got £60 t	o spend to improve	e the energy	
	Use the information in the this money on.	e table to explain w	hat the homeowne	r should spend	

(2) (Total 5 marks)

Page 3

Q2. (a) The diagram shows hot water being poured into a mug.



(i) Complete the sentence by choosing the correct words from the box. Each word may be used once or not at all.

air mug table water

Heat energy is being transferred from the to the

(1)

(1)

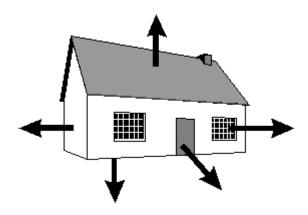
(ii) When will this transfer of heat energy stop?

(b) In the box are the names of four types of fuel used to heat homes.

coal	gas	oil	wood

Which **one** of these types of fuel is renewable?

(c) The diagram shows where heat energy is lost from a house.



(i) Complete the sentences by choosing the correct words from the box. Each word may be used once or not at all.

conduction conductor convection electric evaporation insulator

(3)

(1)

(ii)	Write down one other way of reducing heat loss from a house.	
		(1) (Total 7 marks)

Q3.		People	e do a number of things to reduce the energy loss from their homes.	
	(a)	Des	cribe one thing they may do to cut down the energy loss through:	
		(i)	the roof;	
				(1)
		(ii)	the outside walls;	
				(1)
		(iii)	the glass in the windows;	
				(1)
		(iv)	gaps around the front and back doors.	
				(1)
	(b)	A ho make	ouse is more difficult to keep warm in cold weather. What other type of weather es it difficult to keep a house warm?	
			(Total 5 mar	(1) rks)

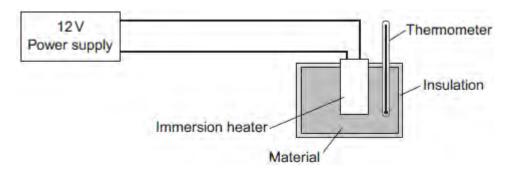
Q4.A student used the apparatus in **Figure 1** to compare the energy needed to heat blocks of different materials.

Each block had the same mass.

Each block had holes for the thermometer and the immersion heater.

Each block had a starting temperature of 20 °C.

Figure 1



The student measured the time taken to increase the temperature of each material by 5 °C.

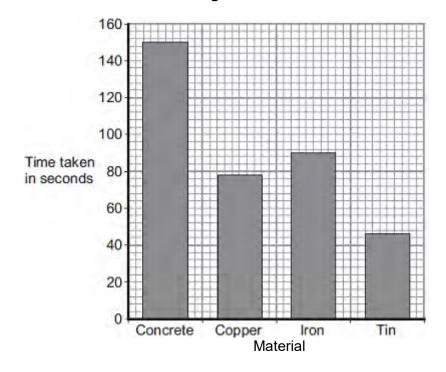
(a) (i) State **two** variables the student controlled.

1	
2	

(2)

Figure 2 shows the student's results.

Figure 2

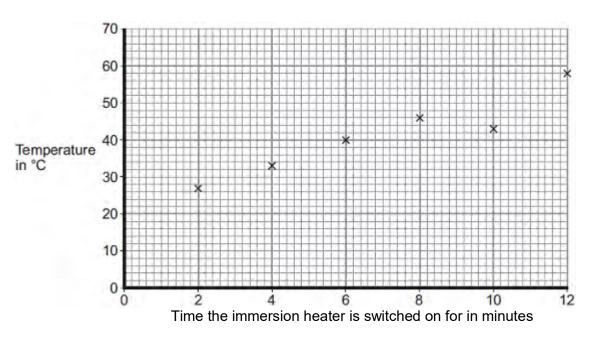


Page 8

(ii)	Why was a bar chart drawn rather than a line graph?		
		(1)	
(iii)	Which material was supplied with the most energy?		
	Give the reason for your answer.		
		(2)	
(iv)	The iron block had a mass of 2 kg.		
(,	Calculate the energy transferred by the heater to increase the temperature of the iron block by 5 °C.		
	The specific heat capacity of iron is 450 J / kg °C.		
	Energy transferred = J	(2)	
T 1			
	student used the same apparatus to heat a 1 kg block of aluminium.		
	ecorded the temperature of the block as it was heated from room temperature.		
The	results are shown in Figure 3 .		

