June 2004

INTERNATIONAL GCSE

MARK SCHEME

MAXIMUM MARK: 40

SYLLABUS/COMPONENT: 0620/05

CHEMISTRY Practical



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Table of results **Experiment 1** 1 Temperature boxes completed Increasing 1 Comparable to supervisor 1 [3] Experiment 2 1 Temperature boxes completed Decreasing 1 Comparable to supervisor 1 [3] (a) All points plotted correctly 4 (-1 for each incorrect) 2 Smooth line graphs Labelled 1 [7] 1. Value from graph 1 (b) (i) Value from graph ± 0.25 \ No unit only (1) 2. 1 [2] (ii) 1. Exothermic 1 1 2. Endothermic [2] 1 (c) Fizz/bubbles/effervescence Solid disappears 1 [2] 1 (d) Carbonate Fizz with acid or similar 1 [2] 1 (e) Solid **A** – value from table/room temperature ± 3°C Solid **B** – value from table/room temperature 1 Reaction finished 1 [3] Sub Total [24] 1 2 (a) White [1] White 1 (c) (i) Precipitate 1 [2] 1 [1] Excess - no change (ii) 1 [1] No precipitate/change (iii) Paper goes blue 1 Fizz/bubbles etc 1 1 Reference to smell [3] 1 [1] (iv) pH greater than 7 (v) Milky/cloudy 1 [1] (d) Calcium 1 [1] (e) Ammonia 1

	Syllabus	Paper
Chemistry – June 2004	0620	5
Chemical Cane 2004	0020	

(f)	Limewater Carbon dioxide	1 1	[2]
(g)	Nitrate Hydroxide	1 1	[2]
		Sub Total	[16]
		Total	[40]