Figure 3

(b)



(i) One of the student's results is anomalous.

Draw a ring around the anomalous result.

(1)

(ii) Draw the line of best fit for the points plotted in Figure 3.

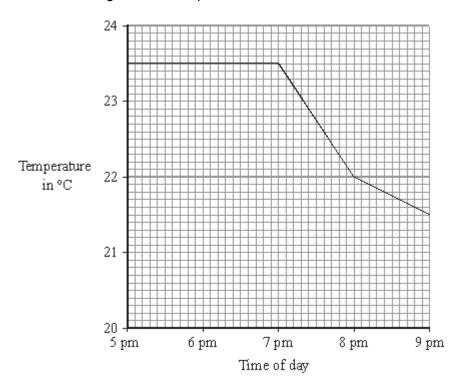
(1)

(iii) What was the temperature of the room?

(iv) What was the interval of the time values used by the student?

Interval = minutes

(1) (Total 11 marks) **Q5.** (a) The graph shows the temperature inside a flat between 5 pm and 9 pm. The central heating was on at 5 pm.



(i) What time did the central heating switch off?

.....

(1)

(2)

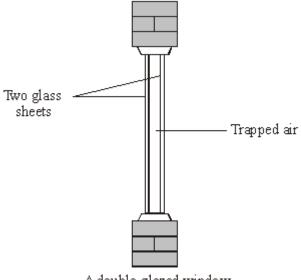
(ii) Closing the curtains reduces heat loss from the flat.

What time do you think the curtains were closed?

Give a reason for your answer.

.....

(b) Less heat is lost through double-glazed windows than through single-glazed windows.



A double-glazed window

Complete the following sentences by choosing the correct words from the box. Each word may be used once or not at all.

conduction	conductor	convection	evaporation	insulator	radiation	
oonaaoaon	Conductor	CONTROCTION	Craporation	modiator	iaaiatioii	

rir is a good		
glass it reduces heat loss by	and	(3)

(c) The table gives information about three types of house insulation.

Type of insulation	Cost to install	Money save each year on heating bills	Payback time
Double glazing	£4000	£200	20 years
Loft insulation	£300	£100	3 years
Cavity wallinsulation	£600	£150	

(1)	insulation.	
		(1)
(ii)	Explain why people often install loft insulation before installing double glazing	
()	or cavity wall insulation.	
	/Total 0 m	(2)
	(Total 9 m	arks)

Q6. (a) The diagram shows two switches on a room heater. The heater has three power settings. The power produced by two of the settings is given in the table.

Tananananananananananananananananananan	Setting	Power in kW
Switches	Low	0.5
	Medium	1.5
	High	

- (ii) The heater is used on the **medium** power setting. It is switched on for three hours.

Use the equation in the box to work out the energy transferred from the mains to the heater in three hours.

energy transferred (kilowatt-hour, kWh)	=	power (kilowatt, kW)	×	time (hour, h)

Show clearly how you work out your answer.	
Energy transferred = kWh	

(2)

(iii) Electricity costs 12 pence per kilowatt-hour.

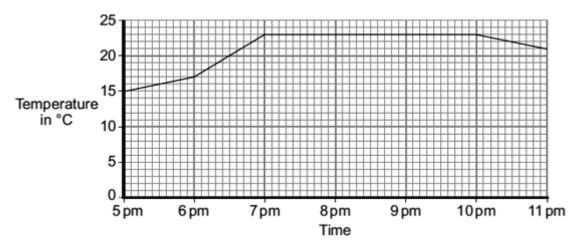
Use the equation in the box to calculate how much the heater costs to use on **medium** power for three hours.

total cost = number of kilowatt-hours × cost per kilowatt-hour

how clearly how you work out your answer.			
Total cost =pence	(2)		

(b) The heater is used to warm a room.

The graph shows how the temperature of the room changes from the moment the heater is switched on.



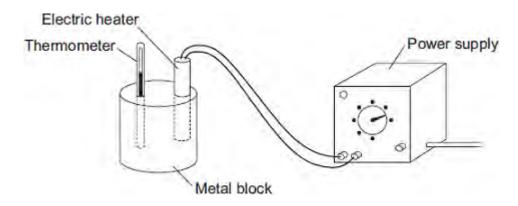
The heater was first used on the medium setting.

(i)	At what time was the heater setting changed to the high setting?
	Give a reason for your answer.

(2)

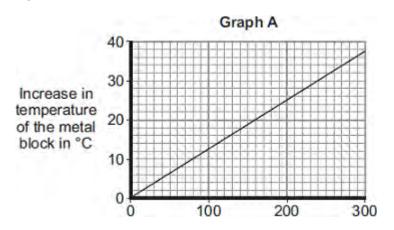
(ii)	From 7 pm until 10 pm, the temperature of the room is not changing.				
	Which one of the following statements gives the reason why the temperature of the room is not changing?				
	Put a tick (\checkmark) in the box next to your answer.				
The roo	m is losing energy as fast as the heater supplies energyenergy				
The roor	m is losing energy faster than the heater supplies energy				
	(Total	(1) 8 marks)			

Q7.(a) A student used the apparatus drawn below to investigate the heating effect of an electric heater.



(i) Before starting the experiment, the student drew **Graph A**.

Graph A shows how the student expected the temperature of the metal block to change after the heater was switched on.

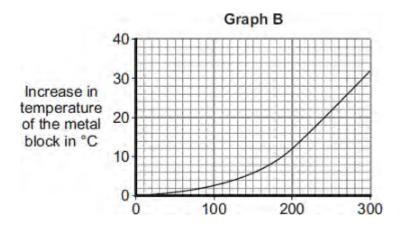


	•	hown i	-			

(ii) The student measured the room temperature. He then switched the heater on and measured the temperature of the metal block every 50 seconds.

(2)

The student calculated the increase in temperature of the metal block and plotted **Graph B**.



After 300 seconds, **Graph B** shows the increase in temperature of the metal block is lower than the increase in temperature expected from **Graph A**.

Suggest **one** reason why.

iii)	The power of the electric heater is 50 watts.
	Calculate the energy transferred to the heater from the electricity supply in 300 seconds.

Energy transferred = J

(1)

(2)

(b) The student uses the same heater to heat blocks of different metals. Each time the heater is switched on for 300 seconds.

Each block of metal has the same mass but a different specific heat capacity.

Metal	Specific heat capacity in J/kg°C
Aluminium	900
Iron	450

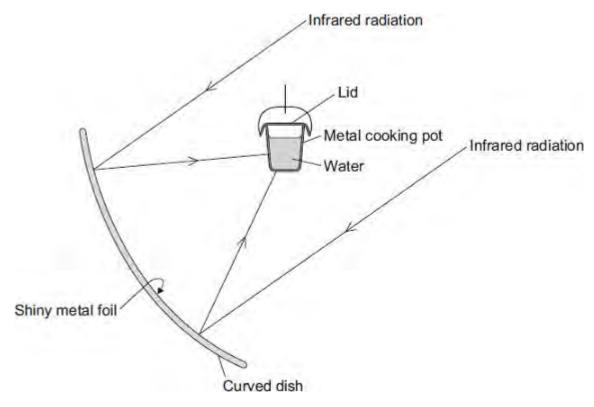
130

Which **one** of the metals will heat up the most?

Draw a ring around the correct answer.

aluminium	iron	lead	
Give, in terms of the amour your answer.	nt of energy needed to	heat the metal blocks, a	reason for
			 (2) (Total 7 marks)

Q8.The diagram shows the design of a solar cooker. The cooker heats water using infrared radiation from the Sun.



(a)	Why is the inside of the large curved dish covered with shiny metal foil?		
		(1)	

silver

white

(2)

(b) Which would be the best colour to paint the outside of the metal cooking pot?Draw a ring around the correct answer.

black

Give a reason for your answ	ver.	

(c)	Why does the cooking pot have a lid?	
		(1)
(d)	Calculate how much energy is needed to increase the temperature of 2 kg of water	
(d)	by 80 °C.	
	The specific heat capacity of water = 4200 J/kg °C.	
	Energy =	(2)
	(Total 6 n	